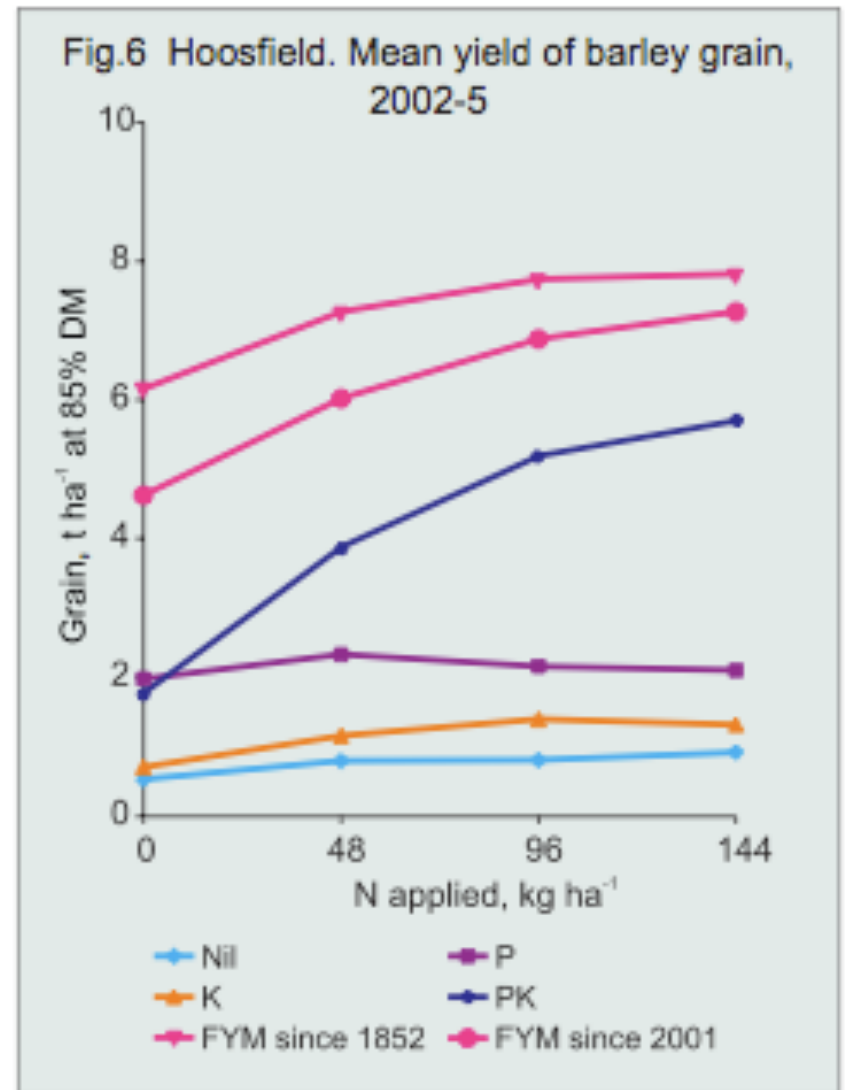




Broadbalk Continuous Wheat
Started in 1843



Hoosfield spring barley
Started in 1852

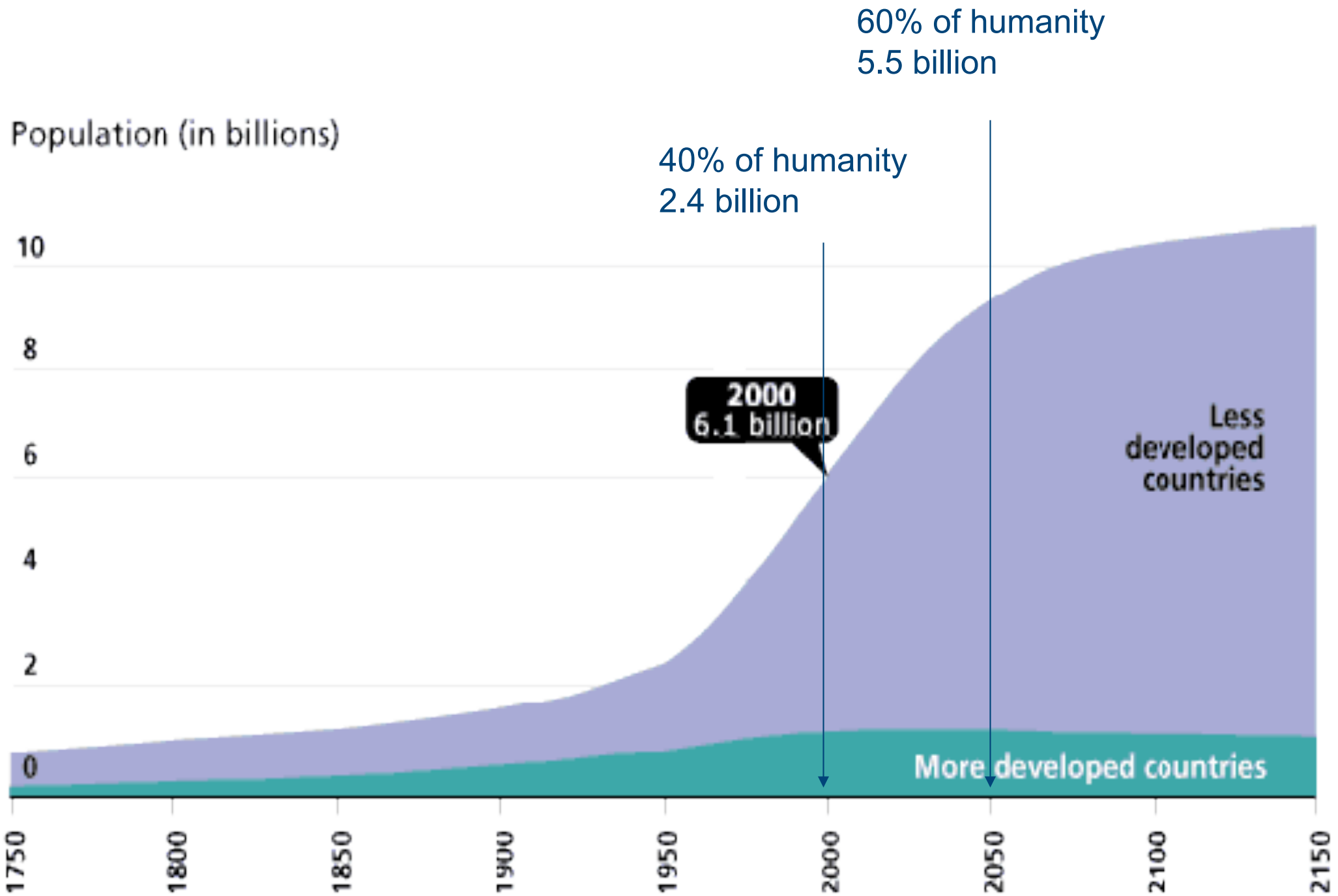
How much of humanity's
“existential dependence” on
synthetic N can be attributed to
P?



“And so the question about the most important invention of the twentieth century—important in the most fundamental existential terms for the largest number of people— elicits almost uniformly a wrong answer, as nothing comes close to the Haber Bosch synthesis of ammon

Vaclav Smil
University of Manitoba





Saturating and topping up soil P
vs. the danger of hitting the
“change point”

Topping up P

- In order to build up soil P to the critical value for crop production, it may be necessary to accept a lower recovery of added P for a number of years. Once the critical level is achieved in many arable cropping systems, the amount of P required to maintain it is often similar to that removed in the (very high P-use efficiency).
- The optimum soil-P status is reached when the optimum P-application rate equals crop P removal (Sibbesen and Sharpley 1997)

TABLE A1.11

P offtake 1856–2001 by arable crops growing on plots that had received no P or a total of 1 410 kg ha⁻¹ from 1856 to 1901 and none since, Exhaustion Land, Rothamsted

Period and number of years	Cropping ¹	Plot 7 (NPK) ² P offtake		Plot 5 (N) ² P offtake (kg ha ⁻¹)		Difference in annual P offtake
		Total	Per year	Total	Per year	
1856–1875 20	Wheat	160	8.00	93	4.65	3.35
1876–1901 26	Potatoes	169	6.50	45	1.73	4.77
1902–1940 39	Barley	235	6.02	131	3.36	2.66
1941–1948 8	Barley	72	9.00	39	4.90	4.10
1949–1974 26	Barley	248	9.54	116	4.46	5.08
1976–1991 16	Barley	135	8.44	65	4.06	4.38
1992–2001 10	Wheat	96	9.63	42	4.18	5.45

¹ P in winter wheat and spring barley grain plus straw and in potato tubers.

² Except 1902–1940 when no N was applied.

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¹ P in winter wheat and spring barley grain plus straw and in potato tubers.

² Except 1902–1940 when no N was applied.

1115

531

P recovery of fertilizer 584 or 41% (difference) or 79% (balance)

Fig.7 Exhaustion Land. Response of spring barley after 3 years fresh P

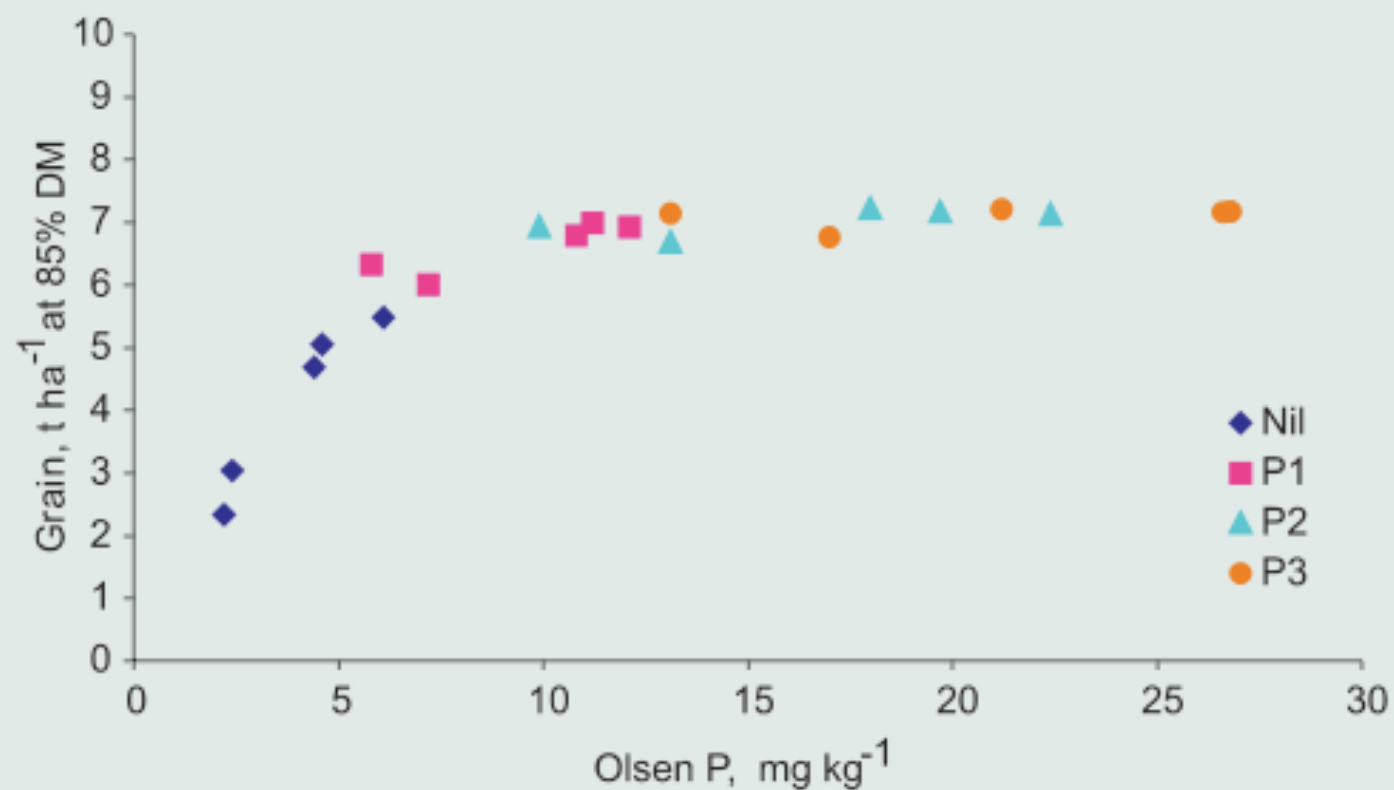


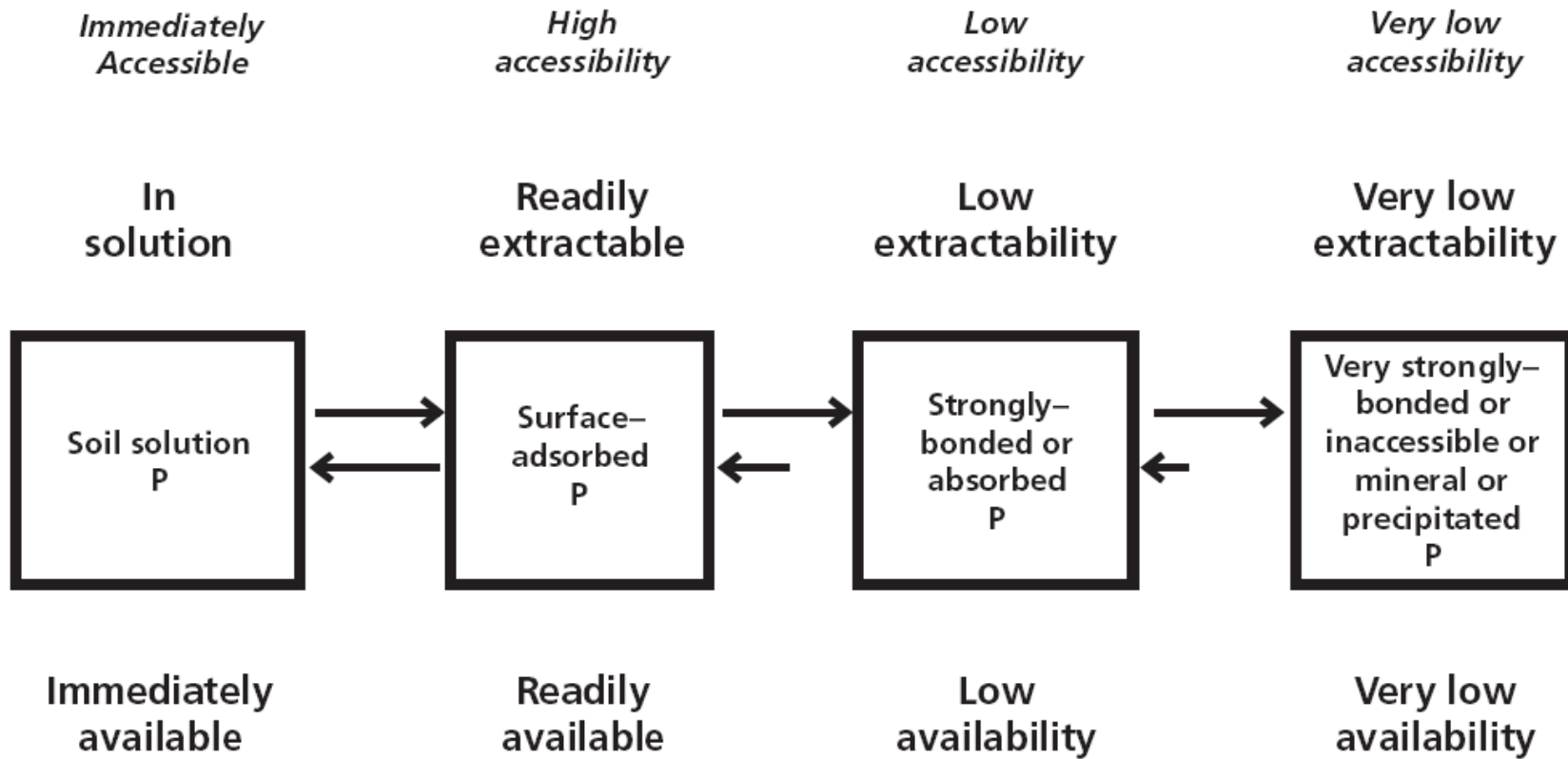
TABLE 11

Relationship between P balance at the end of each treatment period and the change in soil P fractions, Exhaustion Land, Rothamsted

Plot, treatment & period	P balance	Resin	Change in first five P fractions,kg P ha ⁻¹				Total change
			0.5 M NaHCO ₃	0.1 M NaOH	1 M NaOH	0.5 M H ₂ SO ₄	
	(kg ha ⁻¹)				(kg P ha ⁻¹)		
No P							
1903–1993	-300	-49	-20	-195	-82	-19	-365
FYM applied							
1876–1901	1 035	193	129	133	107	335	917
Not applied							
1903–1993	-752	-259	-123	-159	-112	-164	-817
P fertilizer applied							
1856–1901	1 222	158	126	79	7	270	640
Not applied							
1903–1993	-644	-234	-138	-114	45	-253	-694

Source: Adapted from Blake et al. (2003).

FIGURE 3
Conceptual diagram for the forms of inorganic P in soils categorized in terms of accessibility, extractability and plant availability



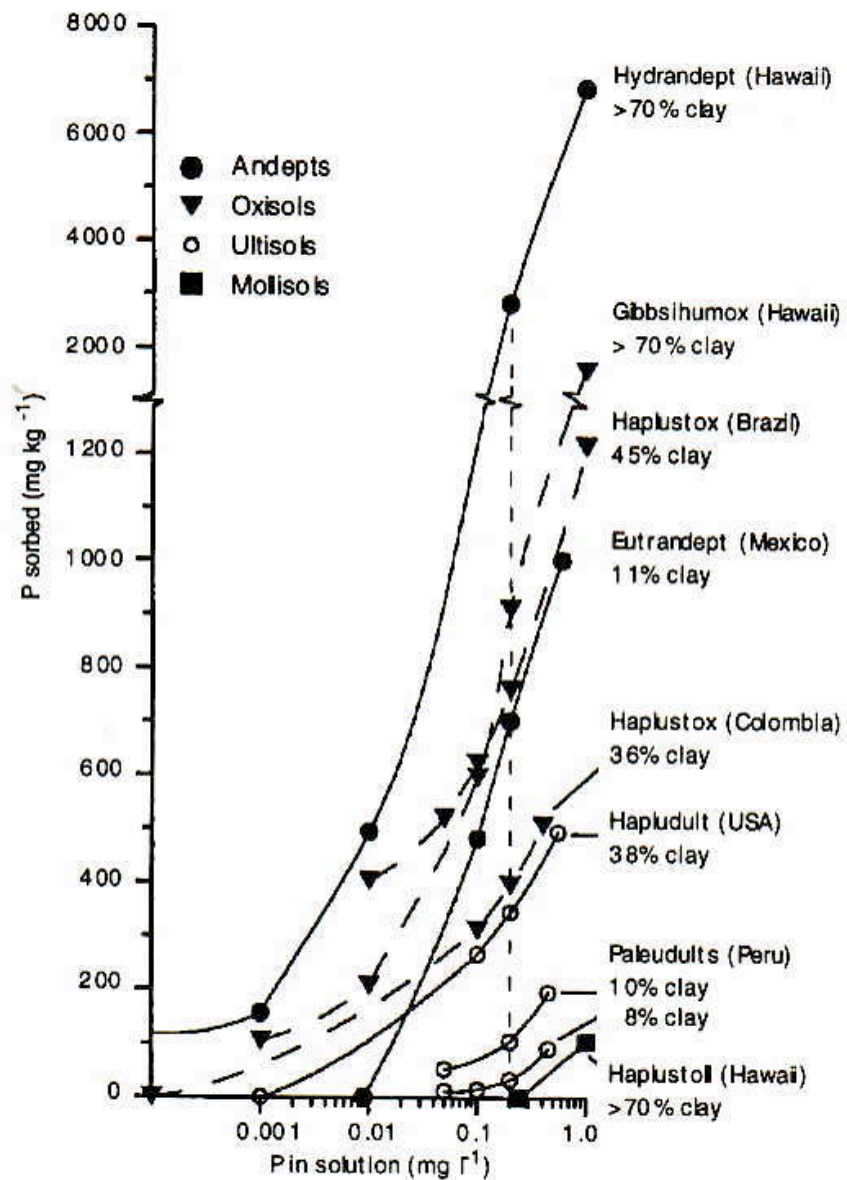


Figure 2. Examples of P sorption isotherms determined by the method of Fox and Kamprath (1970). Source: Sanchez and Uehara (1980).

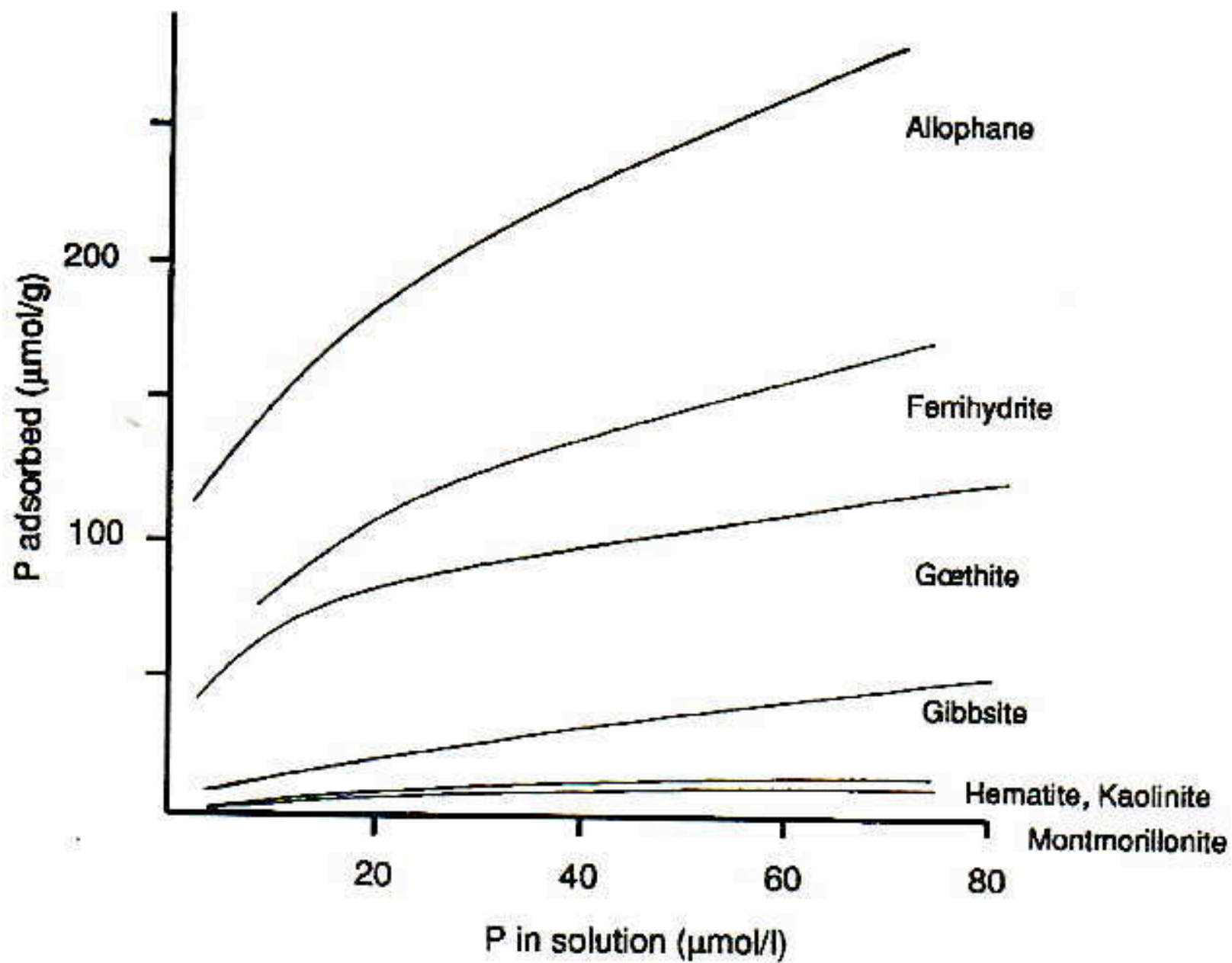


Figure 1. Sorption of P on various minerals (Sollins, 1991).

The “change point”

- An inflection point where P measured in field drainage waters (or in laboratory CaCl_2 soil extracts) shows a distinct linear increase with increasing Olsen-P (bicarbonate P).

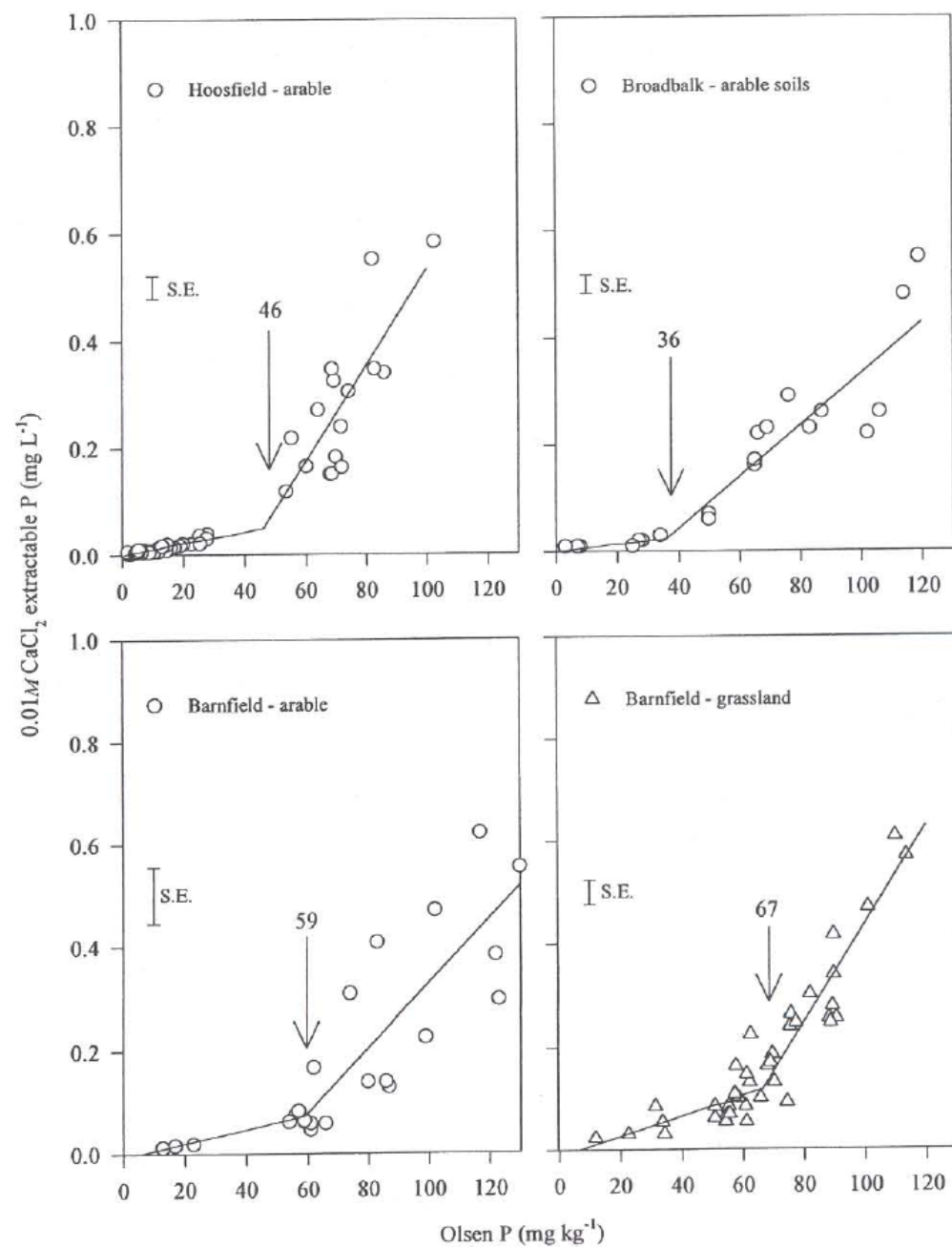


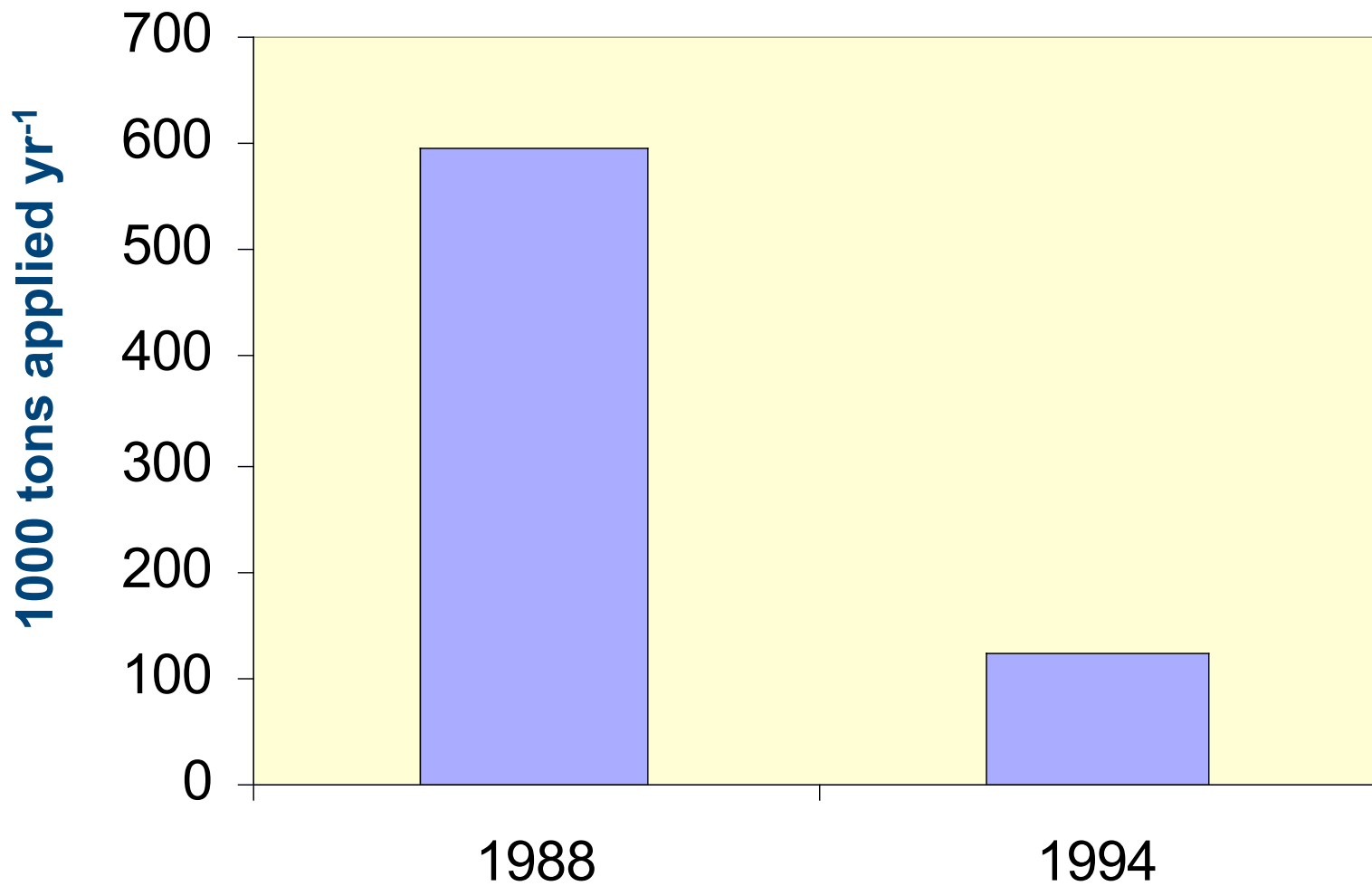
Fig. 1. Relationship between concentrations of CaCl₂-P and Olsen P of the Hoosfield, Broadbalk and Barnfield arable soils and the Barnfield grassland soil (split line model fitted shows the change point as indicated by the arrow; S.E. = standard error).

Change Point and topping up

- Does the “optimum soil P status” maintain soils anywhere near the change point?
- How easy is the proximity to the two states managed by farmers?
- Do soils below the change point still present a significant eutrophication hazard?

The dependence of organic
agriculture on historic P
fertilization or net transfers of
fertilizer P in manure

Change in fertilizers applied in Cuba following the break-up of the Soviet Union



If nutrient returns of P in organic matter (livestock or humanure) were improved, and P losses were reduced, could endogenous P supplies be used to top up?

Alternative agroecosystem models

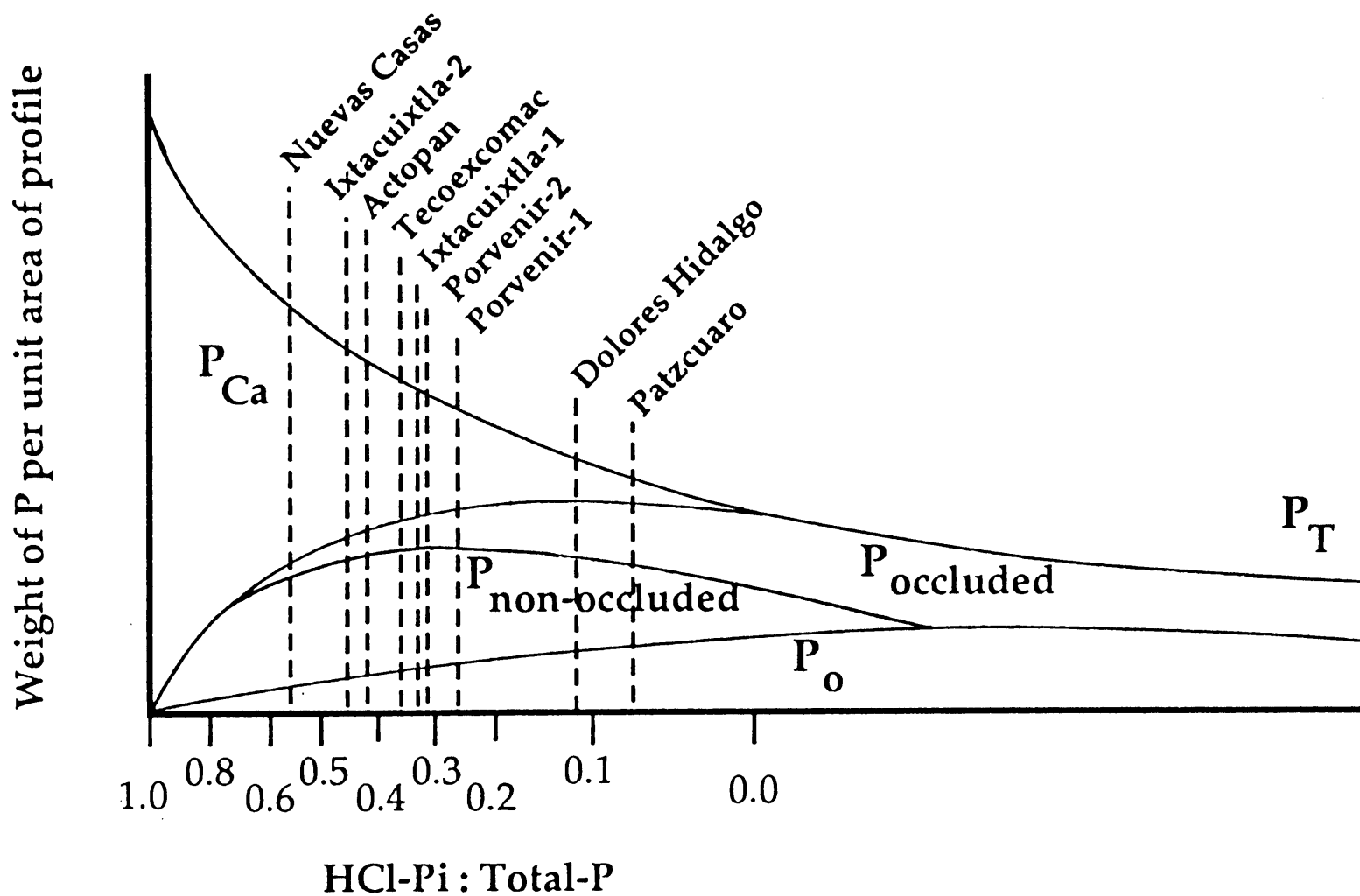
P inputs

kg ha⁻¹ yr⁻¹

Soil mineral weathering	0.05 - 1.0 (5.0 possible)
Atmospheric deposition	0.07 - 1.7
Total	0.12 - 2.7

Other P inputs that may be significant in particular ecosystems include: pollen, volcanic ash, fossil fuel combustion (e.g., fly ash), forest fires, ocean spray

Newman (1995)



P inputs from weathering

Reference or site	kg P ha ⁻¹ yr ⁻¹
Newman--review (1995)	0.05-1 (5 possible)
Smil (2000)	1.3
Hopi continuous maize (P in harvest)	2.2
Hayed prairie meadow (Glover)	5.9
Broadbalk continuous wheat, U.K.	3.7
Park Grass perennial meadow	5.3

P export in harvests from fertilized agriculture

Crop	kg P ha ⁻¹ yr ⁻¹
cereals	15-35
legumes and root crops	15-25
vegetables and fruits	5-15

Smil 2000

Sept

Dec

March

June

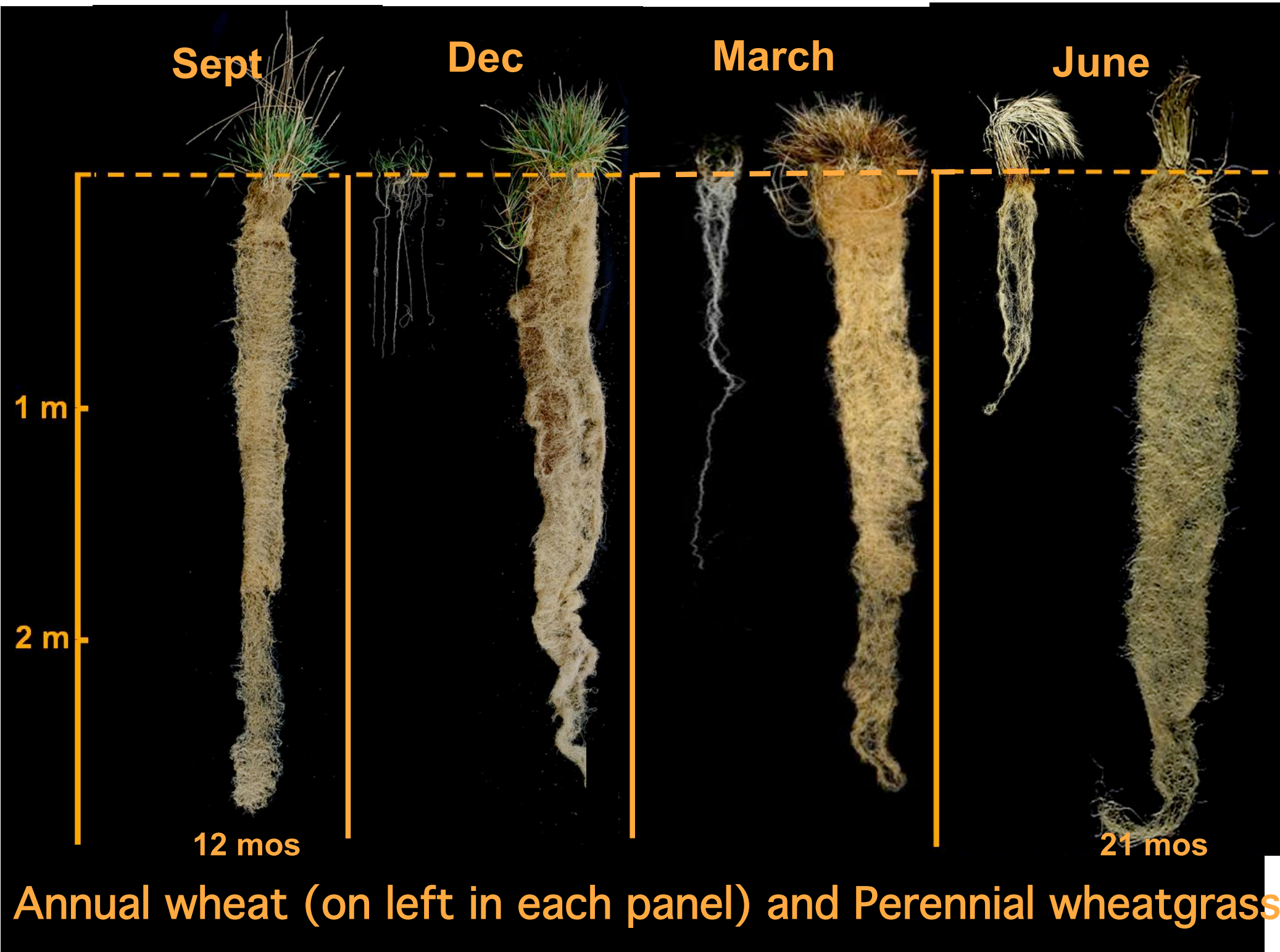
1 m

2 m

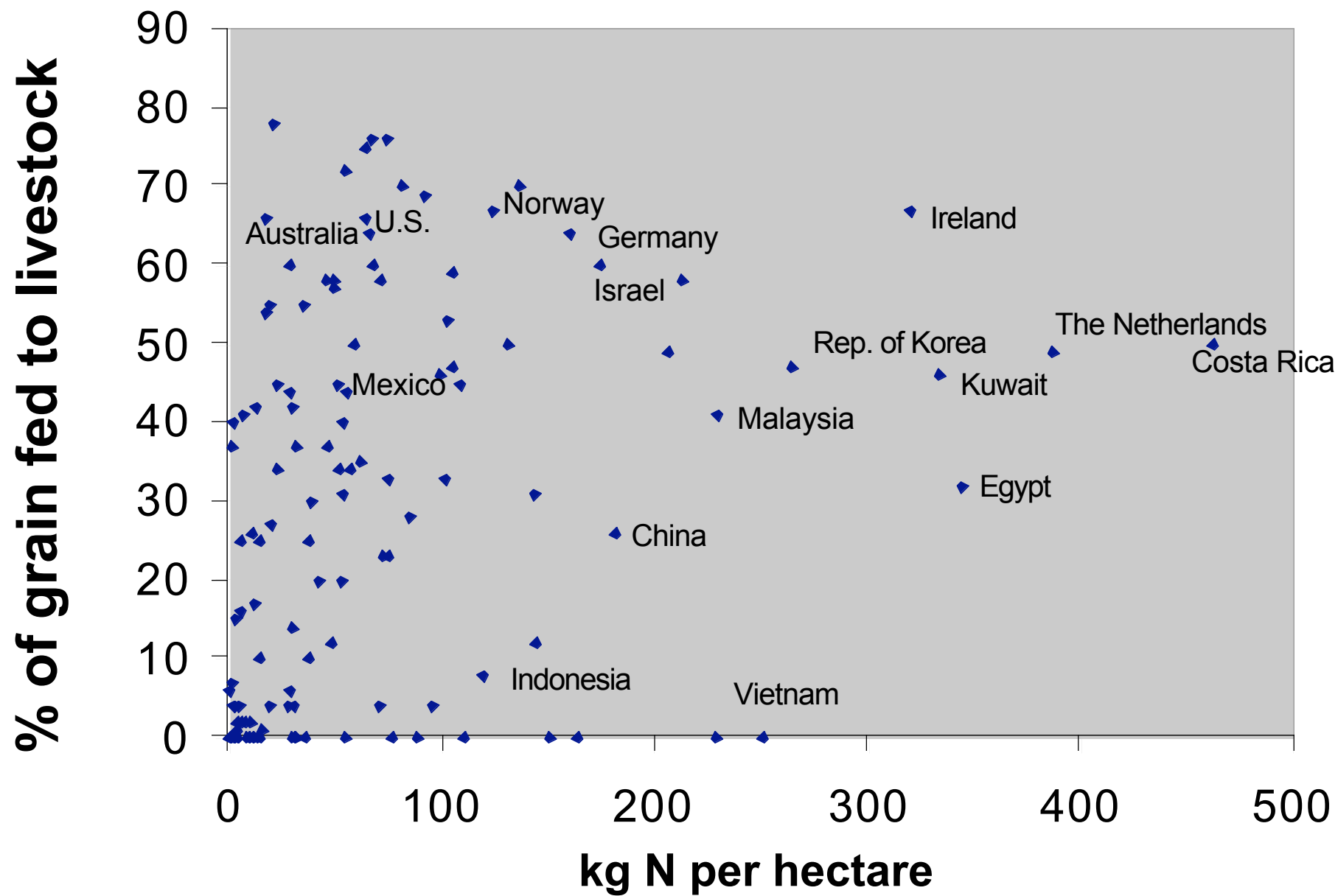
12 mos

21 mos

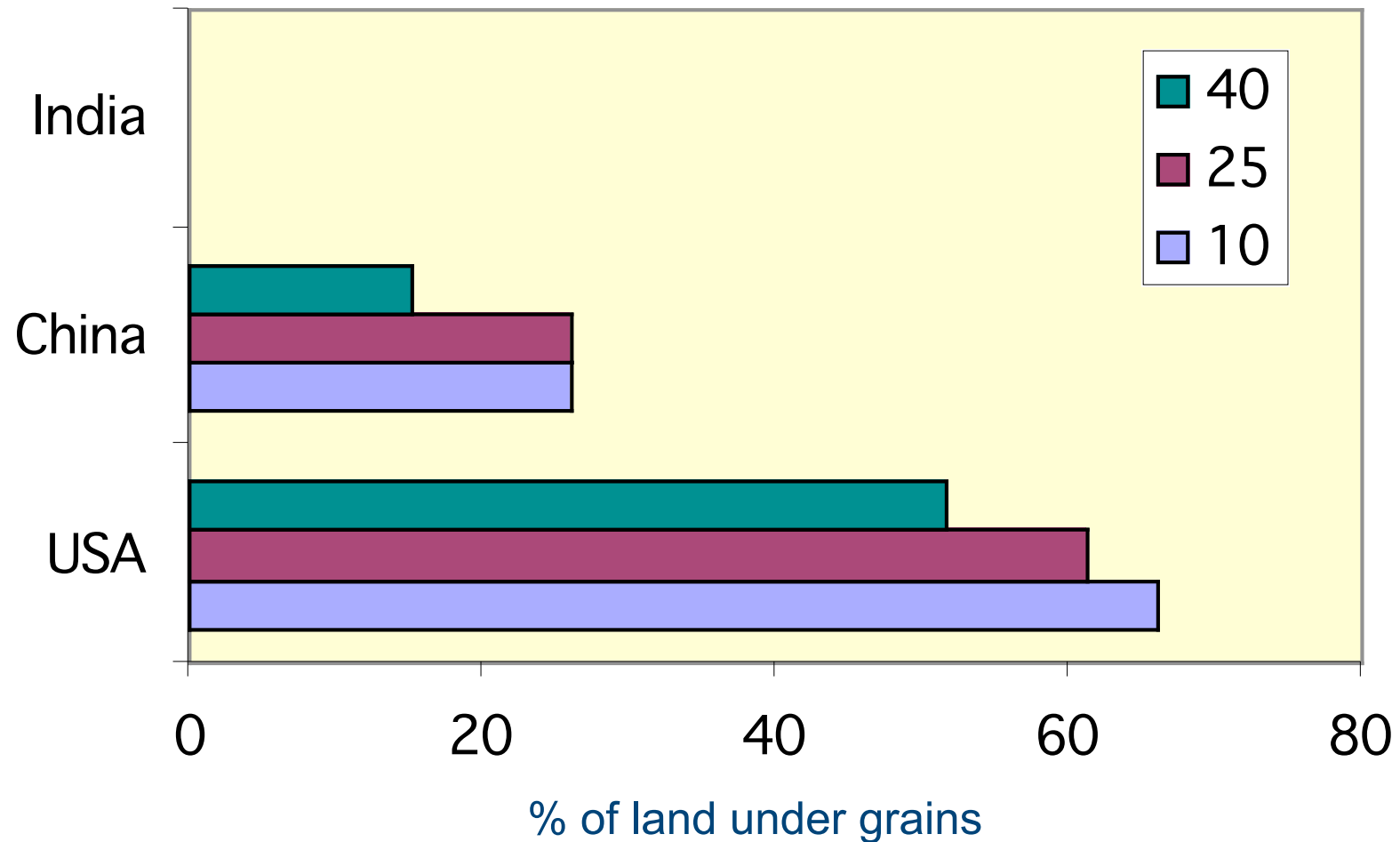
Annual wheat (on left in each panel) and Perennial wheatgrass



How different dietary scenarios
determine the need for P
fertilizers

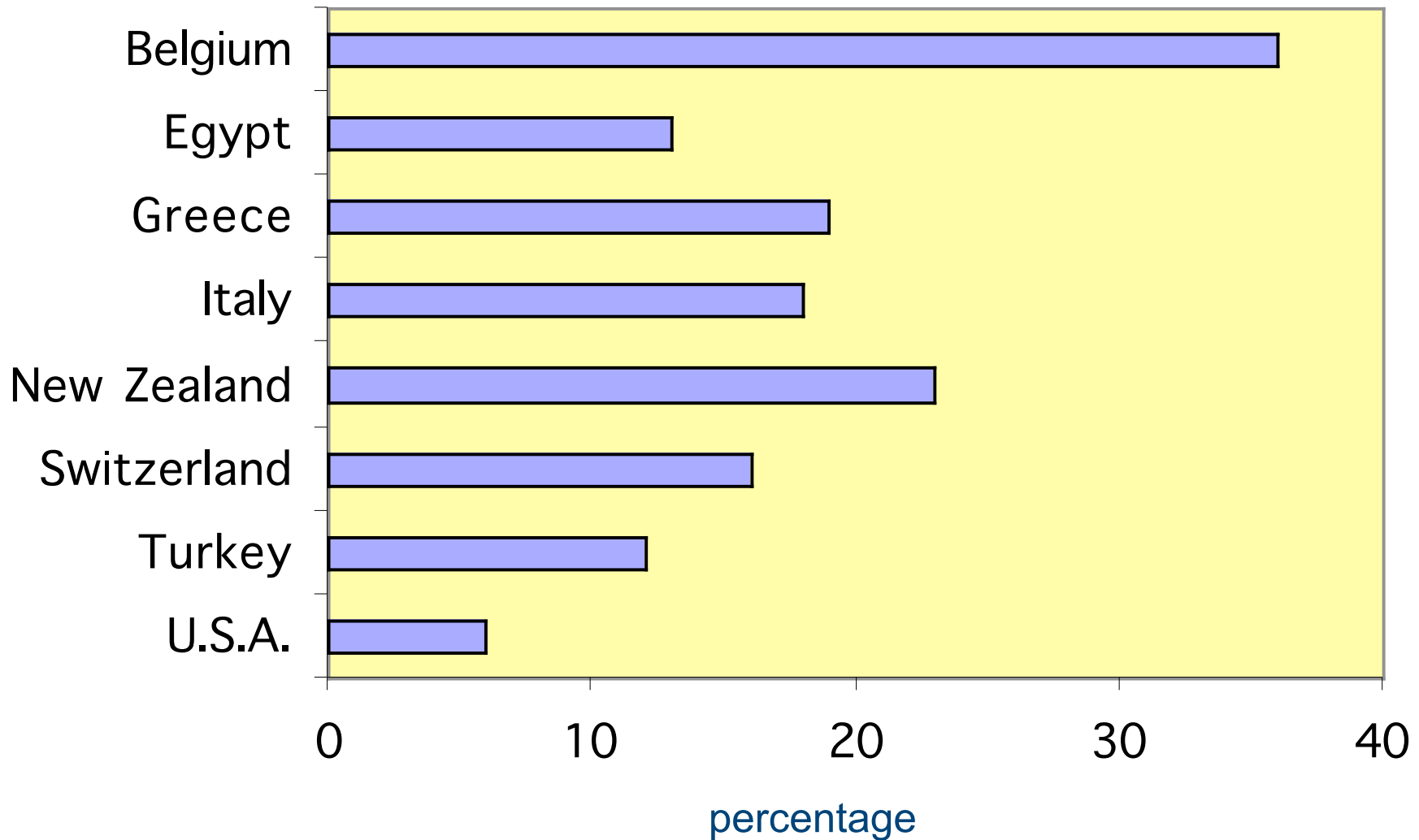


% of land under grain cultivation that could be freed up if the protein derived from meat were reduced to 40, 25 or 10 percent of daily requirements



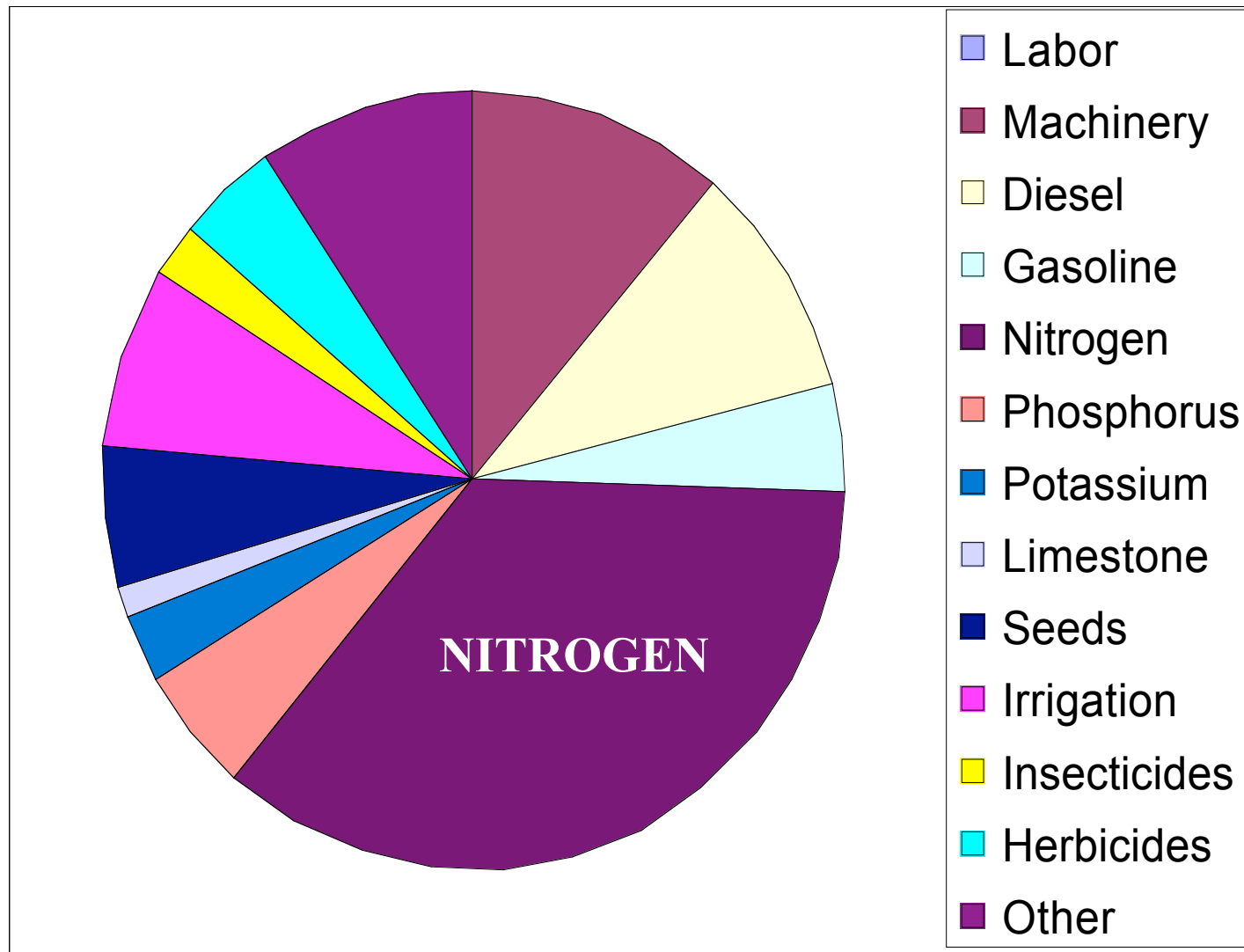
Crews and Peoples (2004)

**% of land under domestic grain production that could be freed-up
if per capita food supply was reduced to 3000 Cal day⁻¹**



From Crews and Peoples (2004)

Proportion of total energy cost of different inputs into US corn production



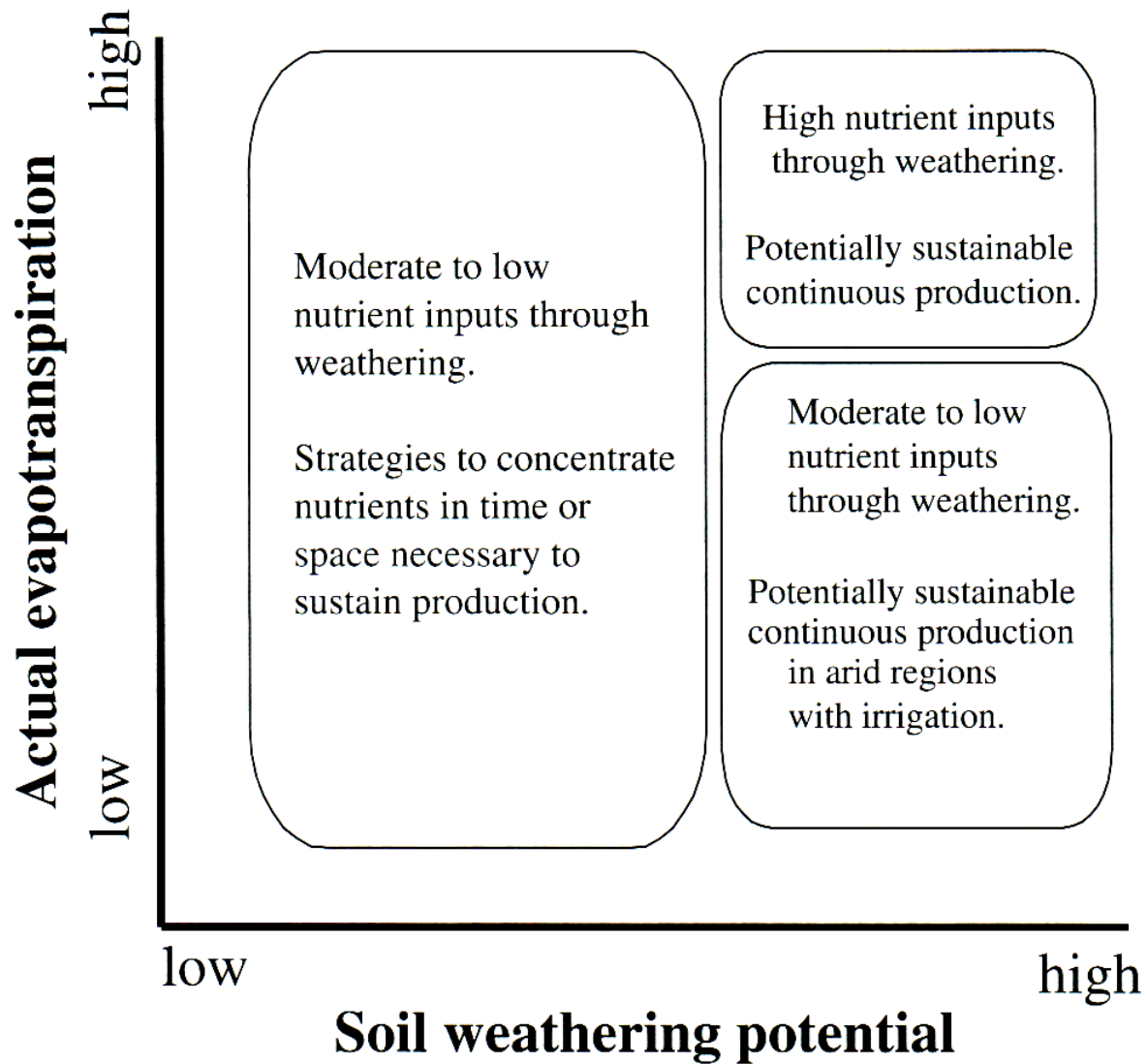
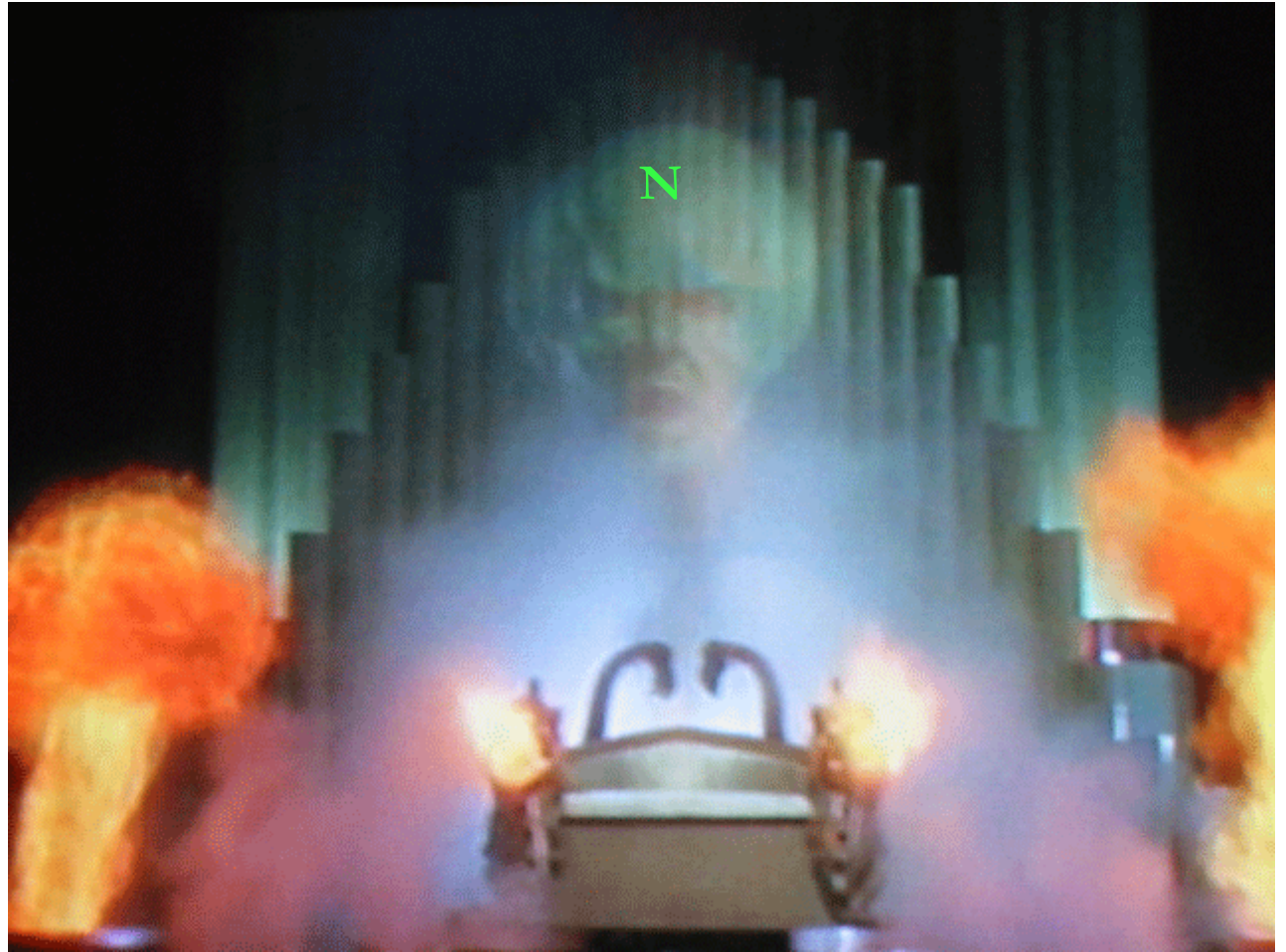


Table 2. Examples of calculated weathering potential index (WPI) values⁵⁶ and percentages of calcium (%Ca) and phosphorus (%P) for selected parent materials⁵⁴.

Rock type	WPI	% Ca	% P
Rhyolite	2.5	0.81	0.03
Granite	7	1.31	0.05
Gneiss	7	0.09	0.04
Schist	7	0.81	0.04
Diorite	13	4.67	0.13
Andesite	14	4.82	0.09
Basalt	18	6.72	0.15
Gabbro	22	6.80	0.11

$$\frac{100 \times \text{moles}(\text{K}_2\text{O} + \text{Na}_2\text{O} + \text{CaO} + \text{MgO} - \text{H}_2\text{O})}{\text{moles}(\text{SiO}_2 + \text{TiO}_2 + \text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3 + \text{Cr}_2\text{O}_3 + \text{K}_2\text{O} + \text{Na}_2\text{O} + \text{CaO} + \text{MgO} - \text{H}_2\text{O})}$$

Pay no attention
to the element
behind the screen



Tim Crews
Environmental Studies Program
Prescott College
Prescott, Arizona

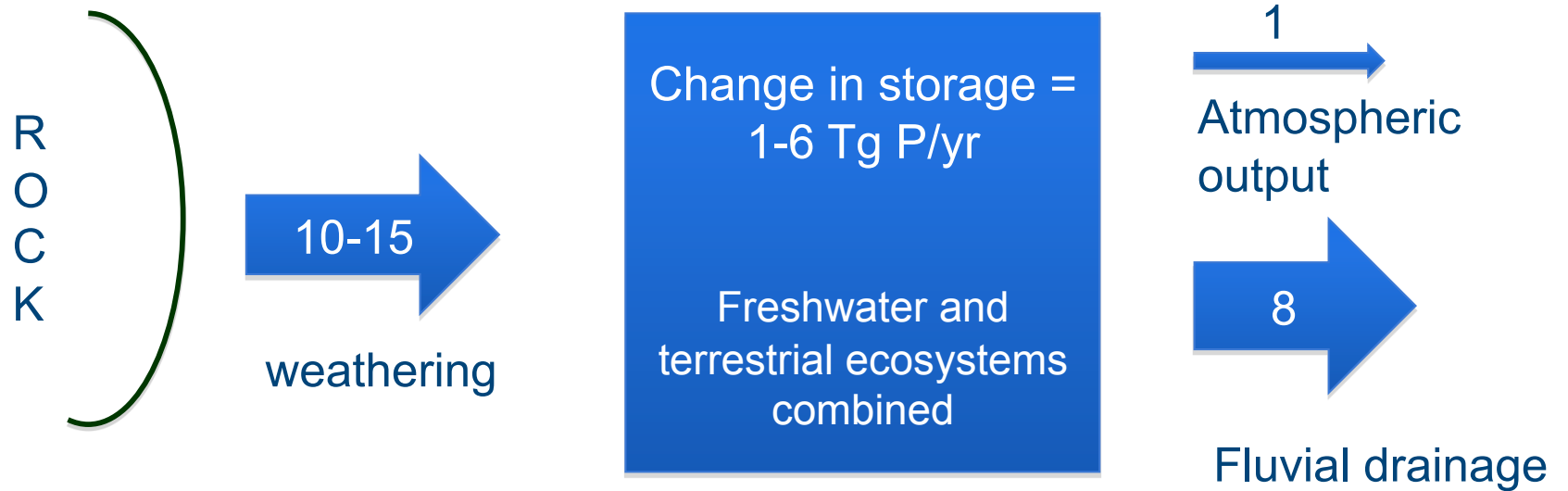








PREINDUSTRIAL GLOBAL TERRESTRIAL P FLUX



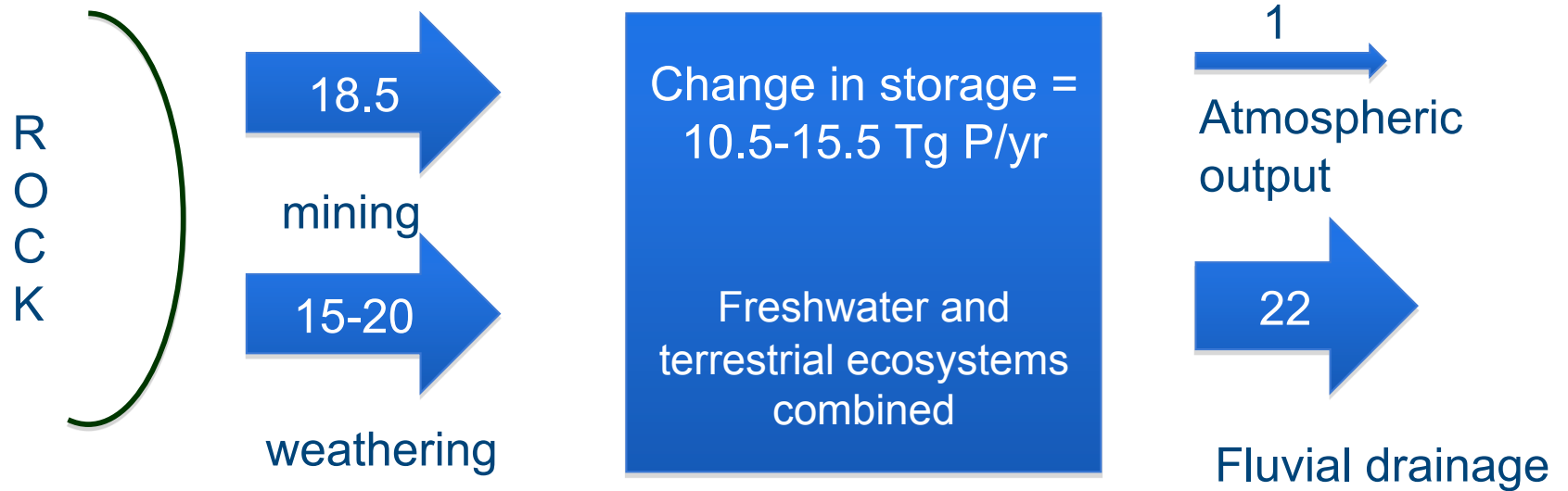
Change in storage =

$$10 - (1 + 8) = 1$$

to

$$15 - (1 + 8) = 6$$

CURRENT GLOBAL TERRESTRIAL P FLUX



Change in storage =

$$(15 + 18.5) - (1 + 22) = 10.5$$

to

$$(20 + 18.5) - (1 + 22) = 15.5$$

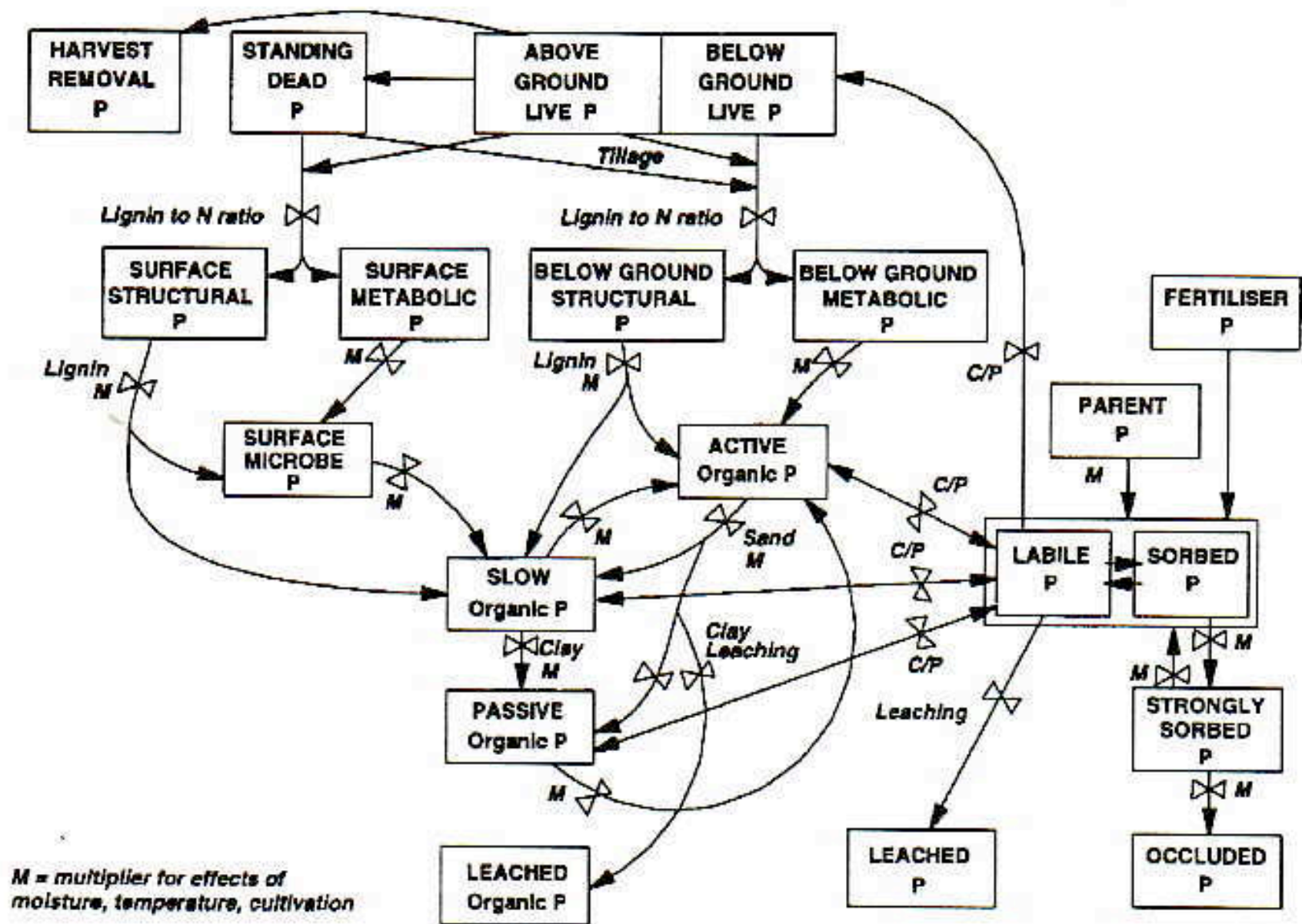


Figure 8. The pools and flows of phosphorus in the CENTURY model

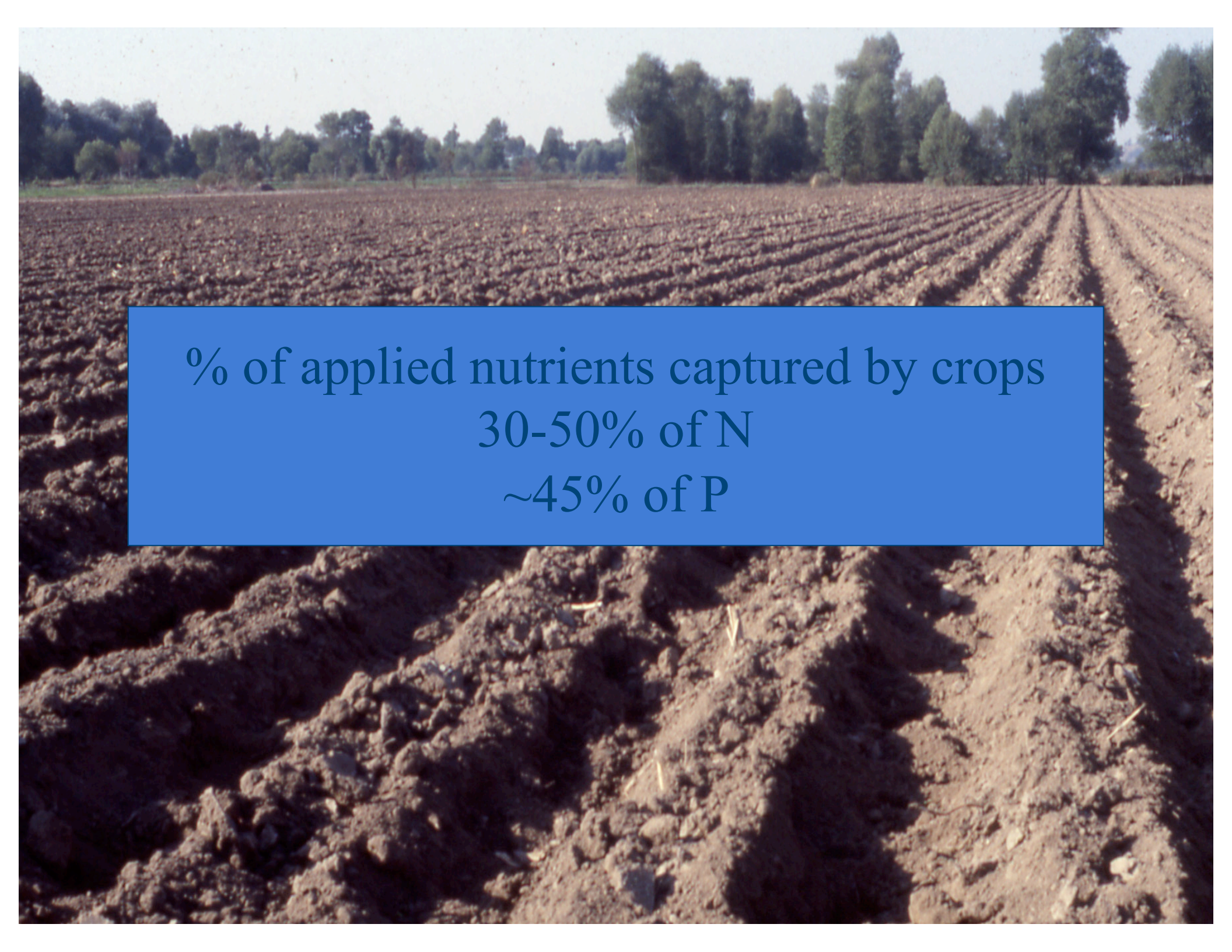
P FERTILIZER
(ORGANIC OR INORGANIC)



SOIL



CROP



% of applied nutrients captured by crops
30-50% of N
~45% of P

P FERTILIZER
(ORGANIC OR INORGANIC)



Clay mineralogy

High sorption capacity:

Amorphous aluminosilicates

Fe oxides

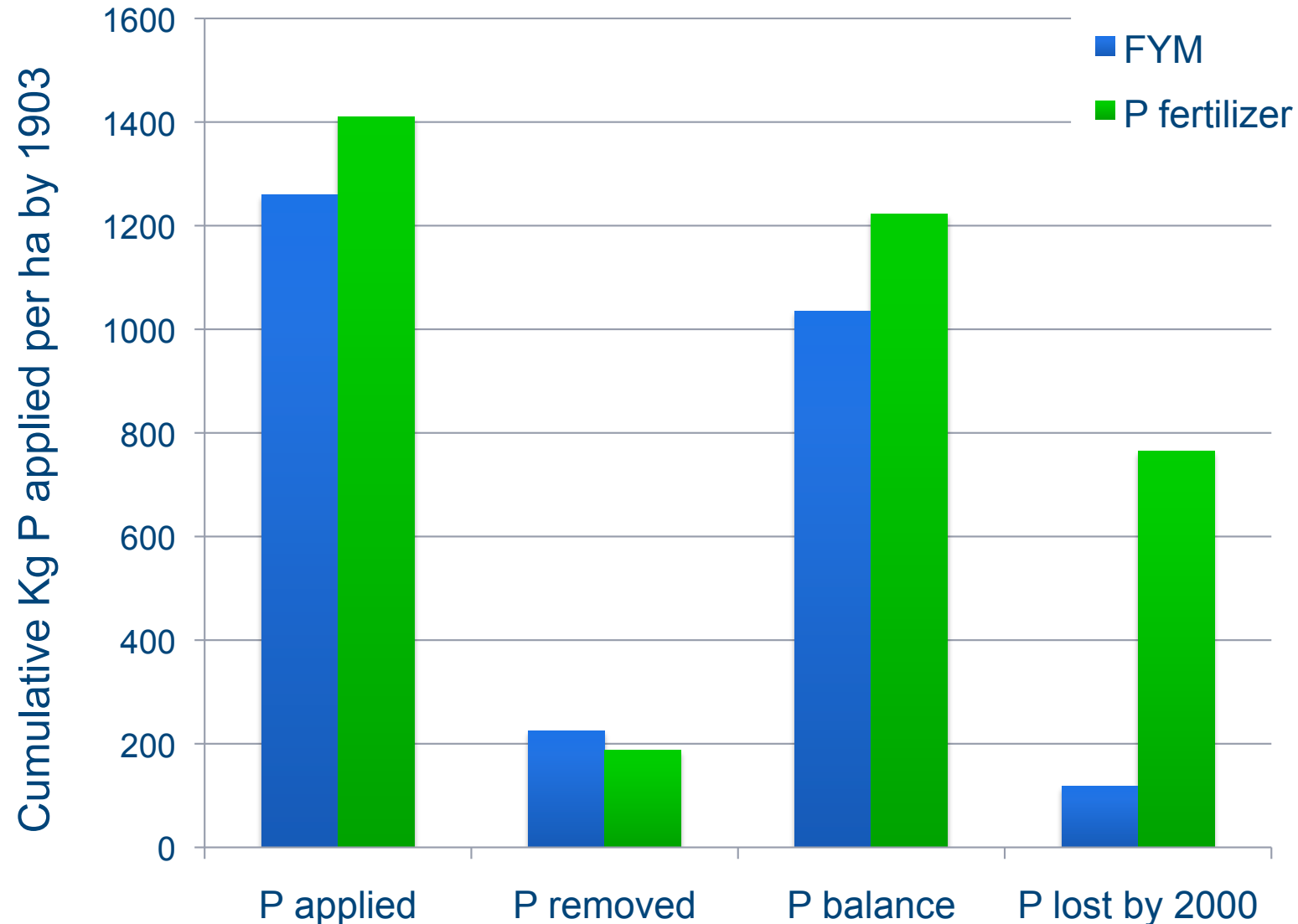
Al oxides

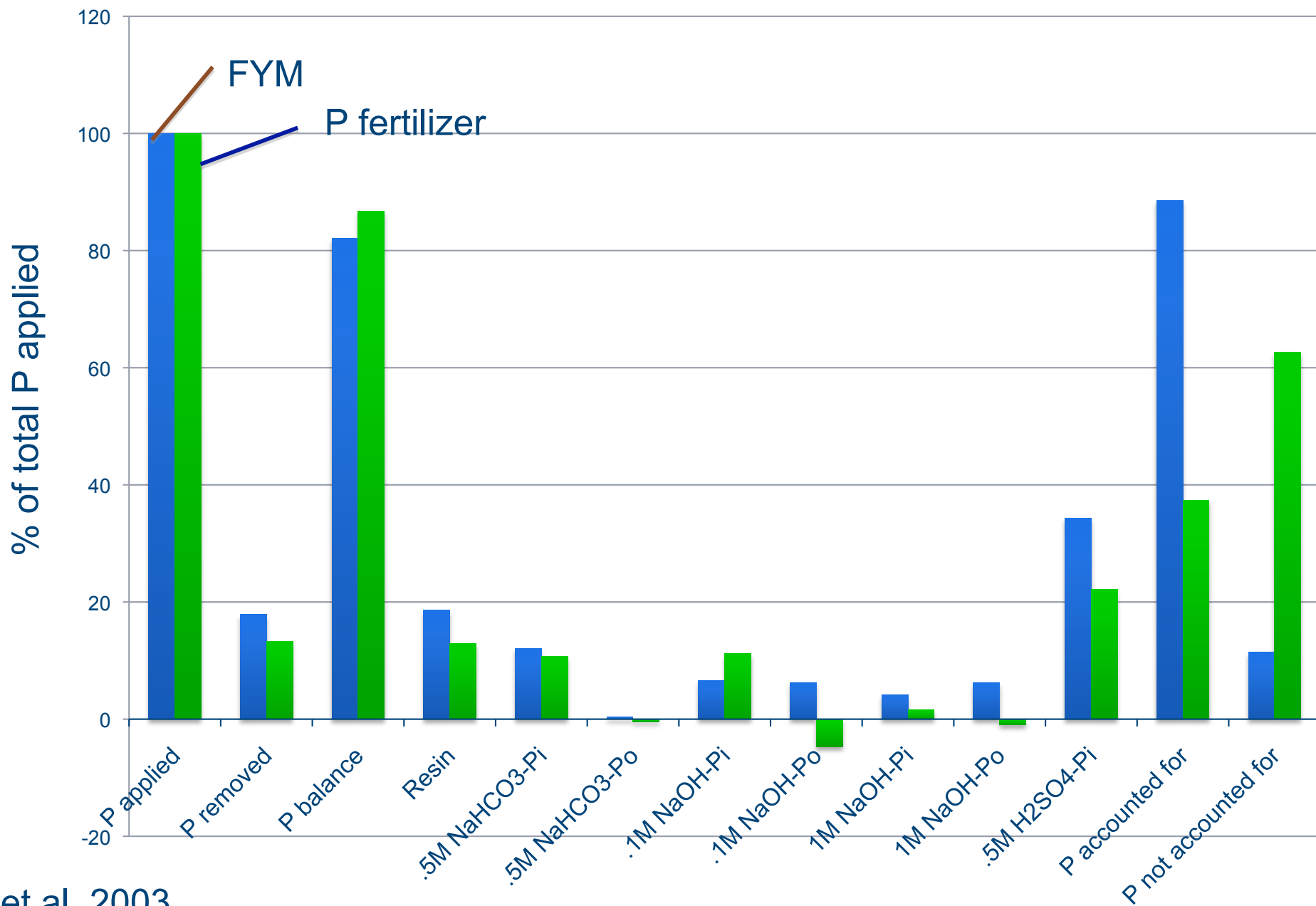
Moderate sorption capacity:

Crystalline aluminosilicates

Humic compounds

Cumulative P additions as Farm Yard Manure (FYM) or P fertilizer by 1903
And P removal in wheat, potato and barley harvests to present at Exhaustion
Land Plots—Rothamsted Research, U.K. (Blake et al. 2003)





P FERTILIZER
(ORGANIC OR INORGANIC)



SOIL



CROP

P-enriched organic and inorganic particles

Water and wind erosion



Leaching of Dissolved Po or Pi



wetland



river



lake

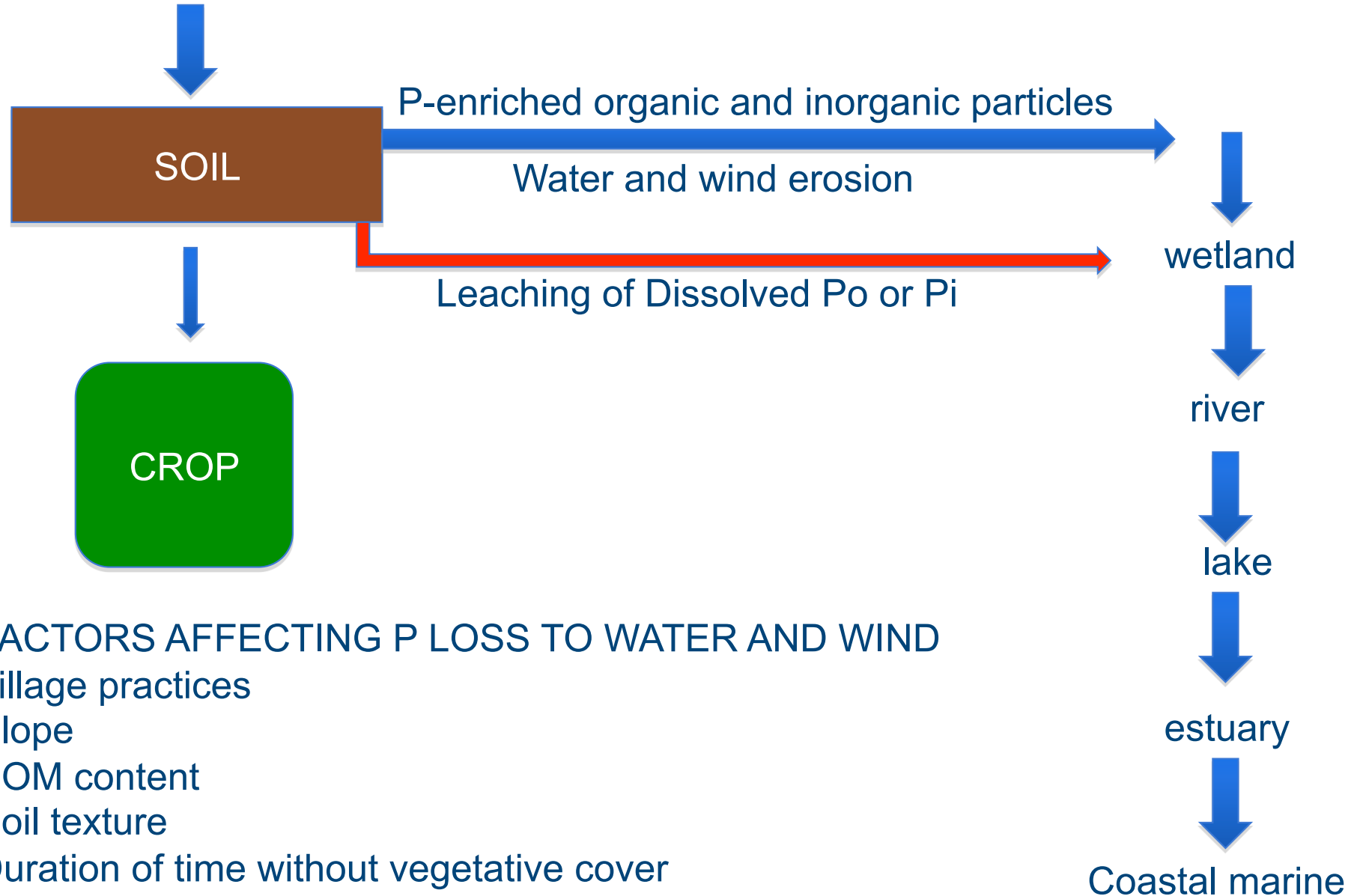


estuary



Coastal marine

P FERTILIZER
(ORGANIC OR INORGANIC)



FACTORS AFFECTING P LOSS TO WATER AND WIND

Tillage practices
Slope
SOM content
Soil texture
Duration of time without vegetative cover
Intensity of rainfall events

Table 5. Effect of fertilizer P application on the loss of P in surface runoff.

Land use	Fert. Appld.	P losses				Fert. loss	
		Solub.	Partic.	Solub.	Partic.	Solub.	Partic.
		----- kg ha ⁻¹ y ⁻¹ -----		----- mg l ⁻¹ -----		----- % -----	
Contour corn ¹	40	0.12	0.45	0.19	0.71		
	66	0.15	0.76	0.25	1.27	0.1	1.2
Grass ²	0	0.01	0.20	0.01	0.06		
	75	0.04	0.29	0.03	0.14	0.04	0.1
No till corn for silage ³	0	0.70	1.30	0.23	0.43		
	30	0.80	1.00	0.39	0.49	0.3	+23.1+
No till corn for grain ³	0	1.10	2.20	0.23	0.46		
	30	1.80	1.60	0.57	0.51	2.3	+27.3+
Conventional corn ³	15	0.30	15.10	0.07	3.57		
	30	0.20	17.50	0.11	9.71	+3.3+	16.0
Wheat/fallow ⁴	0	0.20	1.40	0.30	1.80		
	54	1.20	2.90	3.70	7.40	1.9	2.8
Grass ⁵	0	0.50	0.67	0.18	0.24		
	50	2.80	2.74	0.98	0.96	4.6	4.1

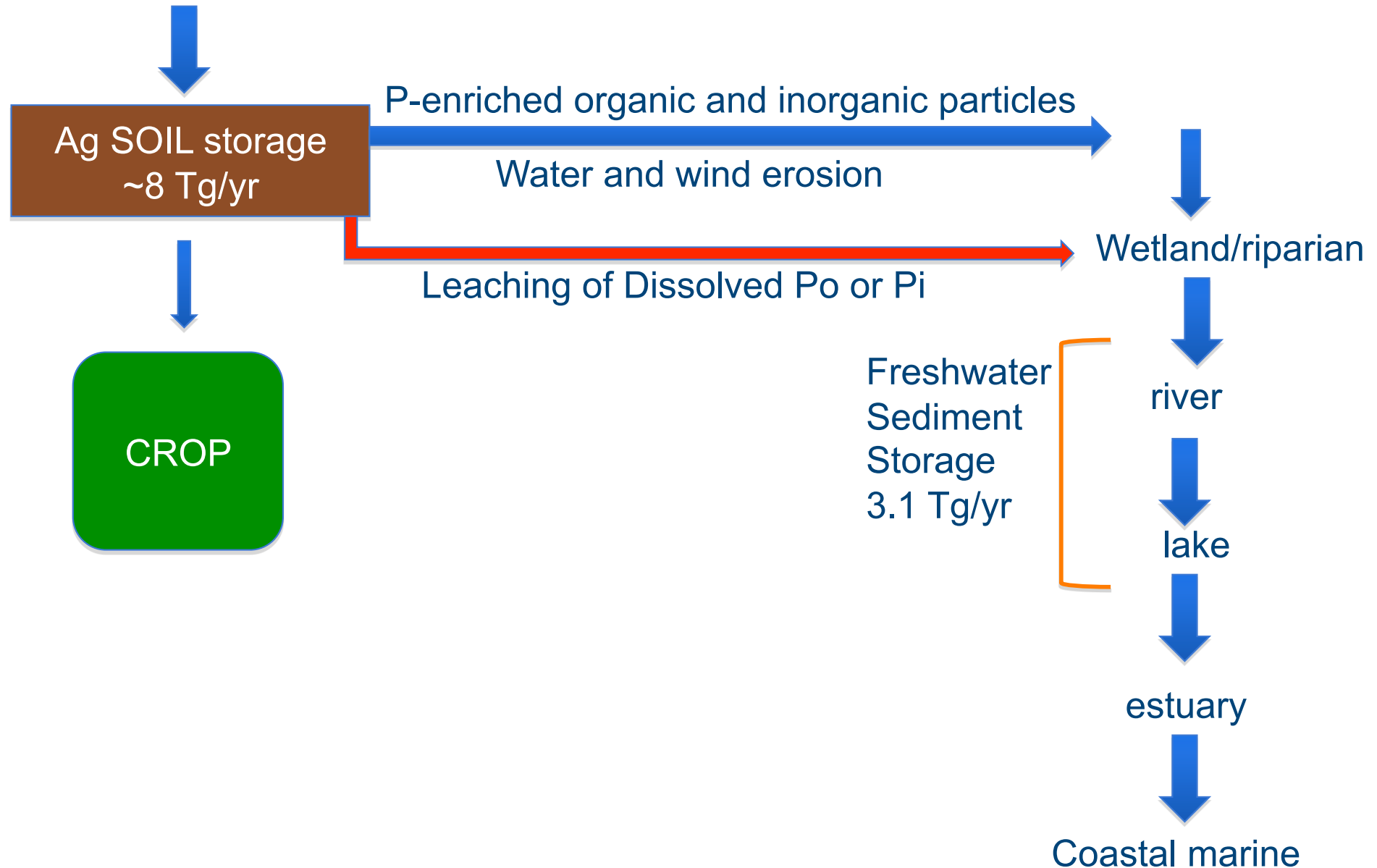
+ Percent decrease in P loss from fertilised compared to check treatment

¹ Burwell et al. (1977) Minnesota; ² McColl et al. (1977) New Zealand;

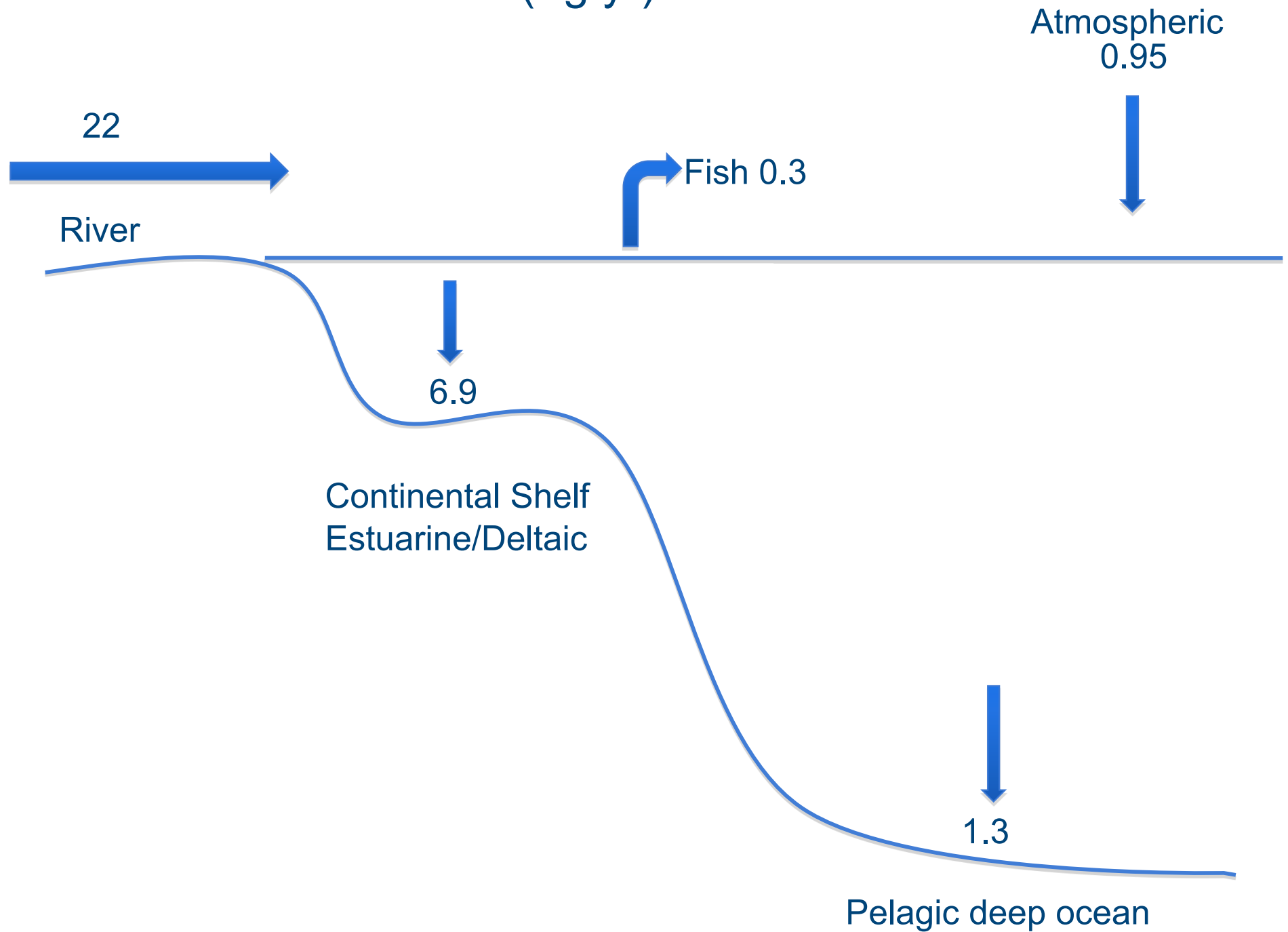
³ McDowell and McGregor (1984) Mississippi; ⁴ Nicholaichuk and Read (1978)

Western Canada; ⁵ Sharpley and Syers (1979) New Zealand

P FERTILIZER
(ORGANIC OR INORGANIC)

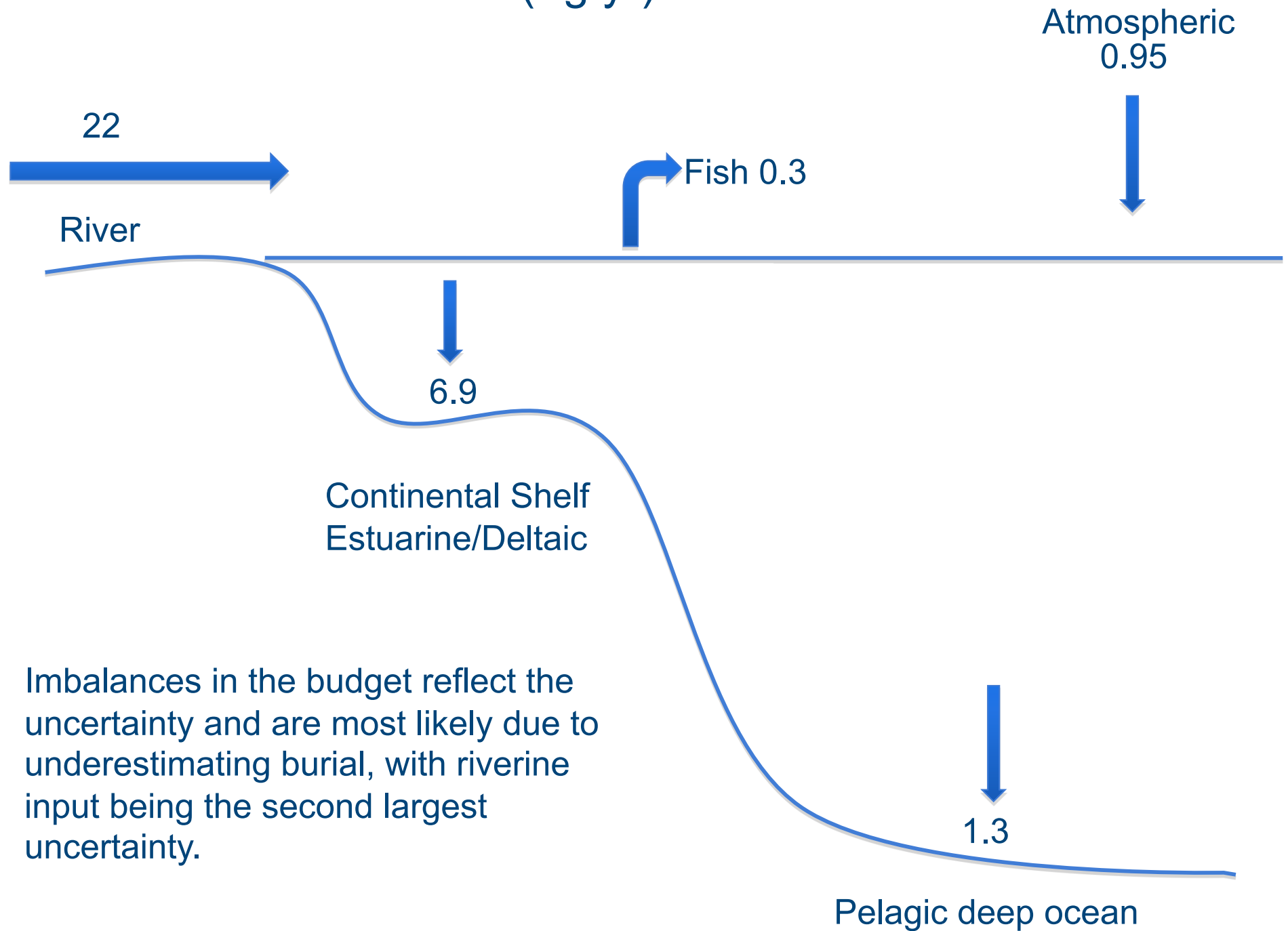


Global ocean P fluxes—(Tg/yr)



(Howarth et al. 1995)

Global ocean P fluxes—(Tg/yr)



Imbalances in the budget reflect the uncertainty and are most likely due to underestimating burial, with riverine input being the second largest uncertainty.

(Howarth et al. 1995)

P FERTILIZER
(ORGANIC OR INORGANIC)



10-90 % of applied P might be taken
Up by the crop

Average ~45%

MANAGEMENT FACTORS

Crop selection

Crop deployment in time and space

Soil pH management (lime)

Alleviation of other crop limiting factors

Mycorrhizal colonization (?)

P FERTILIZER
(ORGANIC OR INORGANIC)



Concentrated Animal Production



Rural to urban food shipments

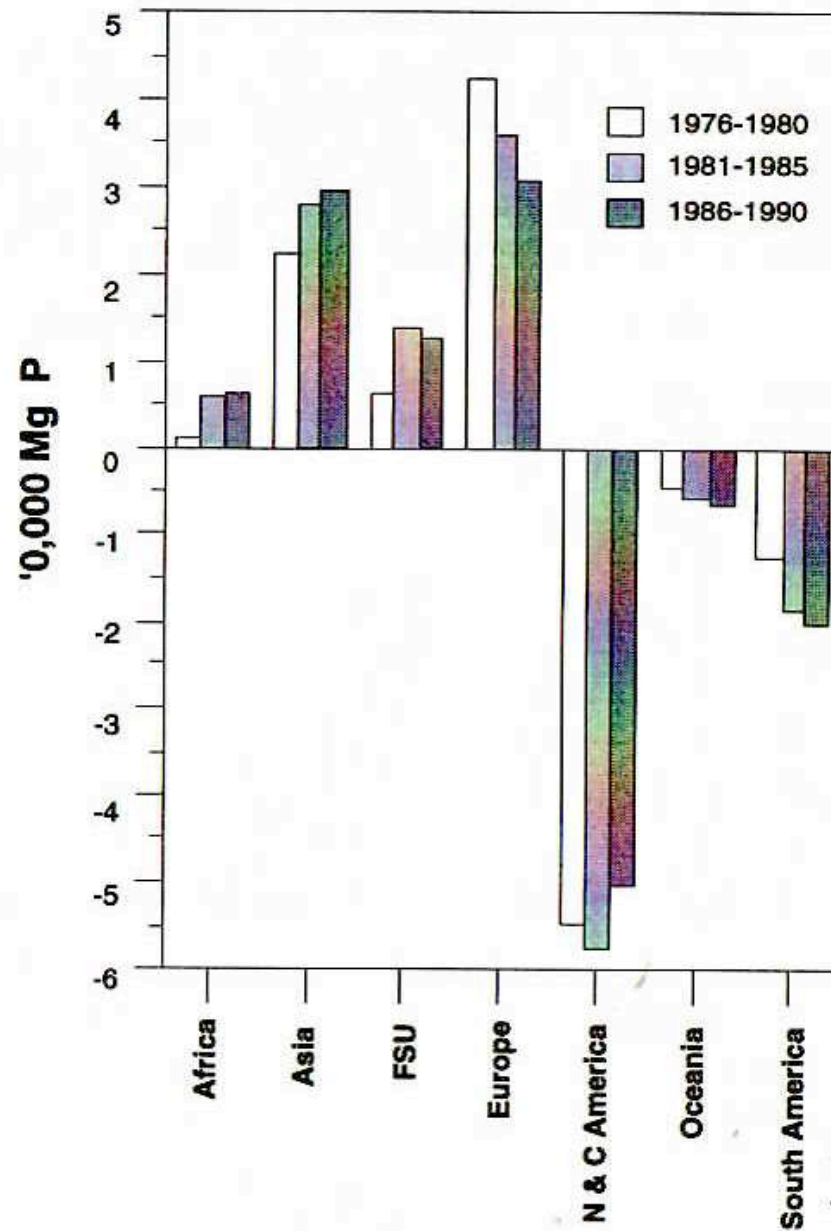
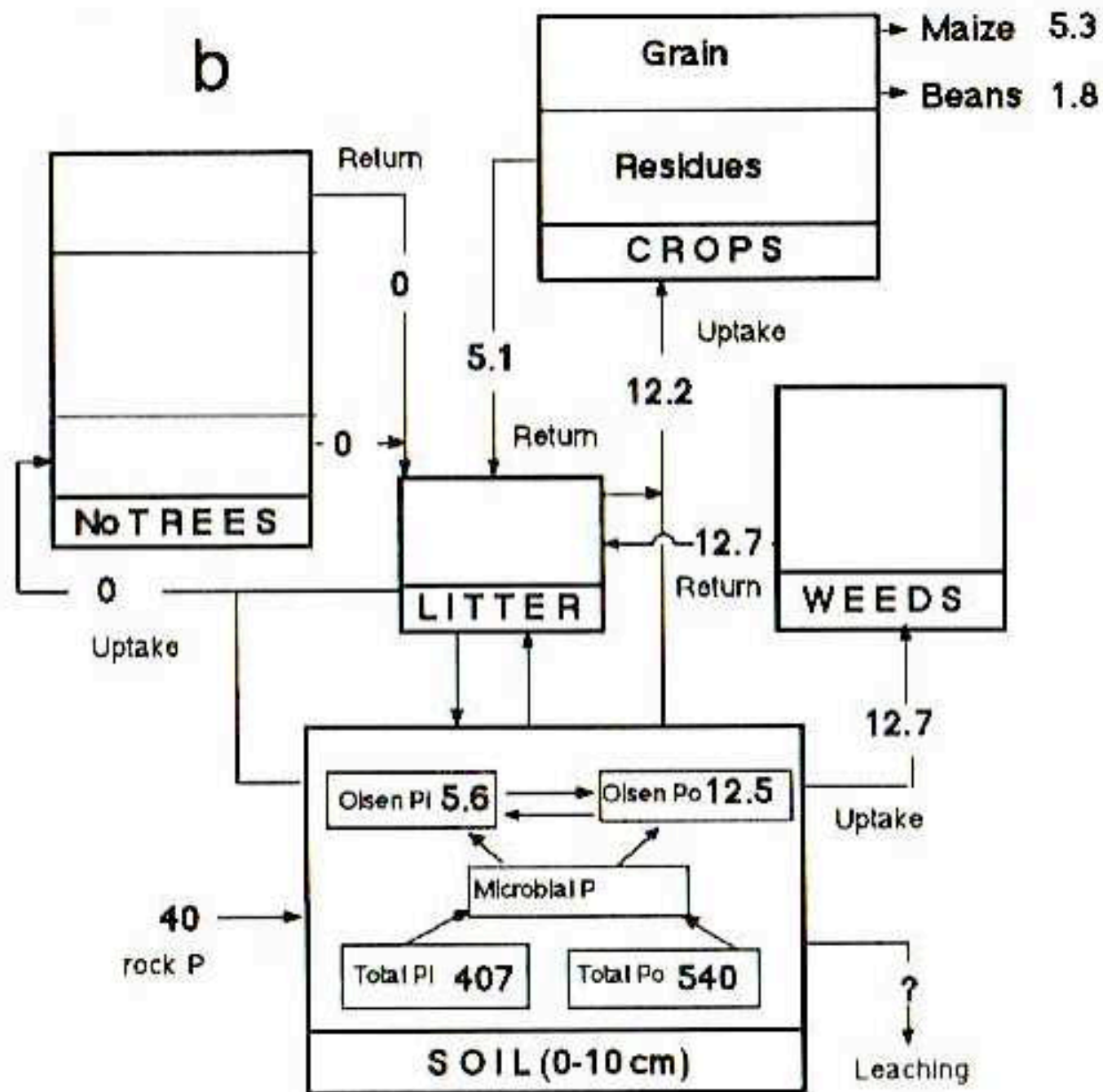


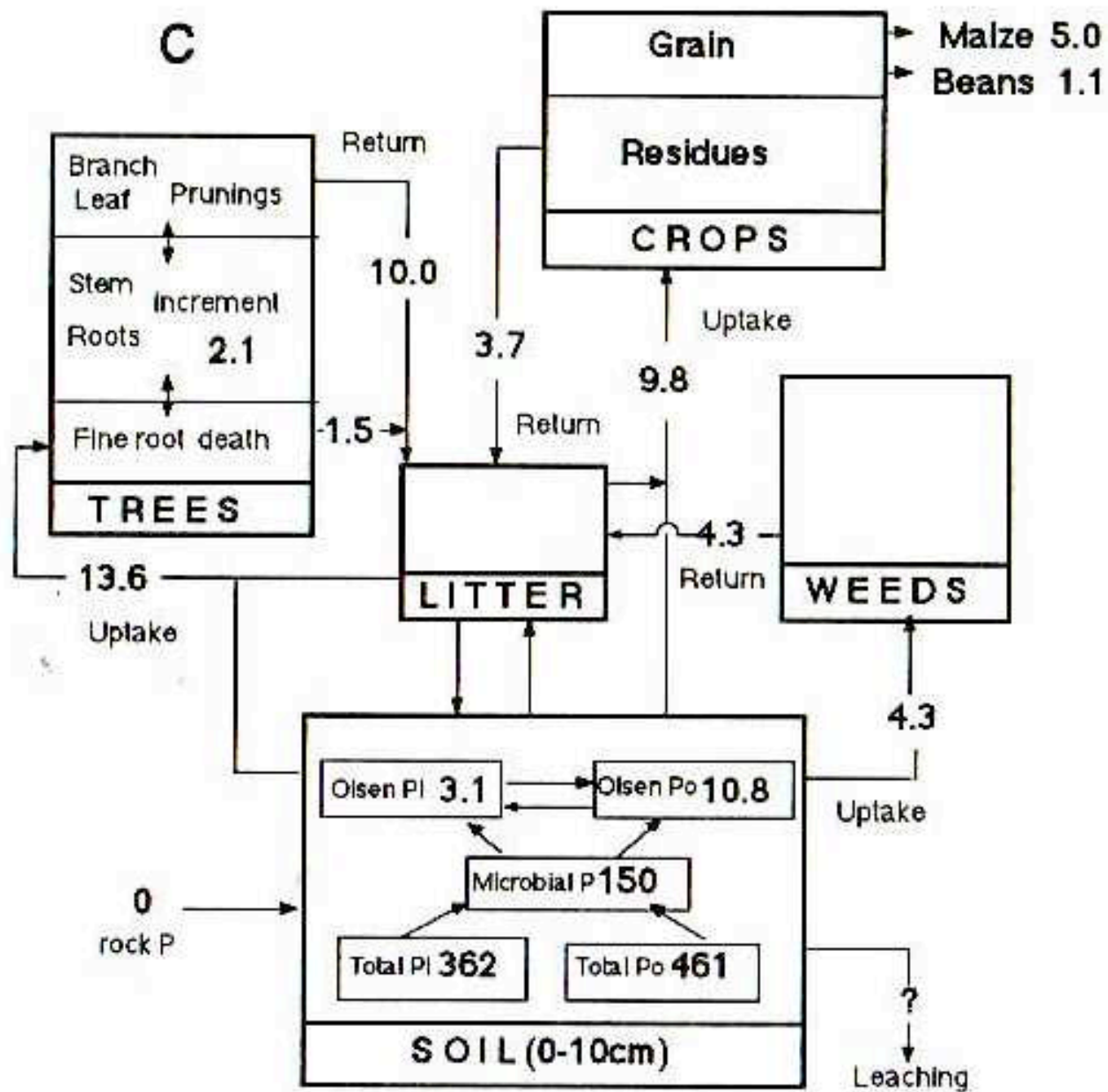
Figure 3. Regional net P balances (total P in imports - total P in exports) for crop commodities averaged over 1976-1980, 1981-1985, and 1986-1990.

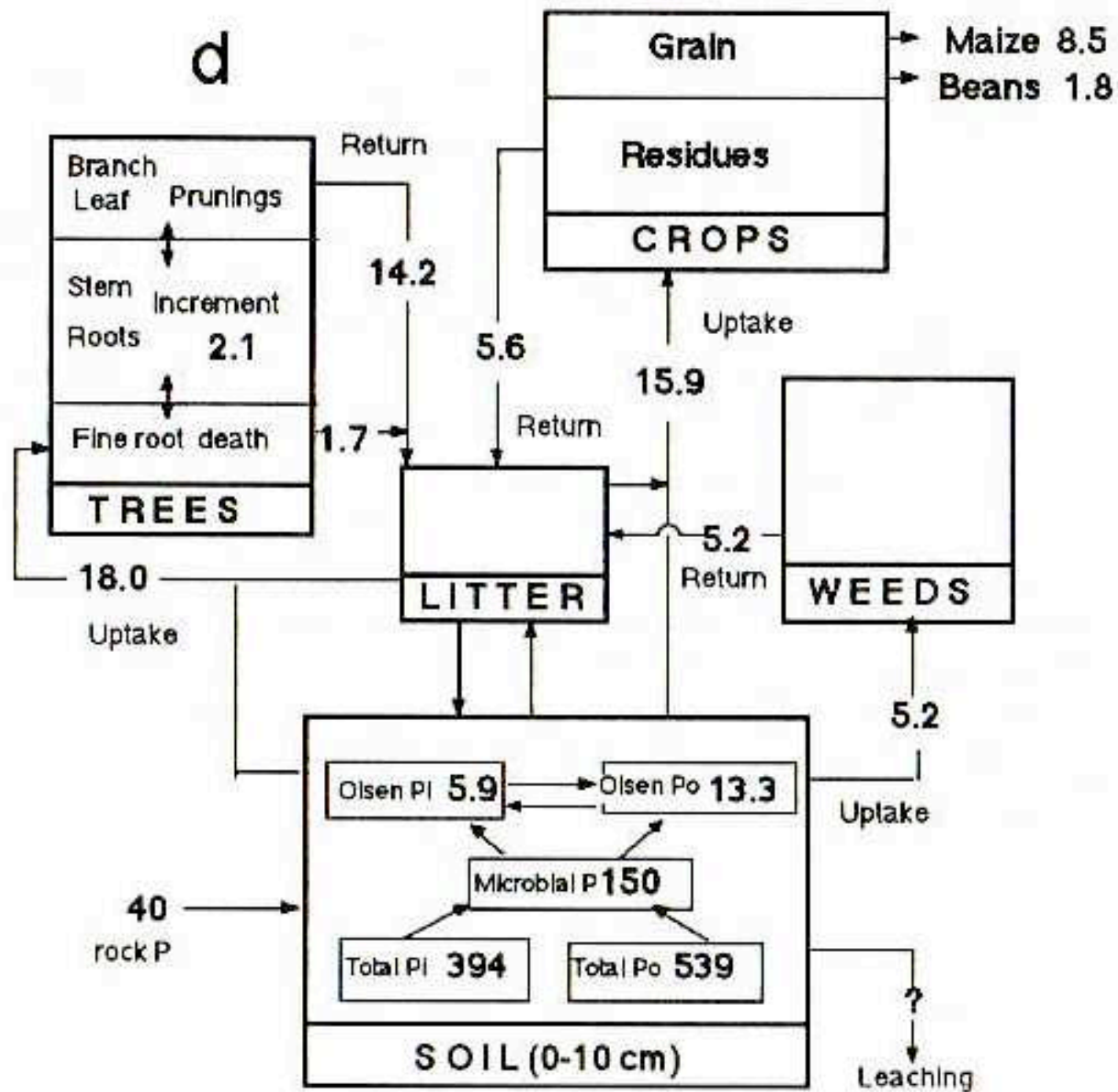
Optimize the utility of P fertilizers, a scarce, non-renewable, non-substitutable and Highly valuable resource, rather than simply try to find the most effective sinks.

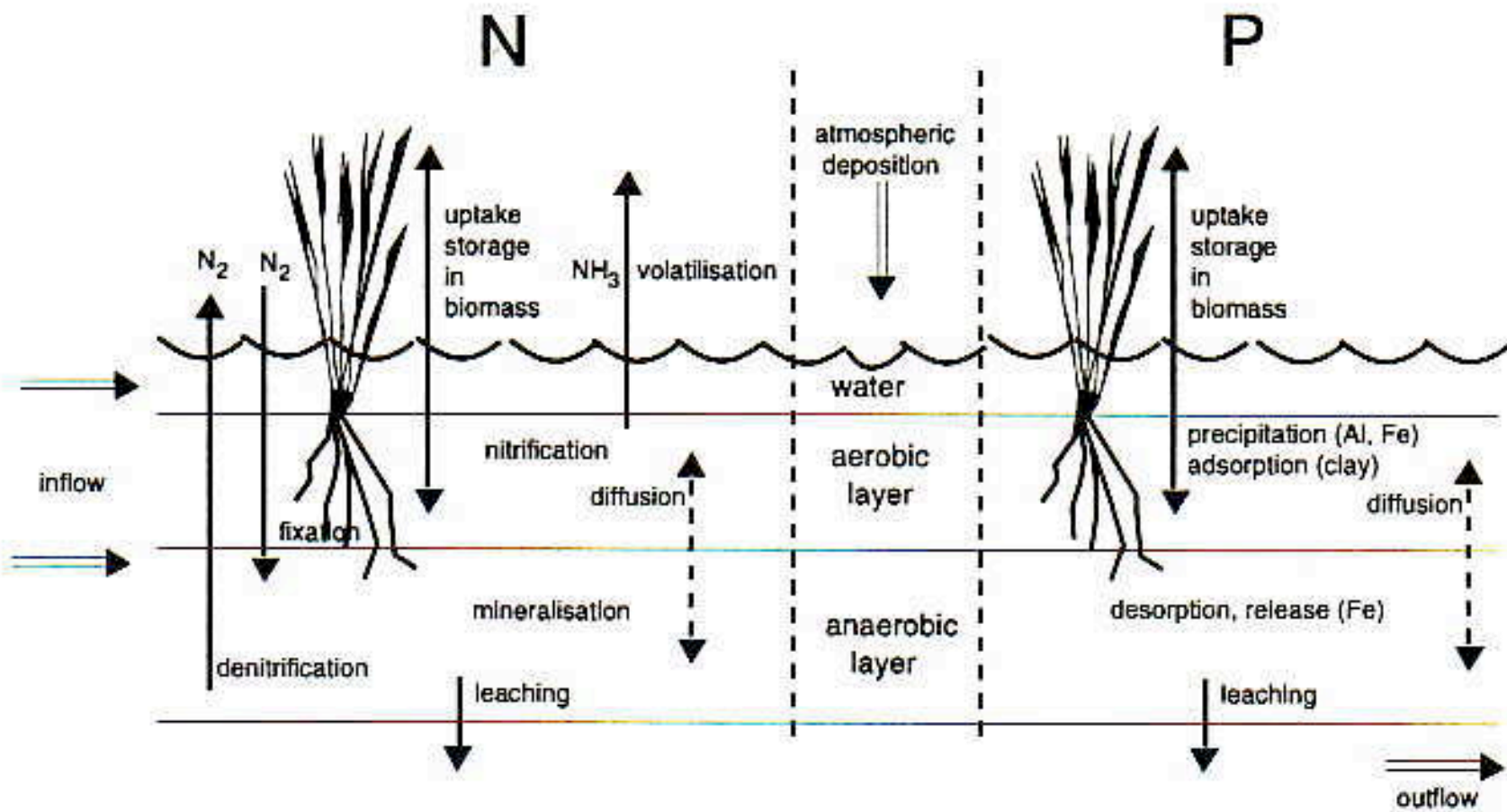
N











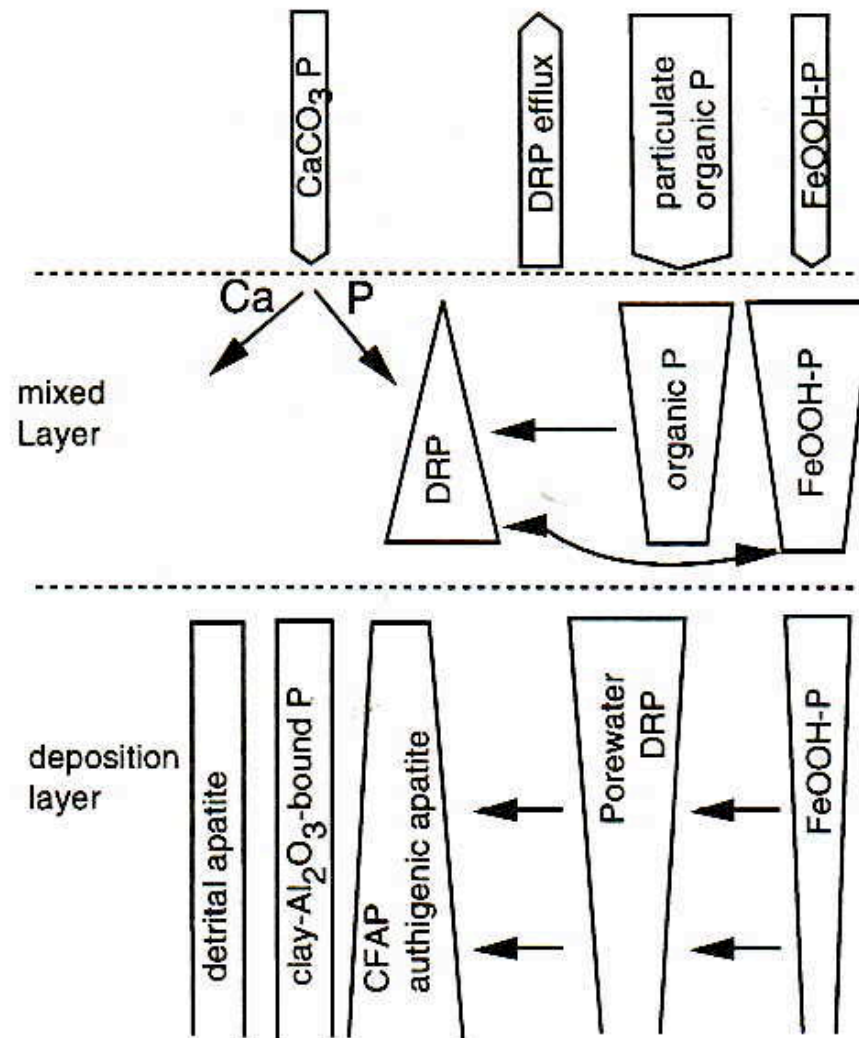


Figure 2. Schematic, general model of P diagenesis in sediments of temperate estuaries and deltaic coastlines. Phosphorus enters sediments primarily in particulate form bound to carbonates and Fe (III) hydroxides and as organic P. Some P also enters sediments as detrital apatite or bound to clays. Some P leaves as dissolved reactive phosphorus (DRP). In the surface mixed layer of the sediments, P is converted into DRP, authigenic apatite, and clay-bound forms; Fe-bound and organic P decrease. Note, however, that the Fe-bound P is important in regulating the flux of DRP to overlying waters (see text). In deeper sediment layers, Fe-bound and organic P further decrease and are converted to apatite.

Table 5. P removed in crops (harvested or grazed), net P fertilisation (all fertilizer P minus P removed in crops), and P in fodder. Net import or export of P in fodder is the difference between P amounts in total fodder used, and potential fodder available from crop removal, 1989.

Country	Crop removal	Net fertilisation	P in fodder		
			Total	Net import	Net export
----- kg P ha ⁻¹ y ⁻¹ -----					
Iceland	0.9	0.7	0.6		0.3
Norway	15.0	15.9	19.2	4.2	
Sweden	15.0	2.3	11.0		4.1
Finland	11.9	20.6	10.7		1.2
Denmark	21.7	15.5	28.3	6.6	
Ireland	12.2	8.4	11.8	0.4	
United Kingdom	14.1	5.4	11.9		2.2
Netherlands	21.8	57.2	81.0	59.2	
Belgium+Luxemb.	21.5	41.5	49.3	27.8	
France	16.6	14.6	12.0		4.6
W. Germany	20.2	24.4	29.2	9.0	
E. Germany	17.7	25.5	22.1	4.4	
Poland	12.0	14.0	10.8		1.2
Czechoslovakia	16.5	26.6	15.3		1.2
Switzerland	13.1	10.6	19.7	6.6	
Austria	13.5	6.3	13.6	0.1	
Hungary	12.8	15.5	11.3		1.5
Portugal	7.8	7.5	7.5		0.3
Spain	7.3	6.5	5.7		1.6
Italy	12.1	17.0	13.3	1.2	
Yugoslavia	7.6	4.6	6.2		1.4
Romania	7.6	7.5	5.5		2.1
Bulgaria	9.5	12.6	7.5		1.9
Albania	7.9	7.5	6.1		1.8
Greece	5.7	7.3	3.7		2.0
Europe	12.3	12.6	12.3		0.1
EC - 12	13.2	12.8	13.7	0.5	

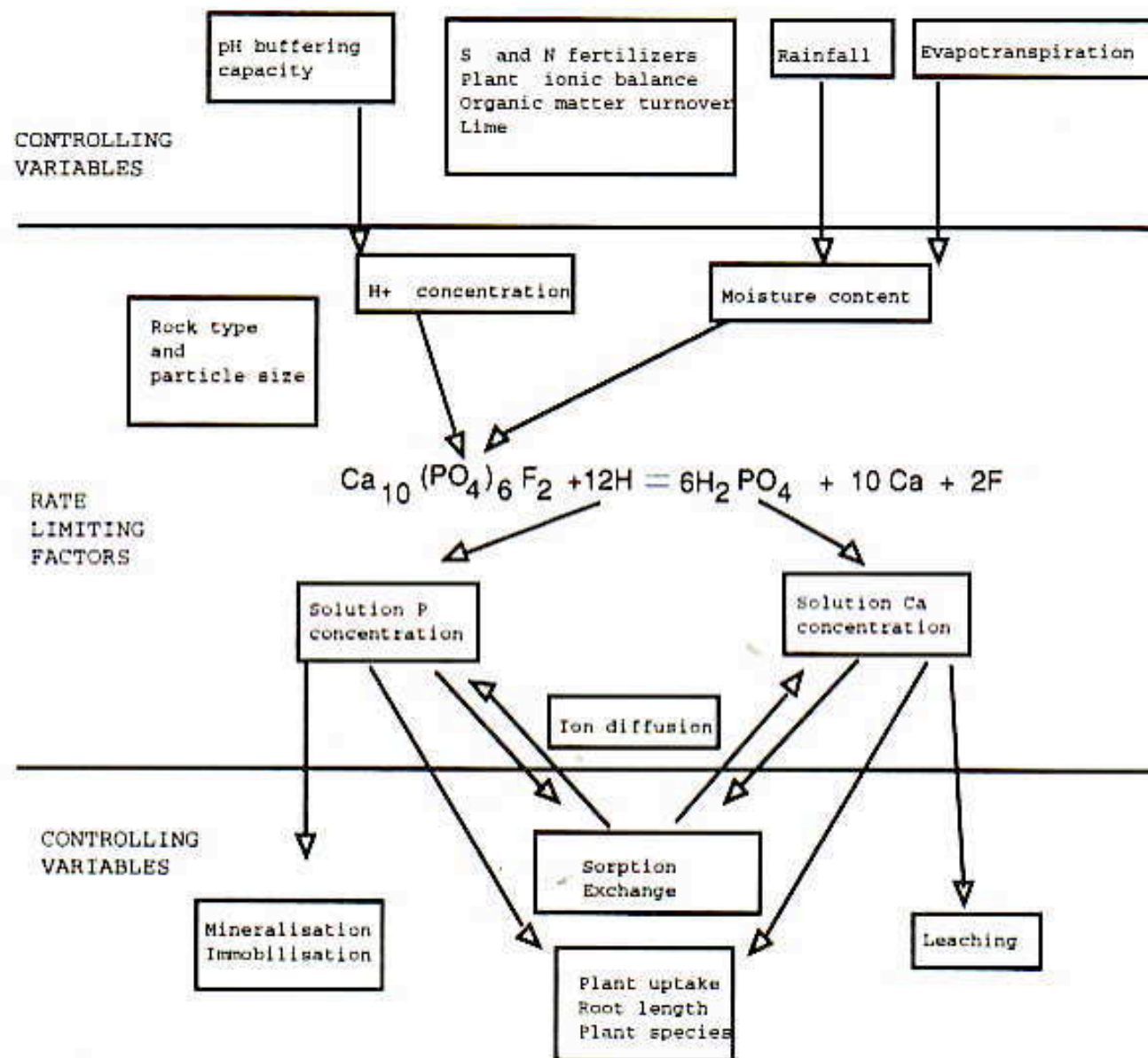


Figure 8. Schematic diagram showing the rate-limiting factors and variables determining phosphate rock dissolution

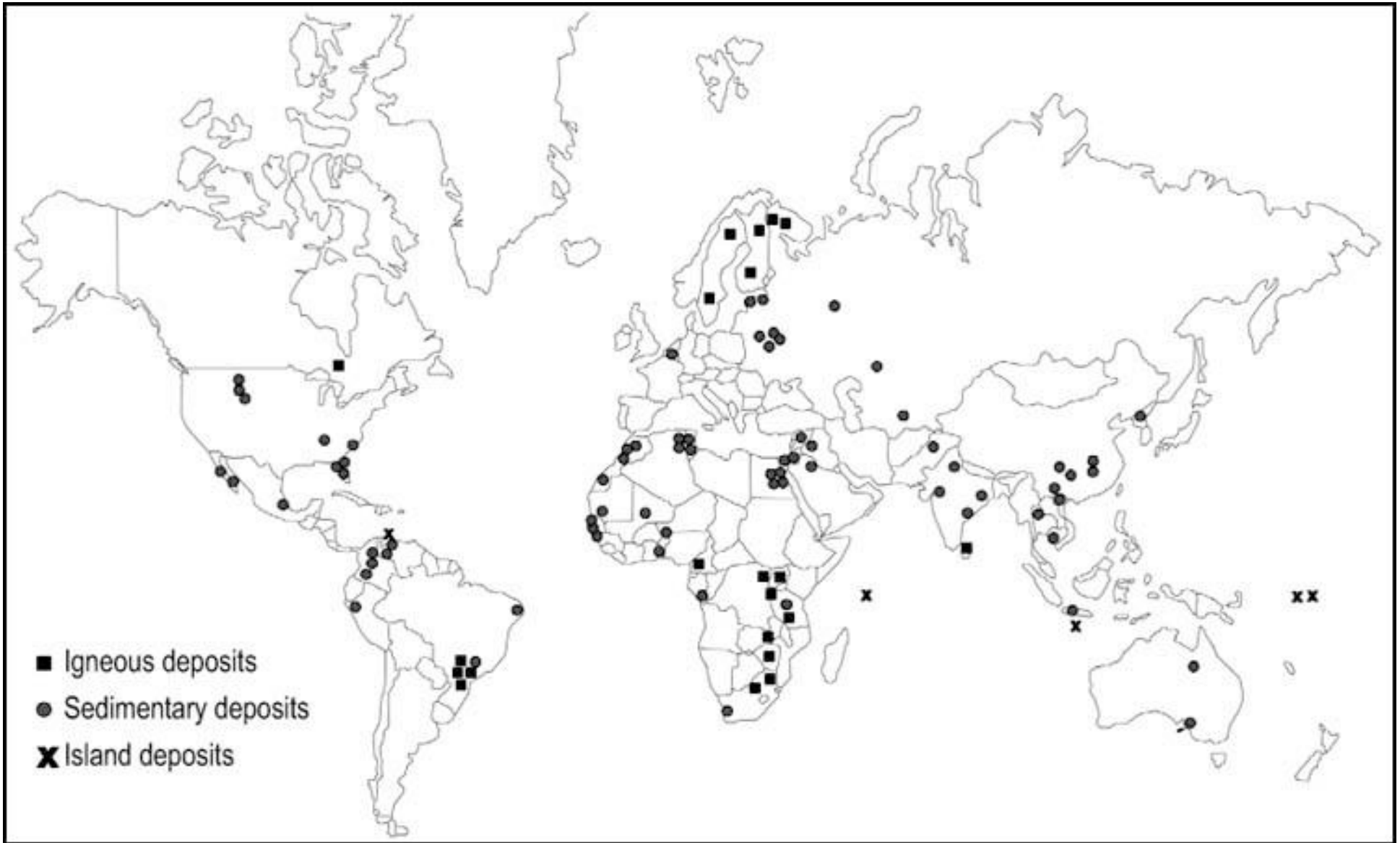
Table 4. P balances and changes in AL-soluble P content of soils between 1979 and 1982 on different slopes in the Zala River watershed (Debreczeni & Sisák, 1990).

Slope	P balance	Changes in AL-P
%	kg ha ⁻¹ y ⁻¹	mg kg ⁻¹
0-5	44	18.2
5-10	40	10.2
10-15	37	9.5
15-20	35	6.5
20-30	31	5.3

52 percent of known unexploited global P reserves



Economic and potentially economic phosphate deposits of the world



FAO 2004

P inputs

kg ha⁻¹ yr⁻¹

Soil mineral weathering	0.05 - 1.0 (5.0 possible)
Atmospheric deposition	0.07 - 1.7
Total	0.12 - 2.7

Other P inputs that may be significant in particular ecosystems include: pollen, volcanic ash, fossil fuel combustion (e.g., fly ash), forest fires, ocean spray

Newman (1995)

P balance for a typical modern wheat farm in Great Britain

P inputs

kg ha⁻¹ yr⁻¹

Soil mineral weathering	0.3	
Atmospheric deposition	net 0	
Fertilizer	25	
Total		25.3

P outputs

kg ha⁻¹ yr⁻¹

Leaching	0.4	
Soil erosion by water	0.5	
Soil erosion by wind	net 0	
Harvested grain	19	
Straw	3	
Total		22.9

Newman (1997)

The U.S. supplies between 38-50% of the world P market.
Florida alone supplies 75% of U.S. and 25% of the world's
P market.

An aerial photograph of a large-scale phosphate mining operation. A long, elevated conveyor system runs across a deep, excavated pit, transporting material from a processing building with a white and green roof. The surrounding landscape is arid and shows signs of industrial activity, with roads and other infrastructure visible in the background.

Kurt Grimm, Dept. Earth and Ocean Sciences at the
University of British Columbia predicts the U.S. will be a net
importer of P within 30 years

Photo by J. Schardt
2002
Florida D.E.P.

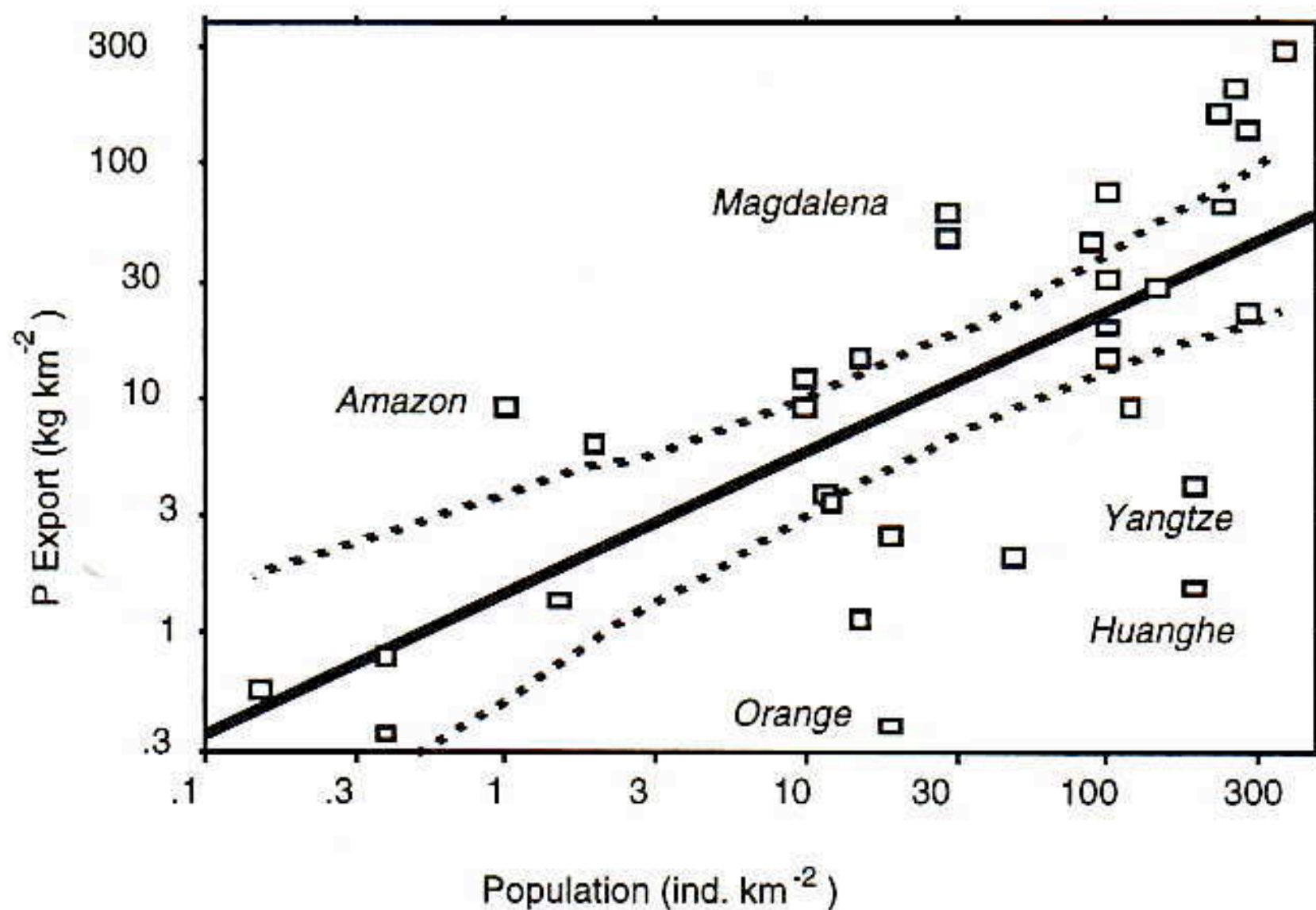
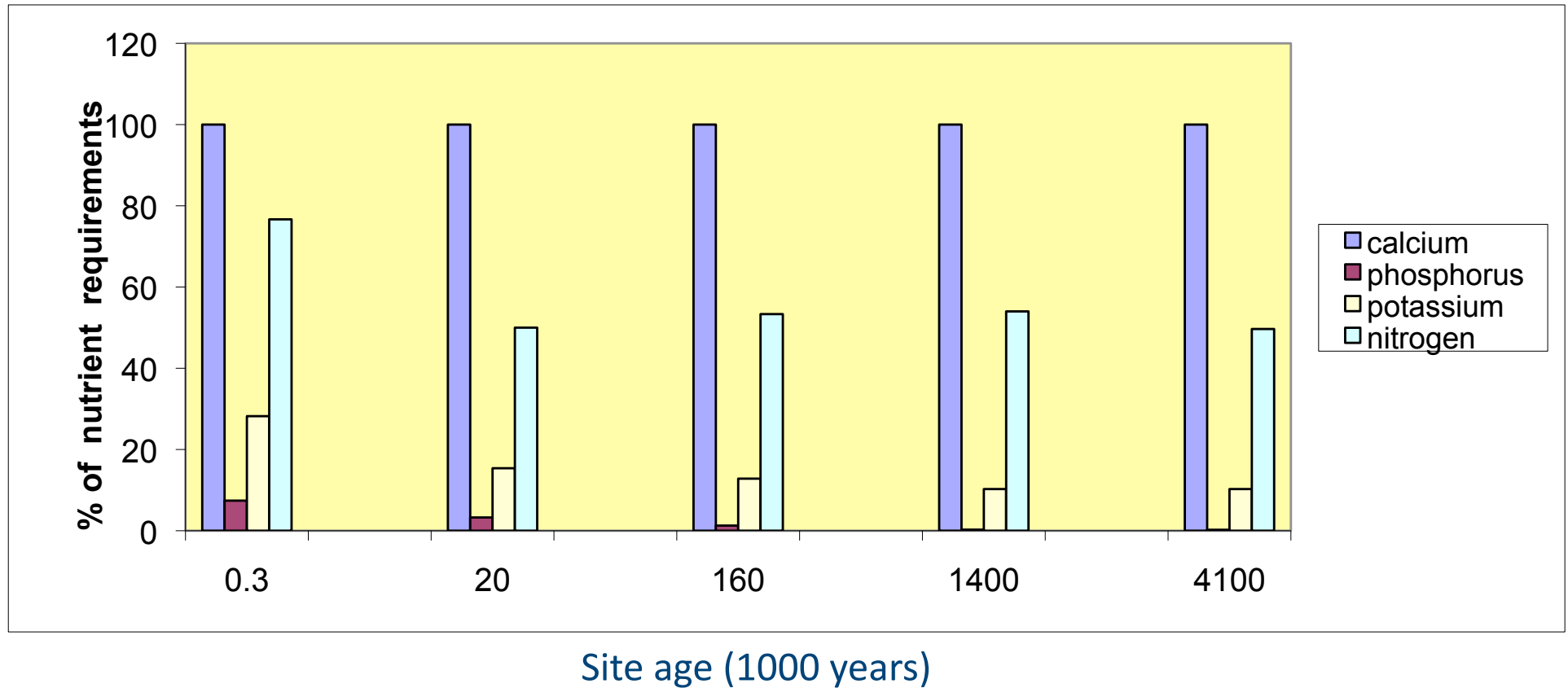
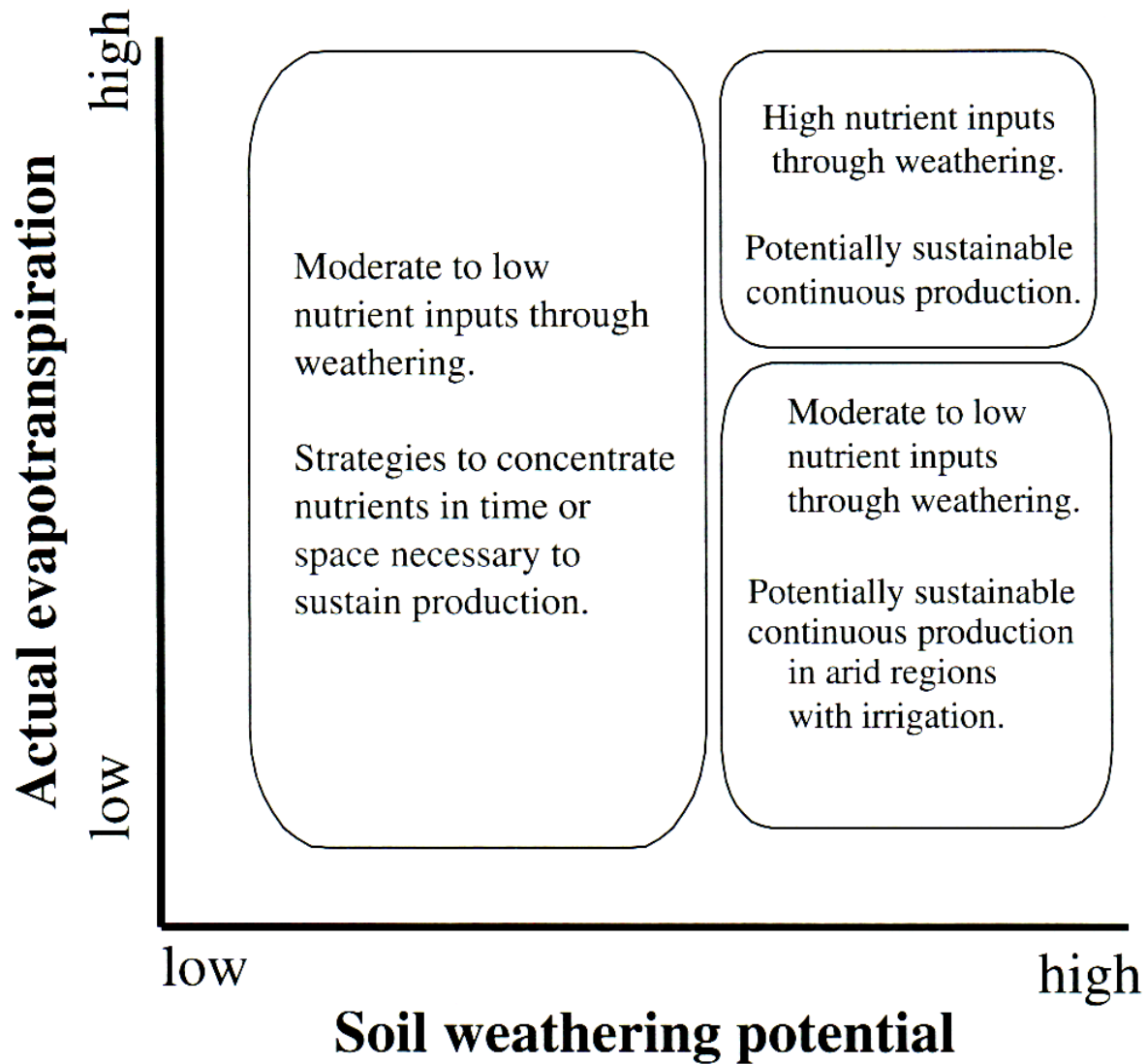


Figure 1. Relationship between population density in the watershed and export of SRP in river water. Data are for 32 major rivers (Table 1).

Percentage of sweet potato nutrient uptake¹ satisfied by
endogenous nutrient supplies estimated by Vitousek (2004)



¹Sweet potato crop yielding 9 Mg ha⁻¹ (fresh yield) grown with no fertilizer inputs
in Papua, New Guinea (Hartemink et al. 2000)



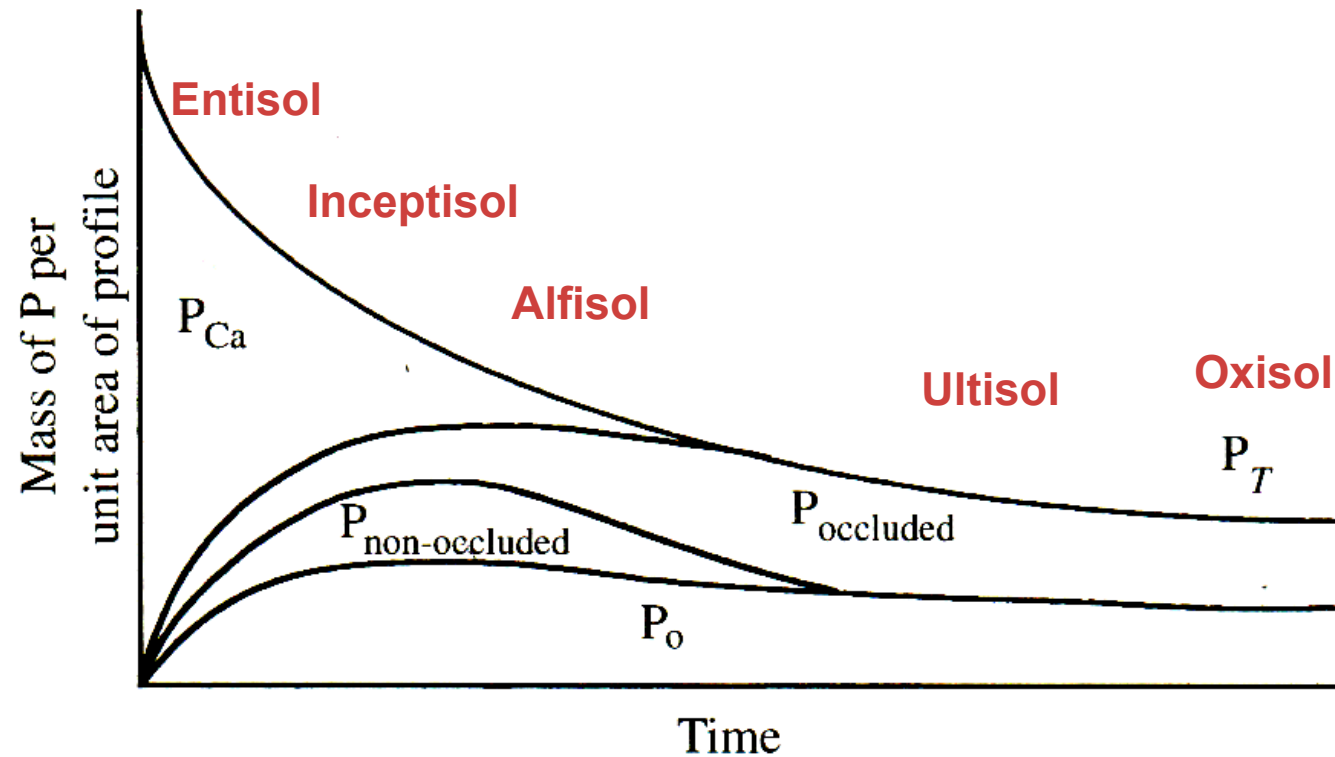
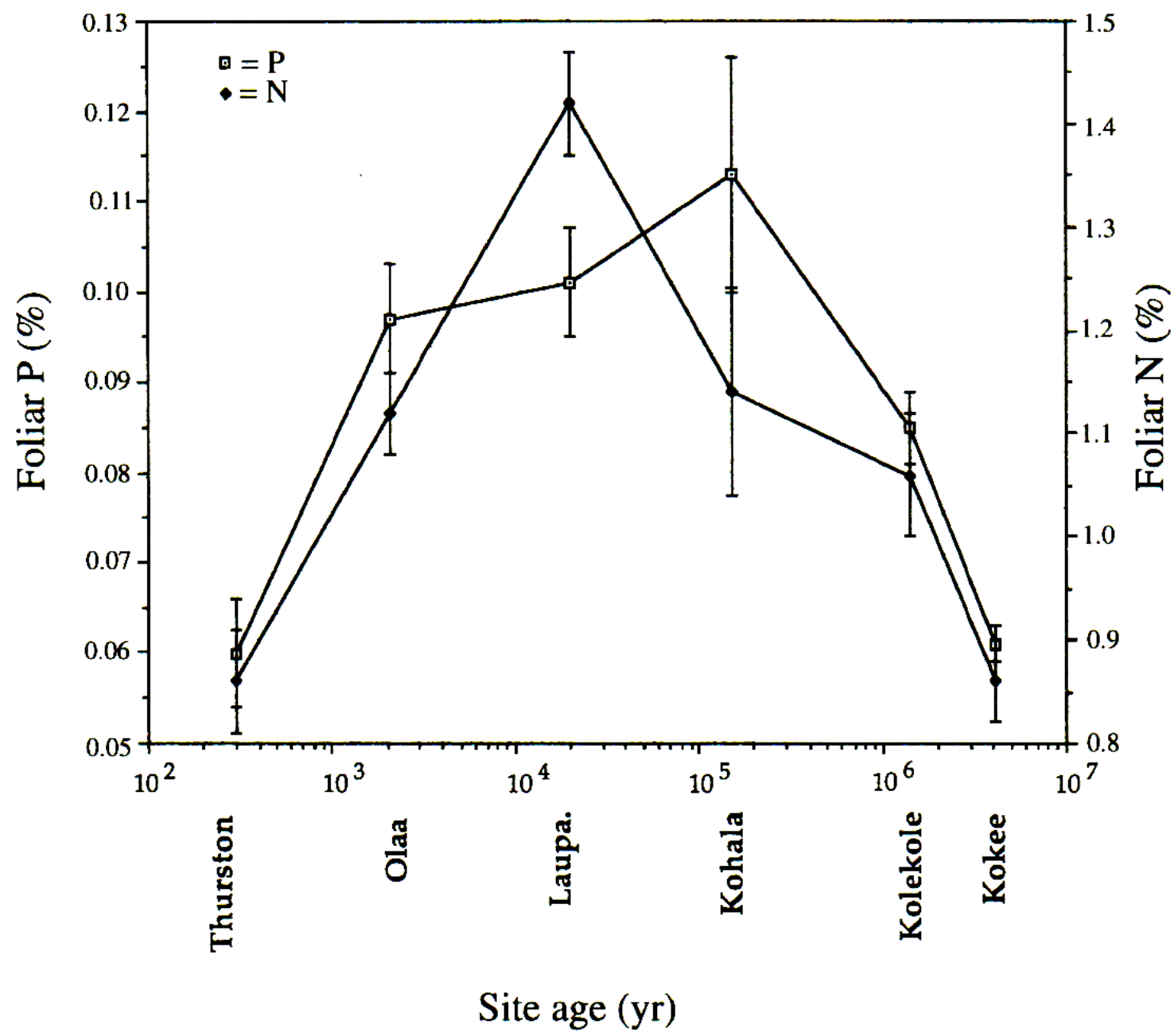


FIG. 1. Walker and Syers' (1976) diagram of P transformations with time. P_{Ca} = calcium phosphates, P_T = total phosphorus, P_o = P bound to organic matter.



P balance for a typical modern wheat farm in Great Britain

P inputs

kg ha⁻¹ yr⁻¹

Soil mineral weathering	0.3		
Atmospheric deposition	net 0		
Fertilizer		25	
Total			25.3

P outputs

kg ha⁻¹ yr⁻¹

Leaching		0.4	
Soil erosion by water	0.5		
Soil erosion by wind		net 0	
Harvested grain	19		
Straw		3	
Total			22.9

Newman (1997)