

The terrestrial P cycle: Ideas and opportunities



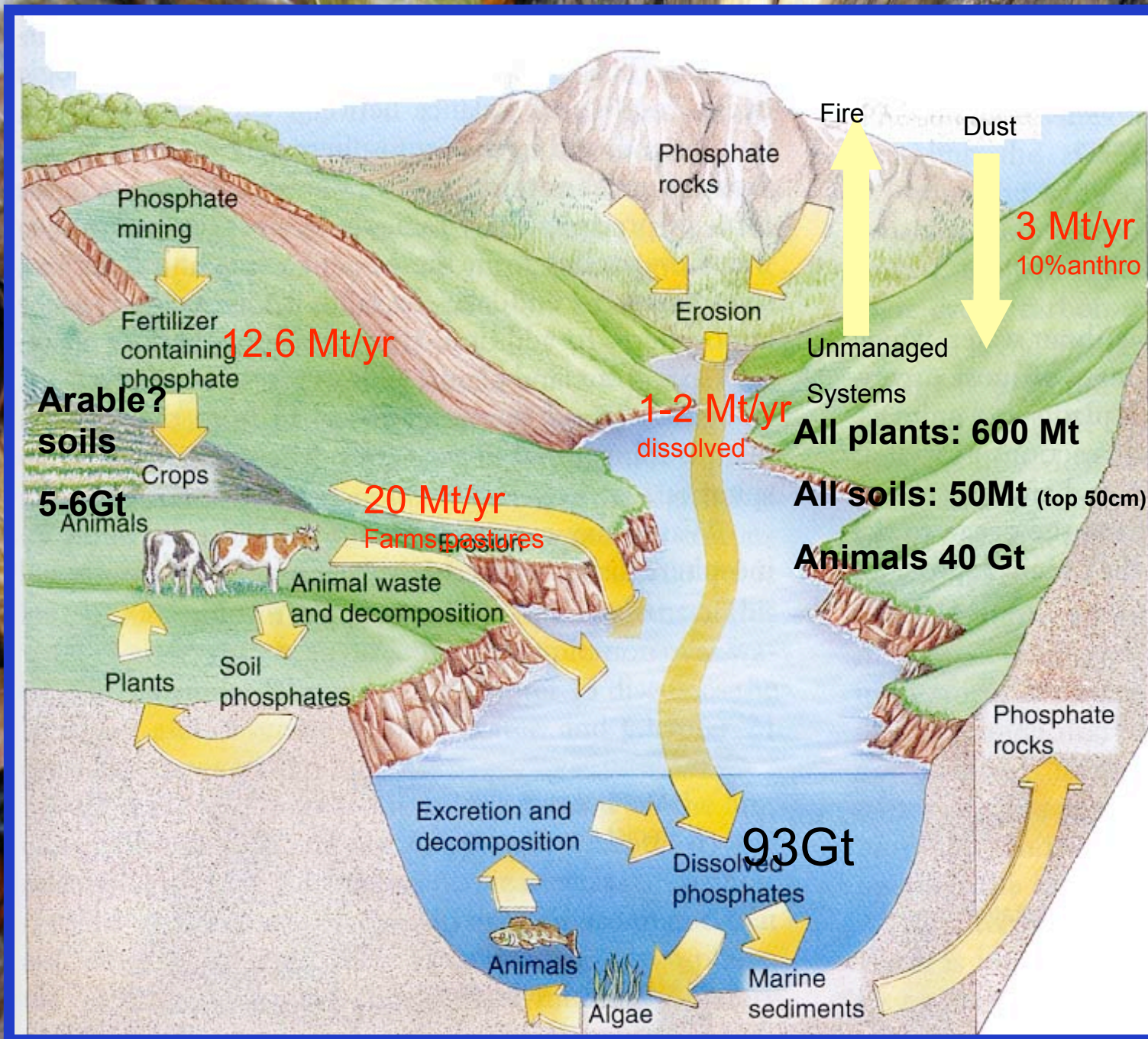
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Questions that might be worth exploring



Agriculture:

- How much P do we need for agriculture?
- Where are we going to get it?
- Will we run out?

Limiting nutrient (?):

- How and why does soil P content vary?
 - What P forms are biologically available?
- 



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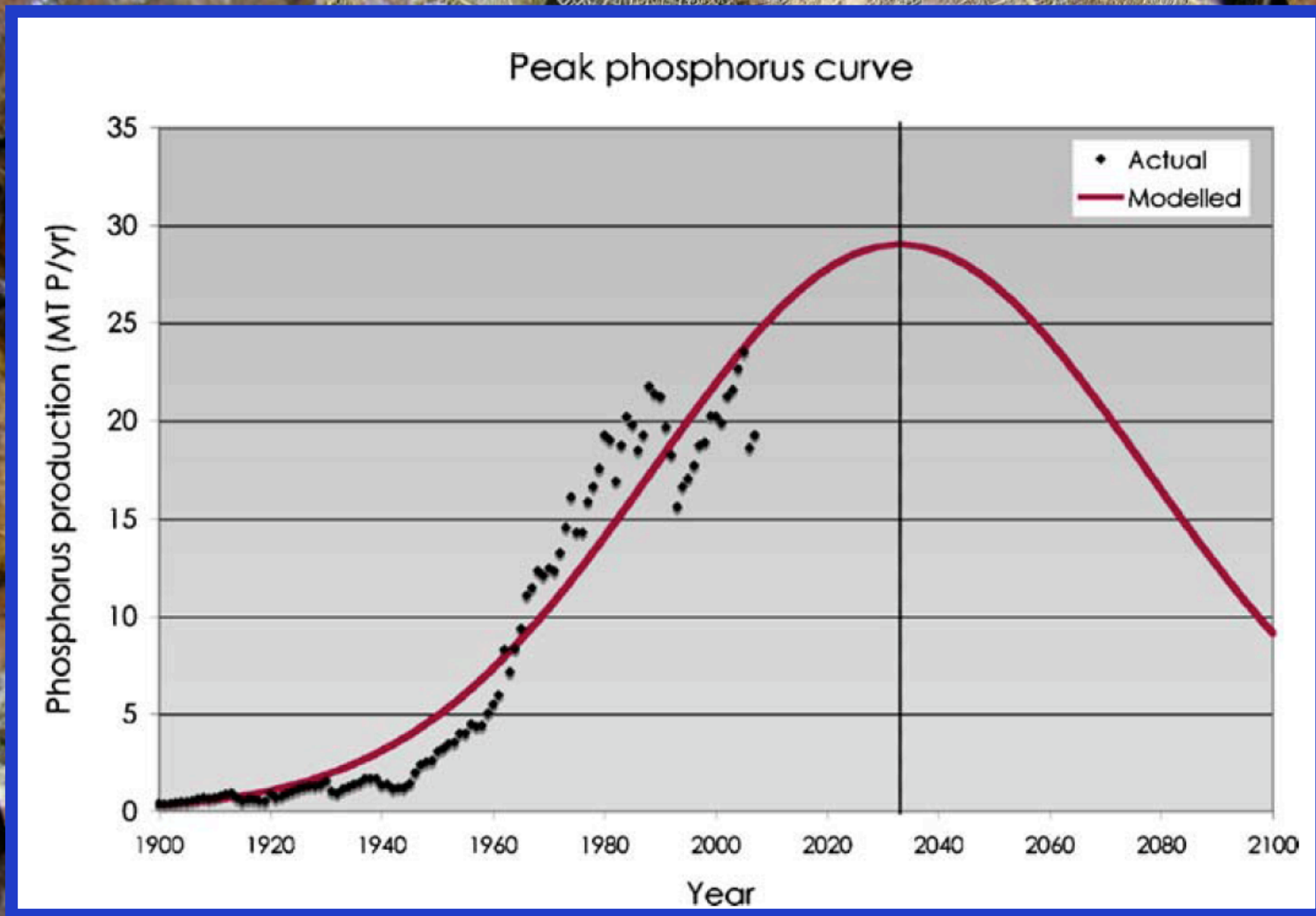
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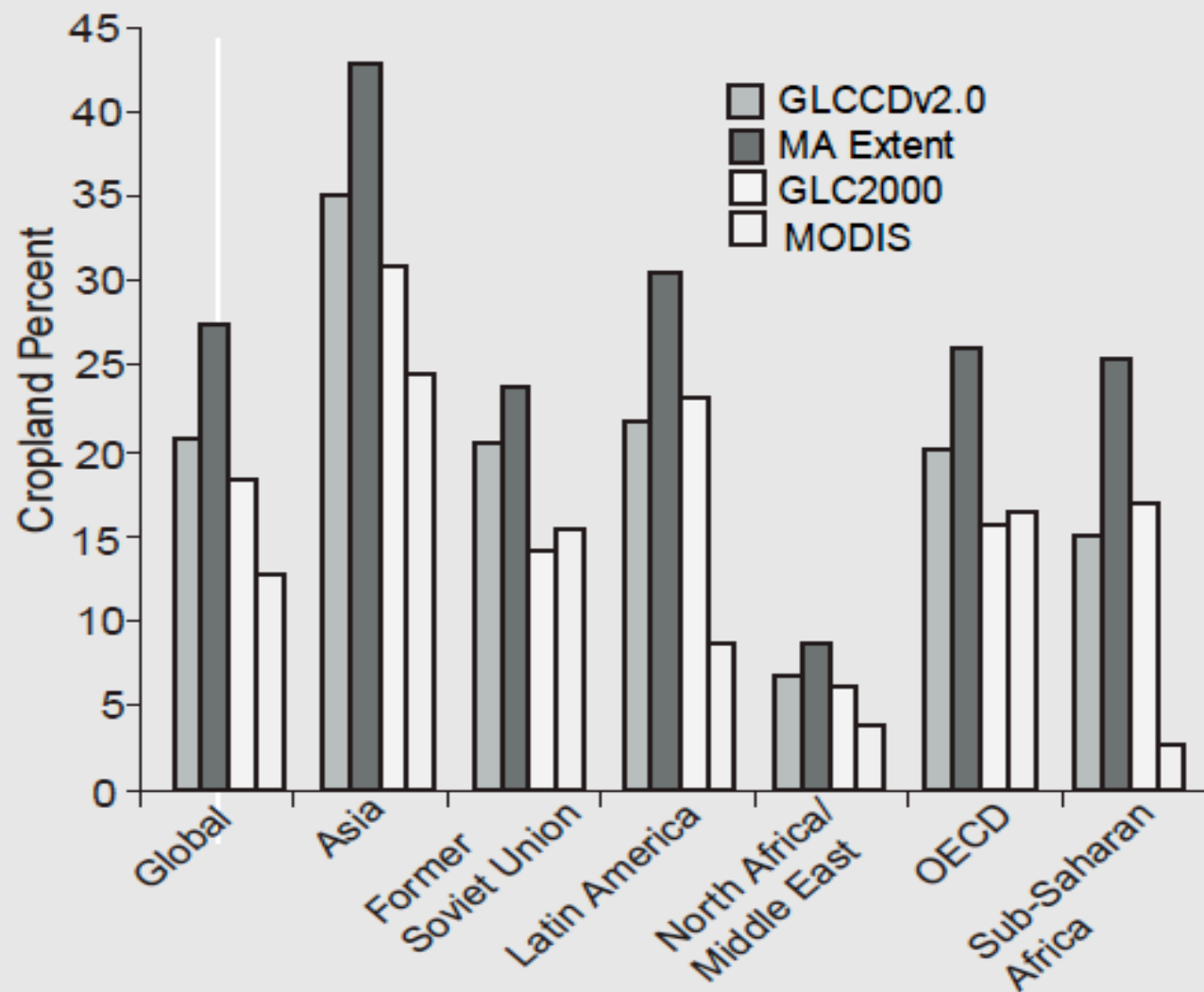


How much P do we need for agriculture?

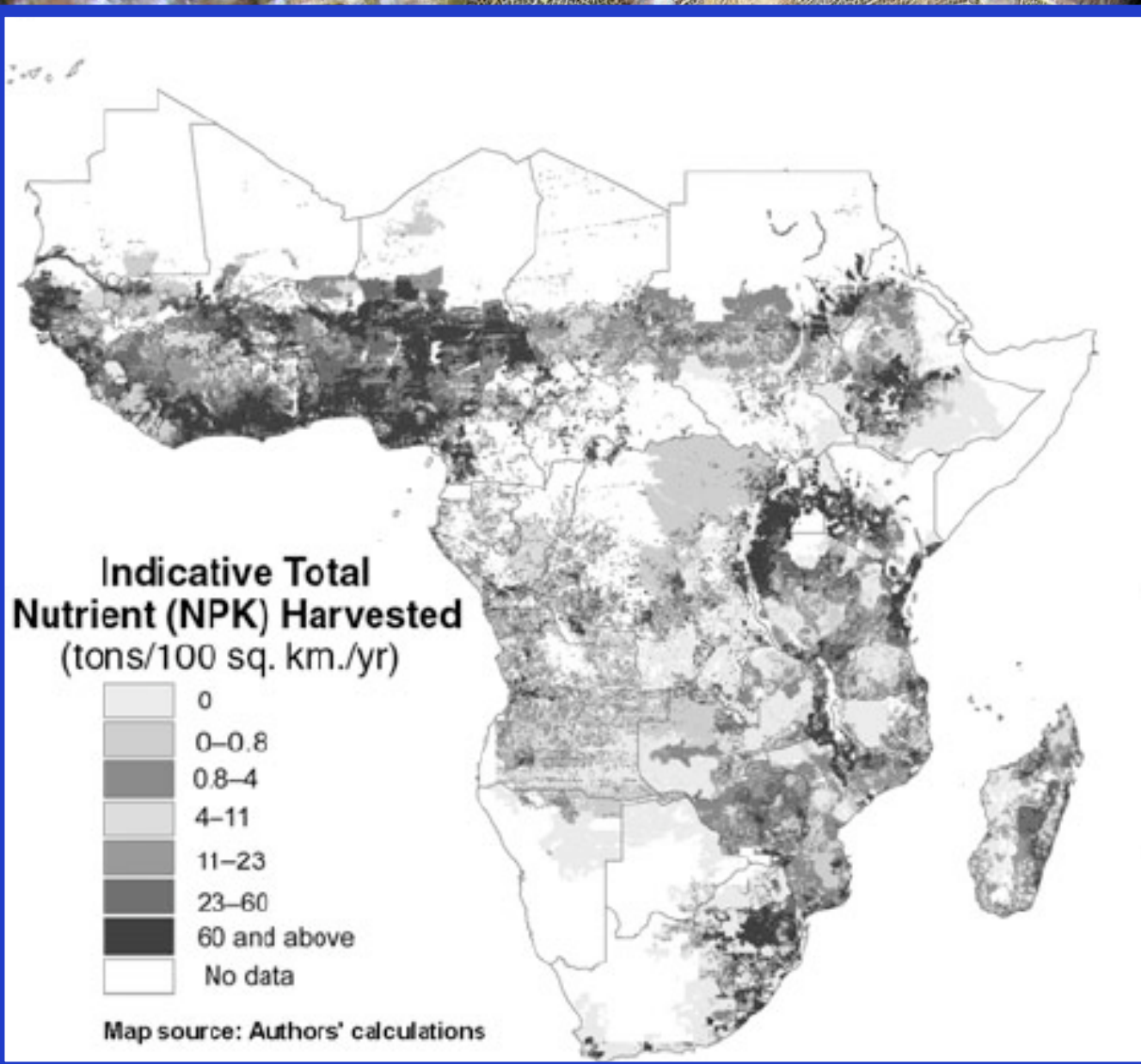


Cordell et al., 2009

How much P do we need for agriculture?



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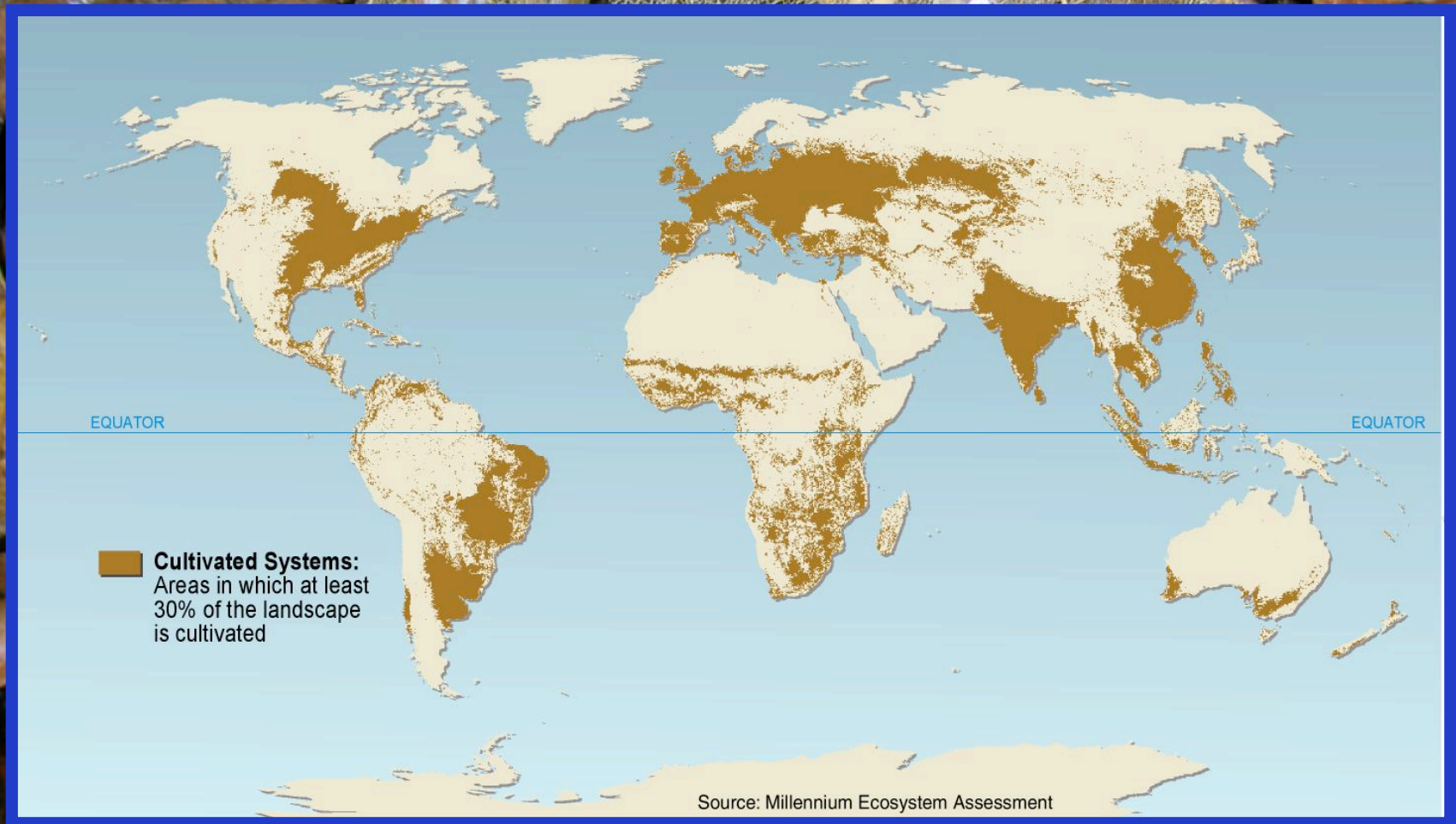


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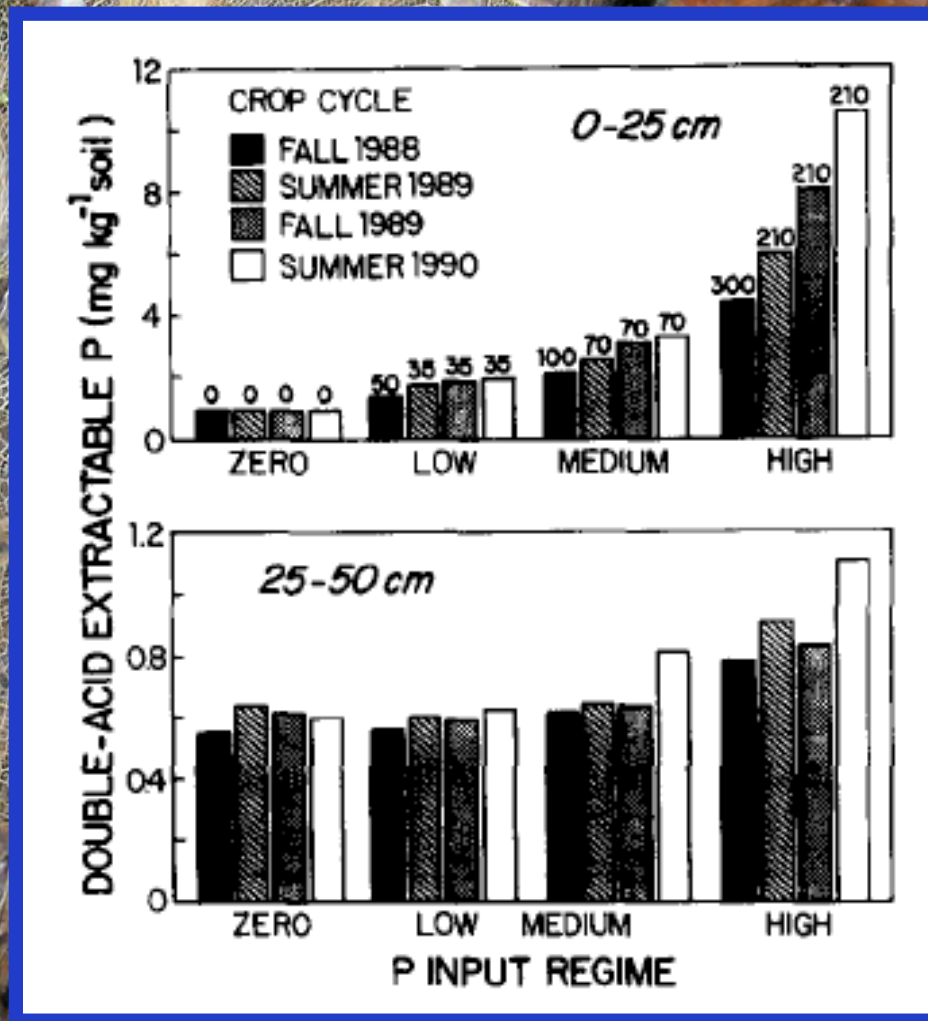
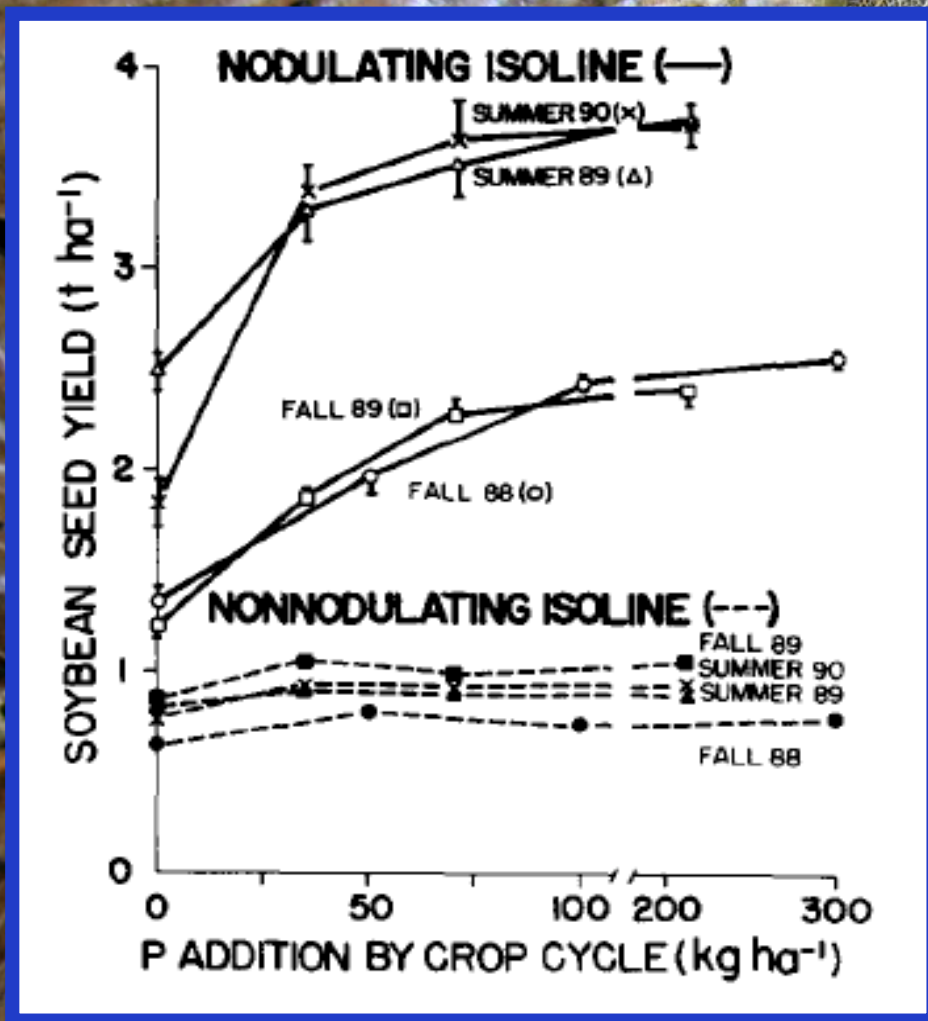


Photos courtesy of Shelby Hayhoe

How much P do we need for agriculture?

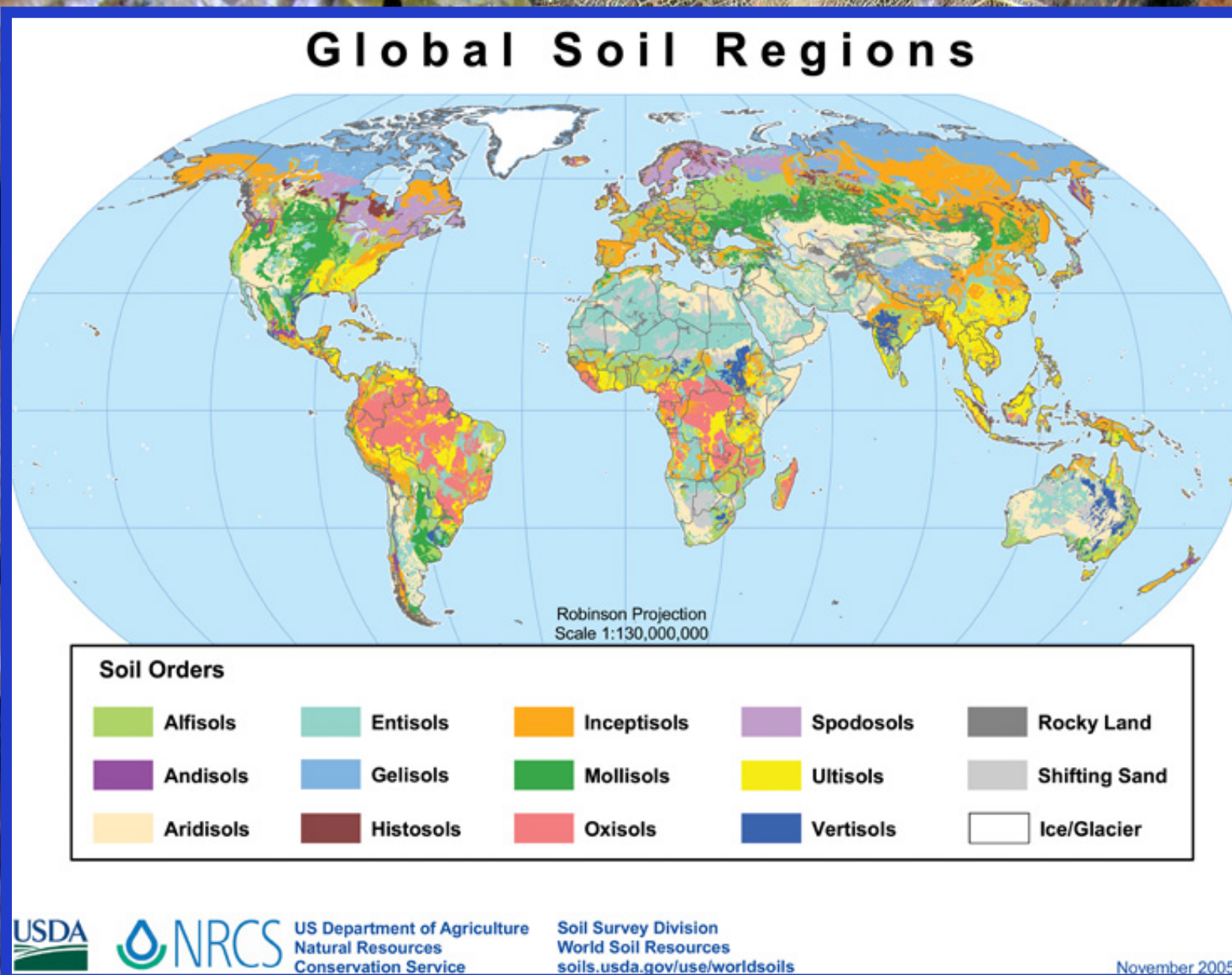


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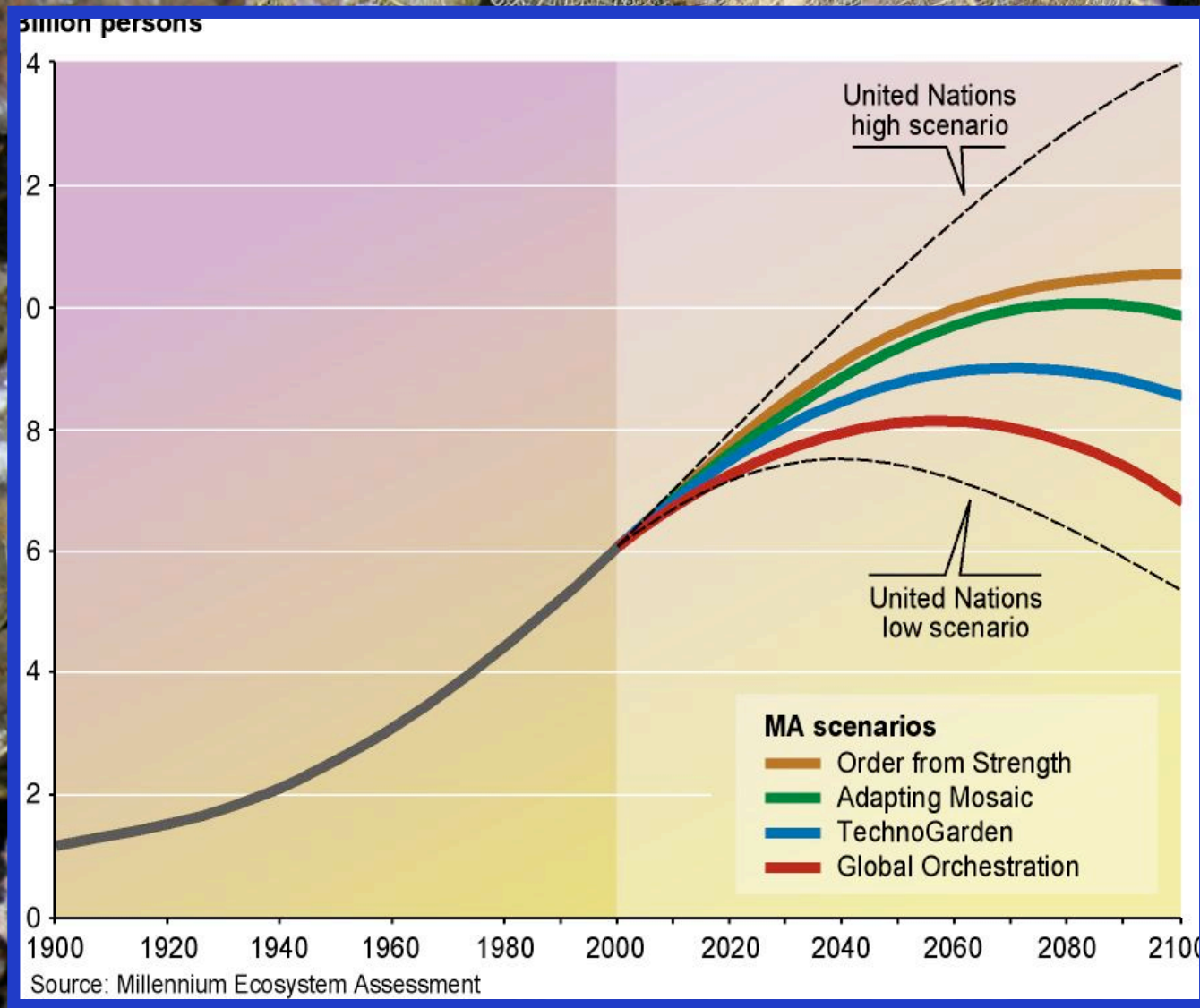


Data from high P Ultisols in Maui

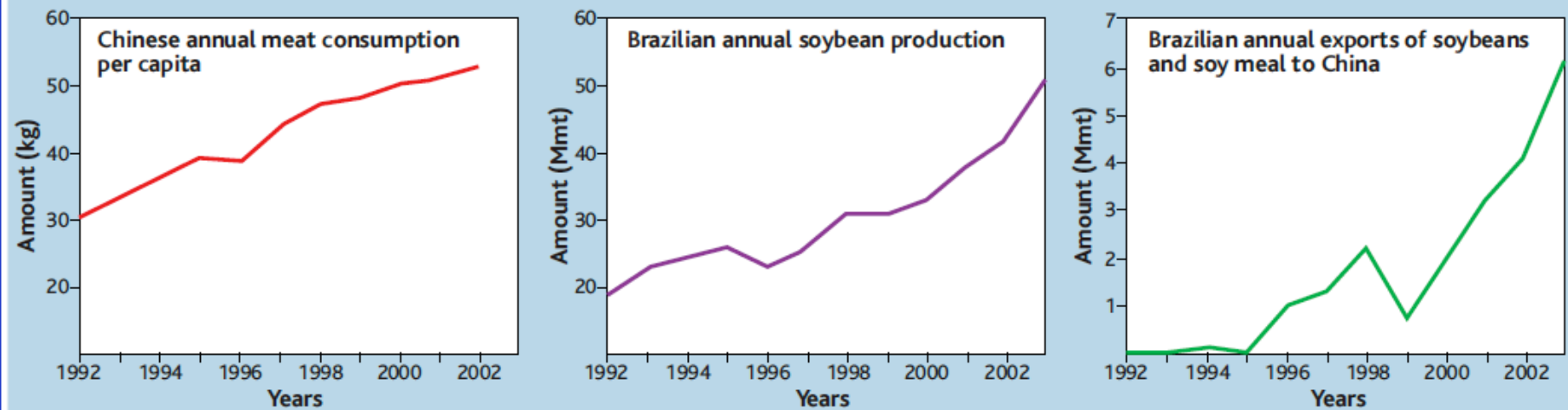
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International linkages in supply and demand of livestock products, 1992–2003 (3). Mmt, millions of metric tons.



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Limiting nutrient (?):

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Where will we get our P from?

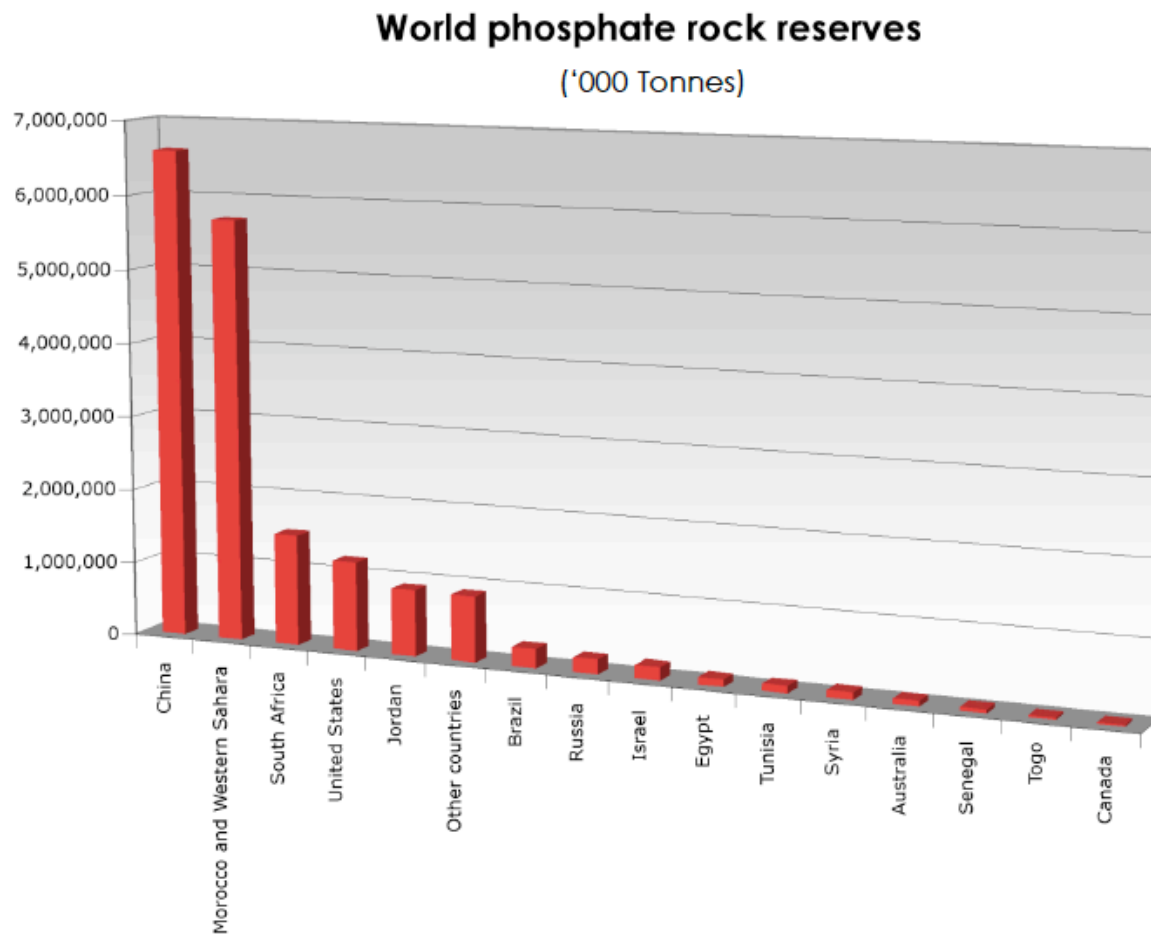
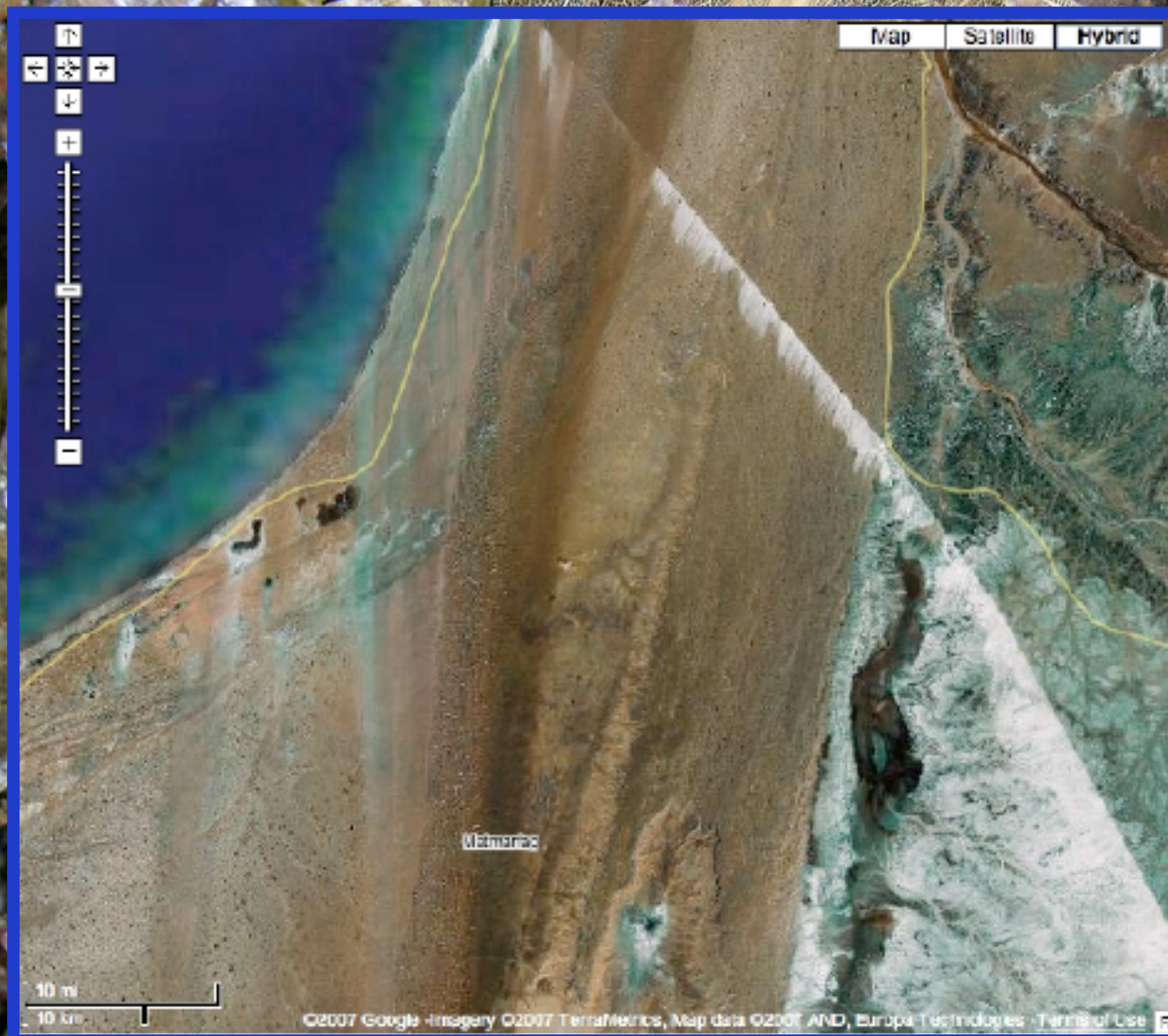


Figure 1: Global phosphorus reserves are highly geographically concentrated and are under the control of only a handful of countries. (data: Jasinski, 2008)

Where will we get our P from?



Cordell 2008 unpubl.



Questions that might be worth exploring



Agriculture:

- How much P do we need for agriculture?
- Where are we going to get it?
- **Will we run out?**


Limiting nutrient (?):

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Will we run out?

- 
- We need to better understand and control P losses from farms.
 - We need a global map of Soil P.
 - We need to know how much of the P that is in soil is available for plant uptake, and over what timescale.





Questions that might be worth exploring



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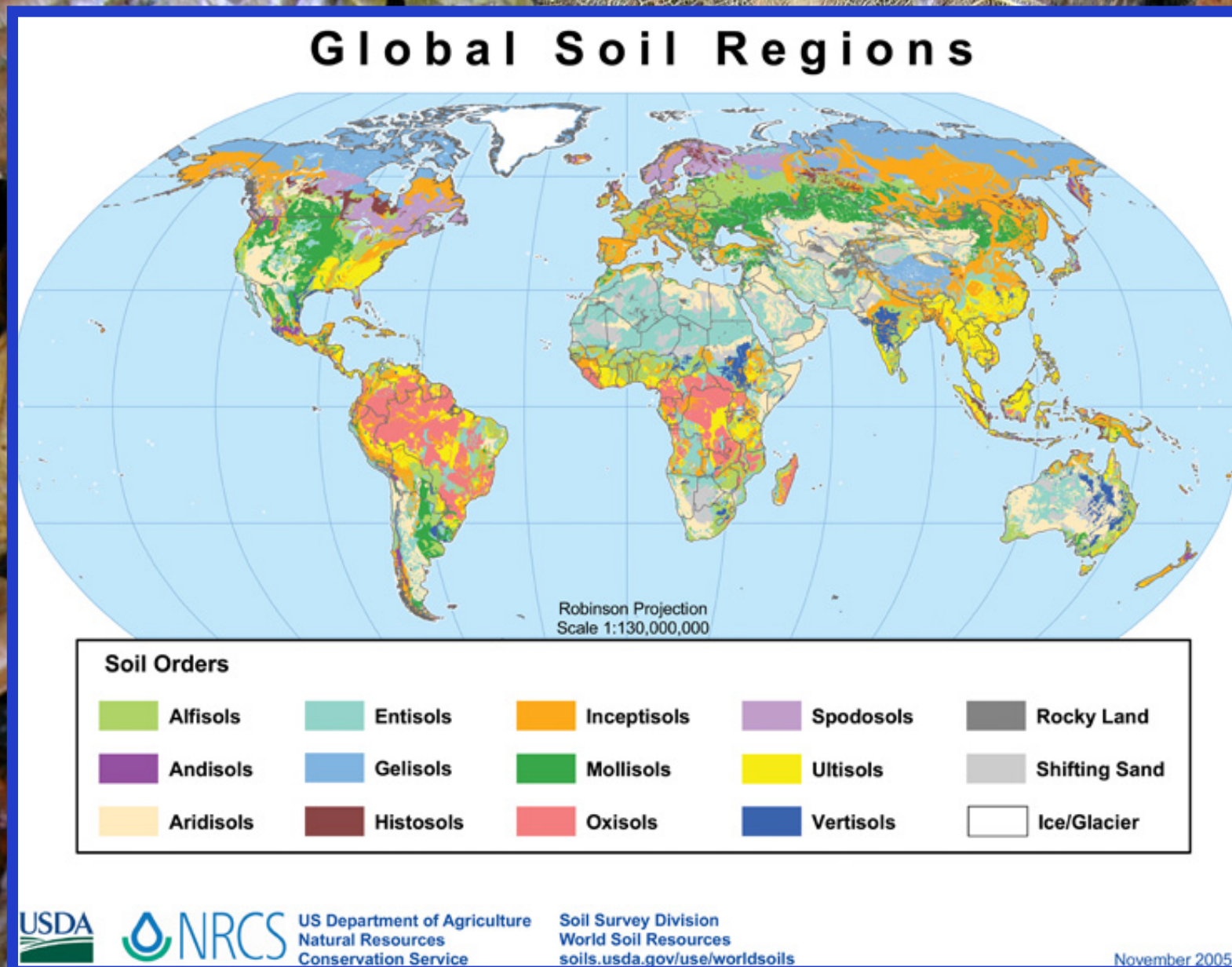
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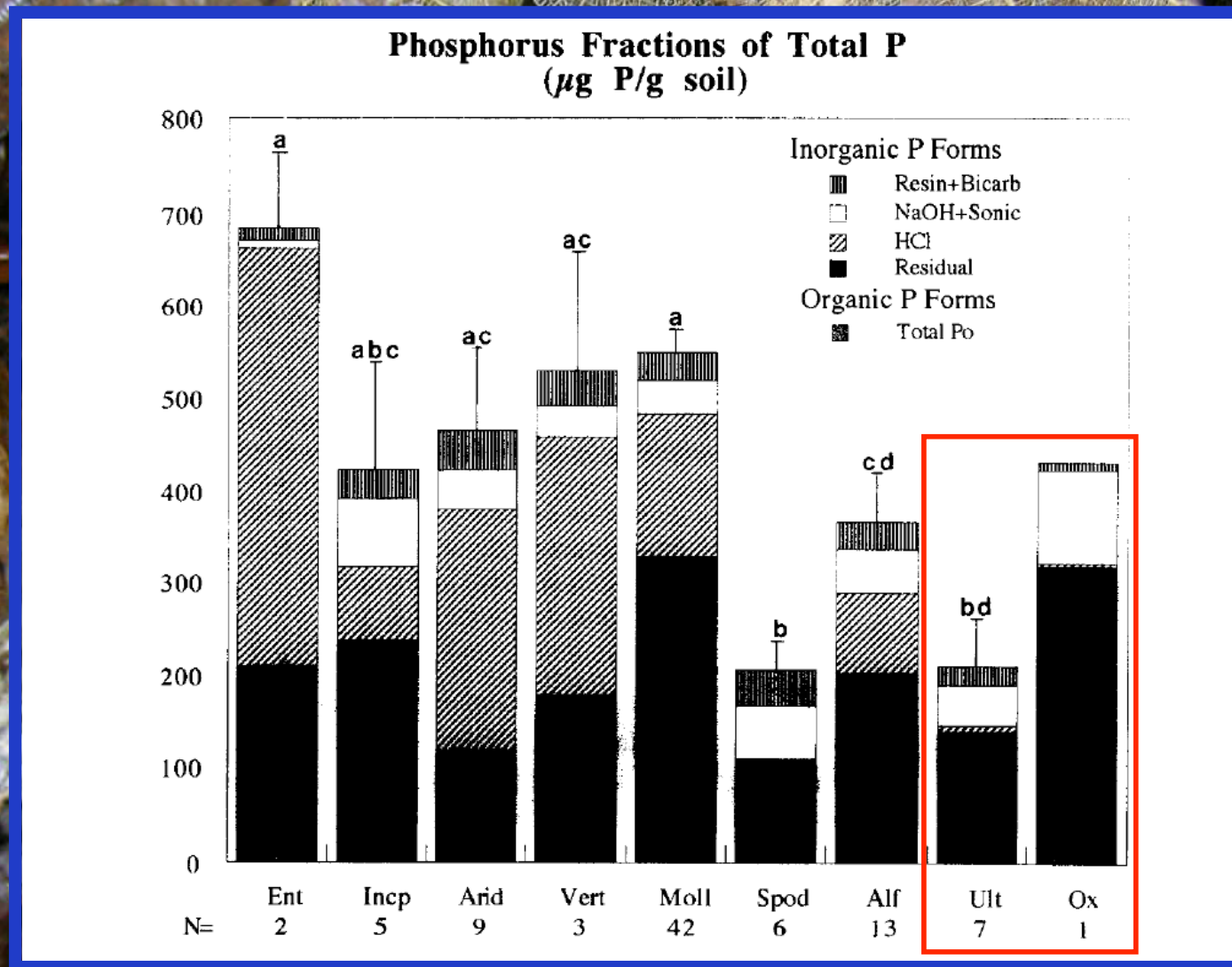
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How and why does soil P content vary?

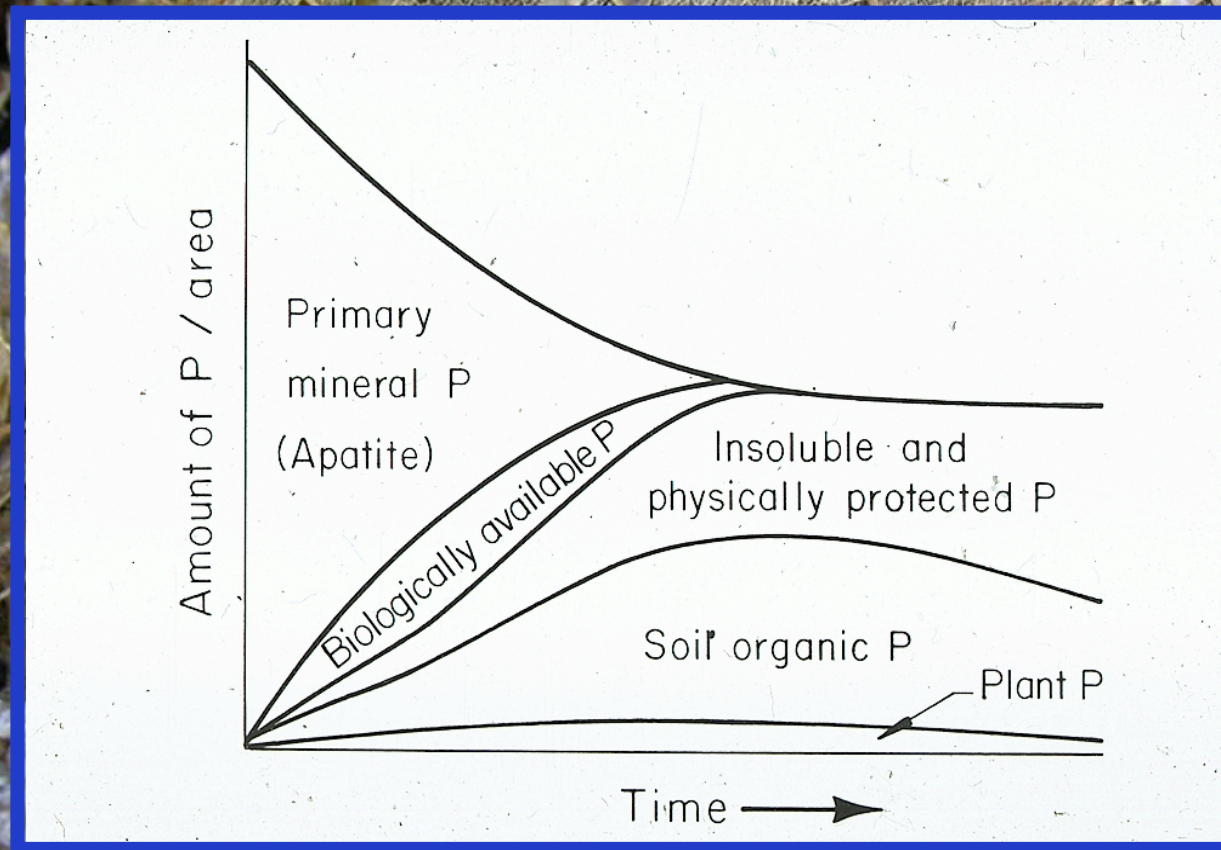


How and why does soil P content vary?



How and why does soil P content vary?

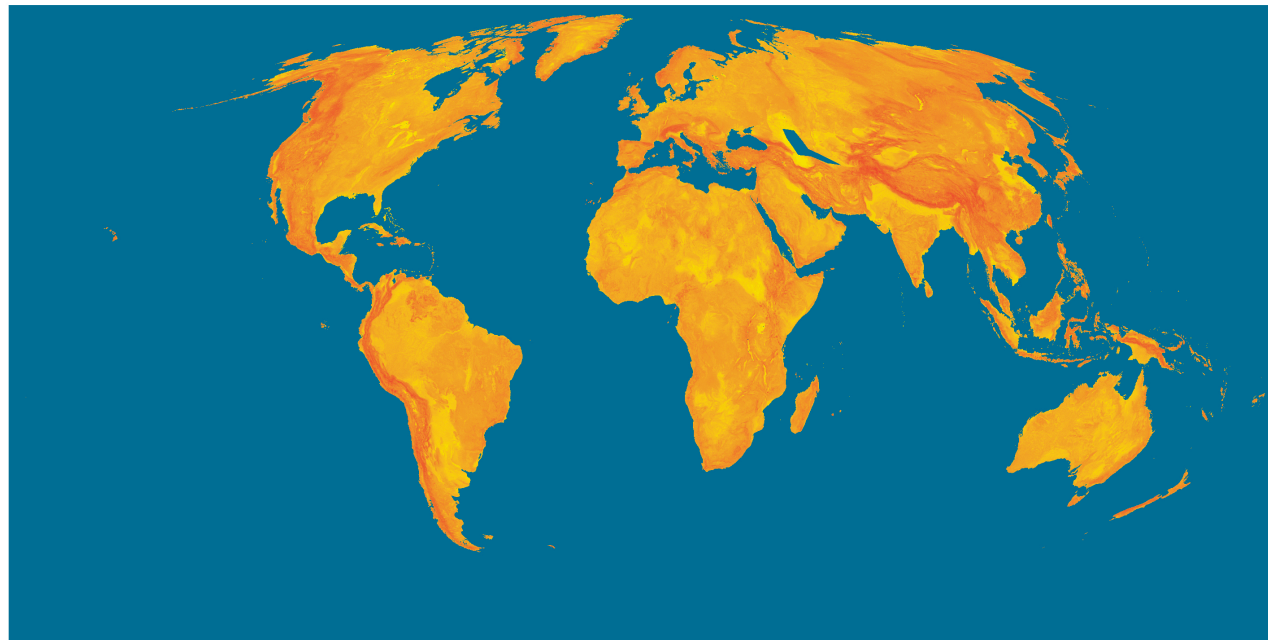
Time plays a big role in determining P status



How and why does soil P content vary?

How does “time” translate in eroding soils?

Worldwide Erosion Rates



\log_{10} Erosion Rate (mm/yr)
-4 1

After Montgomery, 2002

How and why does soil P content vary?

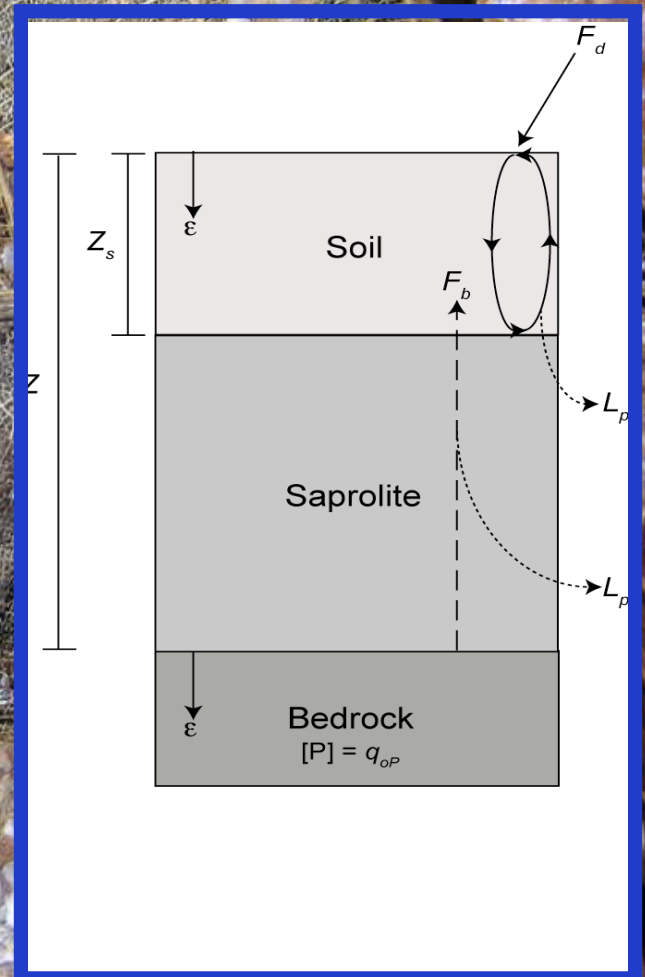
Can we use erosion in place of time to predict soil P?

Modeling Approach

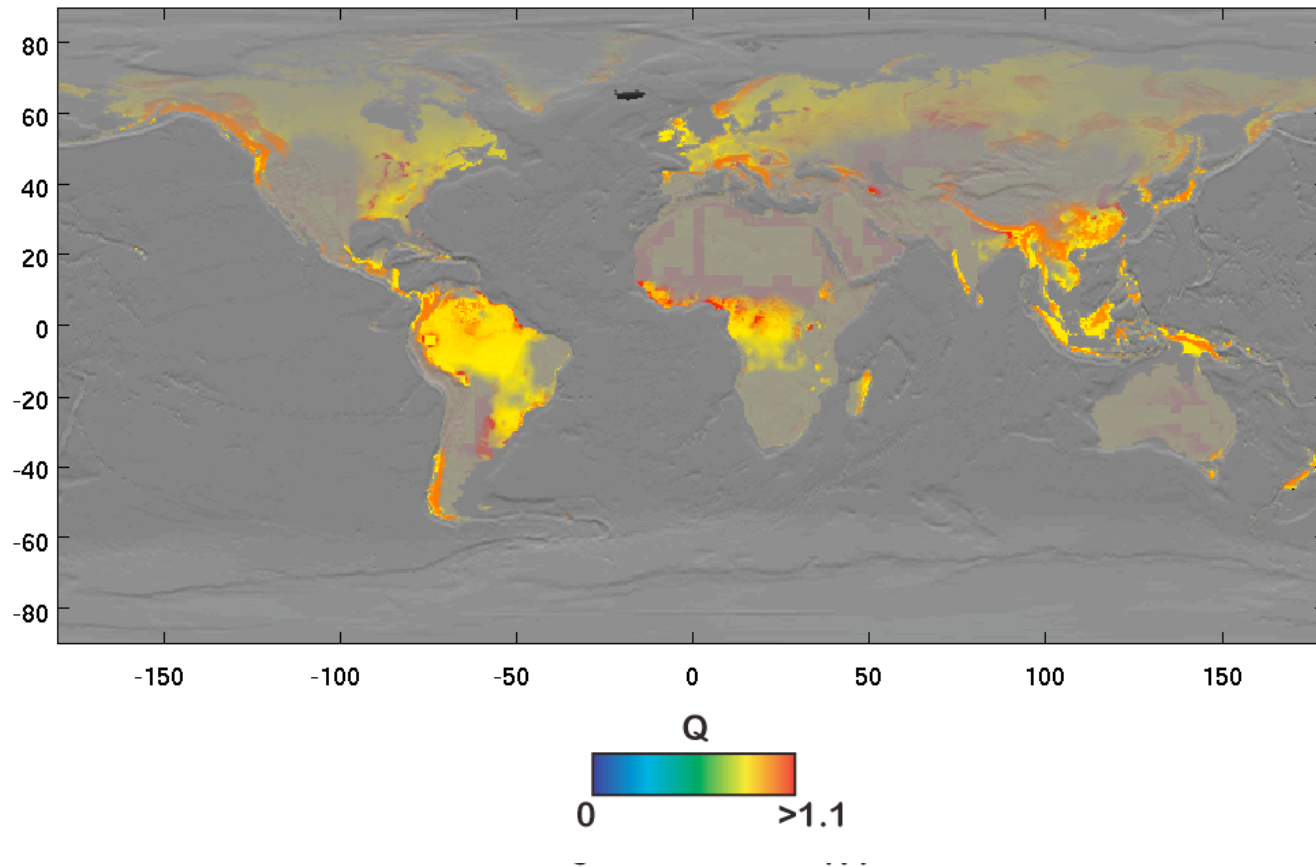
Include:

- Erosion rate
- Weathering zone thickness
- P deposition
- P leaching (currently one size fits all)
- Rock P content (currently one size fits all)

Solve for Q - [Soil P]/[Rock P]



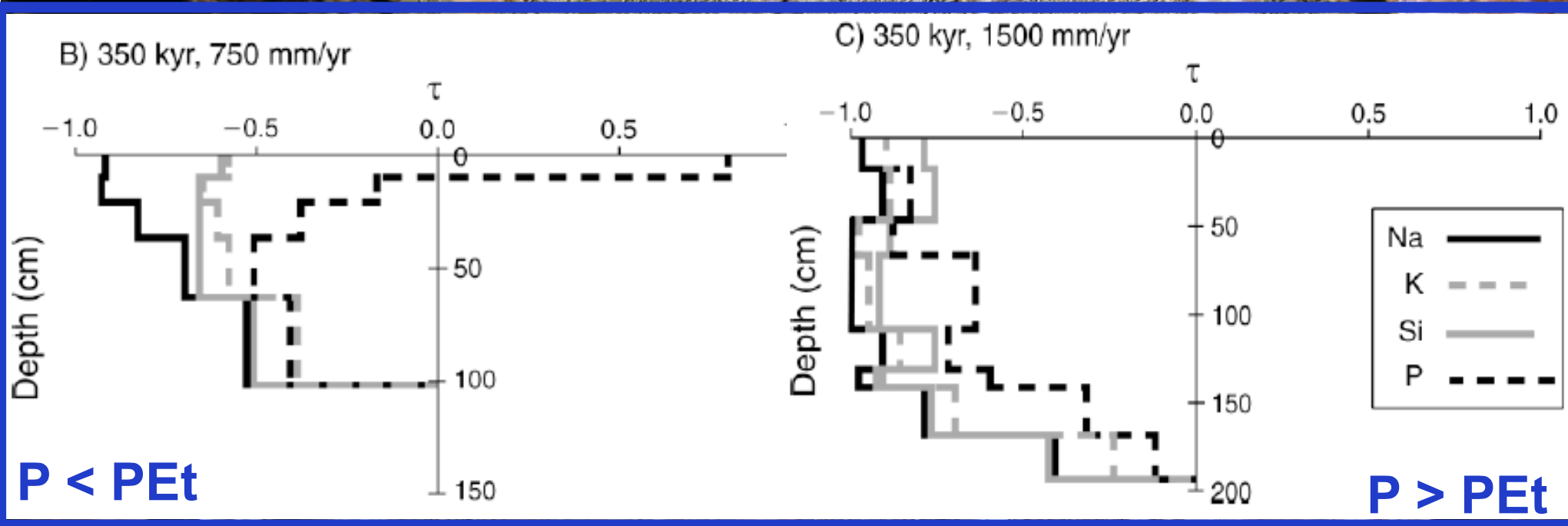
How and why does soil P content vary?



Scenarios vary because of assumptions about weathering zone thickness

How and why does soil P content vary?

Missing data: Loss rates



In fact leaching rate is NOT constant.

It may be more important than time in determining P status.

How and why does soil P content vary?

Missing data: Rock P content

Table 3.5. Composition of the bulk continental crust (see text for sources).

		%	NORM				
SiO ₂		57.3	Q		6.6		
TiO ₂		0.9	Or		6.5		
Al ₂ O ₃		15.9	Ab		26.2		
FeO		9.1	An		26.2		
MgO		5.3	Di		8.7		
CaO		7.4	Hy		24.1		
Na ₂ O		3.1	Il		1.7		
K ₂ O		1.1					
Σ		100.1					
Li	13 ppm	Ni	105 ppm	In	50 ppb	Er	2.2 ppm
Be	1.5 ppm	Cu	75 ppm	Sn	2.5 ppm	Tm	0.32 ppm
B	10 ppm	Zn	80 ppm	Sb	0.2 ppm	Yb	2.2 ppm
Na	2.30%	Ga	18 ppm	Cs	1.0 ppm	Lu	0.30 ppm
Mg	3.20%	Ge	1.6 ppm	Ba	250 ppm	Hf	3.0 ppm
Al	8.41%	As	1.0 ppm	La	16 ppm	Ta	1.0 ppm
Si	26.77%	Se	0.05 ppm	Ce	33 ppm	W	1.0 ppm
K	0.91%	Rb	32 ppm	Pr	3.9 ppm	Re	0.5 ppb
Ca	5.29%	Sr	260 ppm	Nd	16 ppm	Ir	0.1 ppb
Sc	30 ppm	Y	20 ppm	Sm	3.5 ppm	Au	3.0 ppb
Ti	5400 ppm	Zr	100 ppm	Eu	1.1 ppm	Tl	360 ppb
V	230 ppm	Nb	11 ppm	Gd	3.3 ppm	Pb	8.0 ppm
Cr	185 ppm	Mo	1.0 ppm	Tb	0.60 ppm	Bi	60 ppb
Mn	1400 ppm	Pd	1.0 ppb	Dy	3.7 ppm	Th	3.5 ppm
Fe	7.07%	Ag	80 ppb	Ho	0.78 ppm	U	0.91 ppm
Co	29 ppm	Cd	98 ppb				



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What P forms are biologically available?

0.5 g soil samples in 50 mL screw cap centrifuge tubes

Add 30 mL deionized water plus 0.4 g Dowex 1 x 8-50 anion exchange resin in bicarbonate form, shake 16 h, remove resin bag, centrifuge and discard supernatant

Resin P (P_i)

Soil

Add 30 mL NaHCO_3 (pH 8.5), shake 16 h, centrifuge collect supernatant

Bicarbonate P (P_i and P_o)

Soil

Add 30 mL 0.1M NaOH, shake 16 h, centrifuge, collect supernatant

Hydroxide P (P_i and P_o)

Add 20 mL 0.1M NaOH and sonicate in an ice bath at 75 W (Braunsonic 1510) for 2 min, make to 30 mL volume, shake 16 h, centrifuge and collect supernatant

Sonicate/hydroxide P (P_i and P_o)

Soil

Add 30 mL 1.0M HCl, shake 16 h, centrifuge and collect supernatant

Acid P (P_i)

Residual

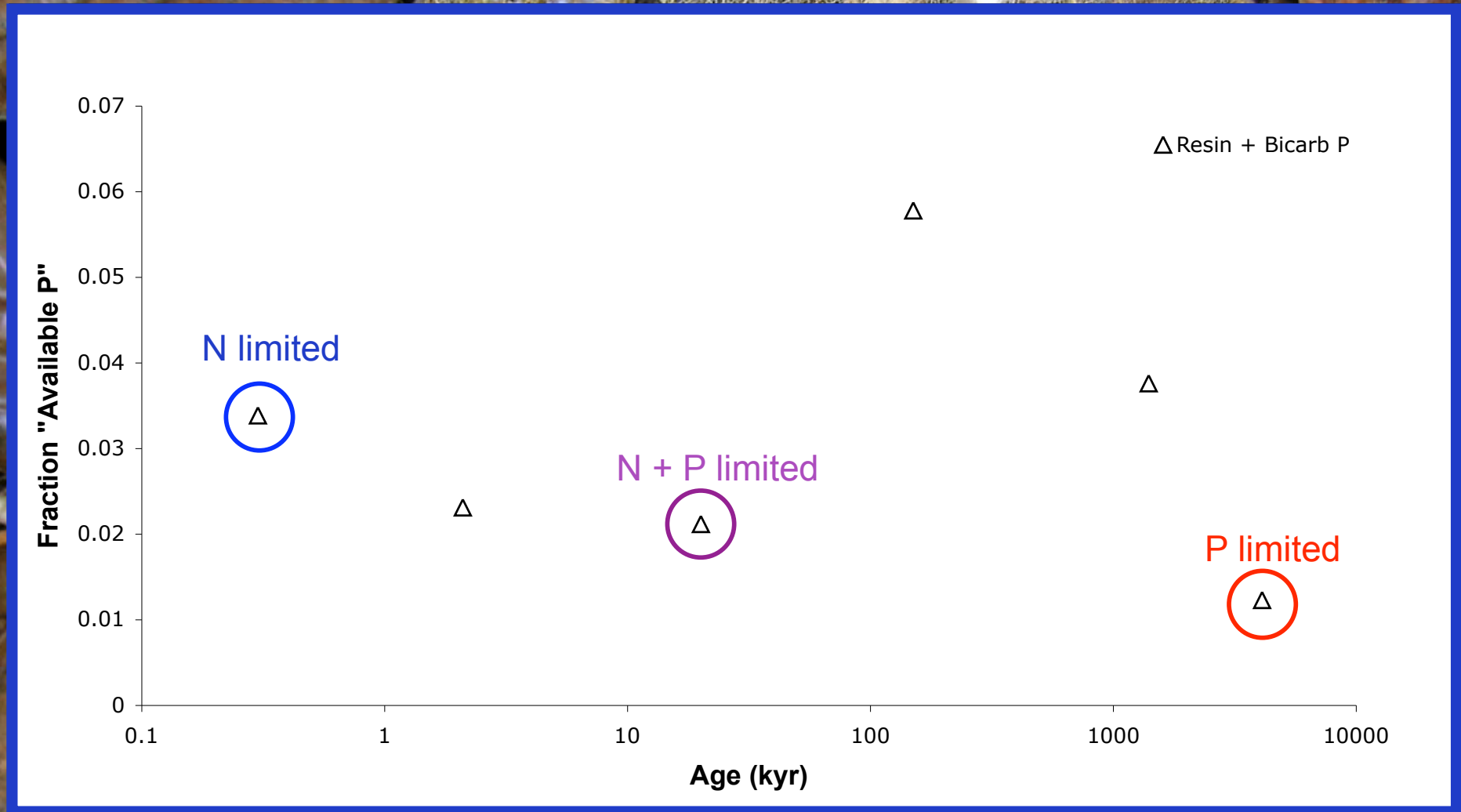
Soil

Digest with 5 mL H_2SO_4 and H_2O_2

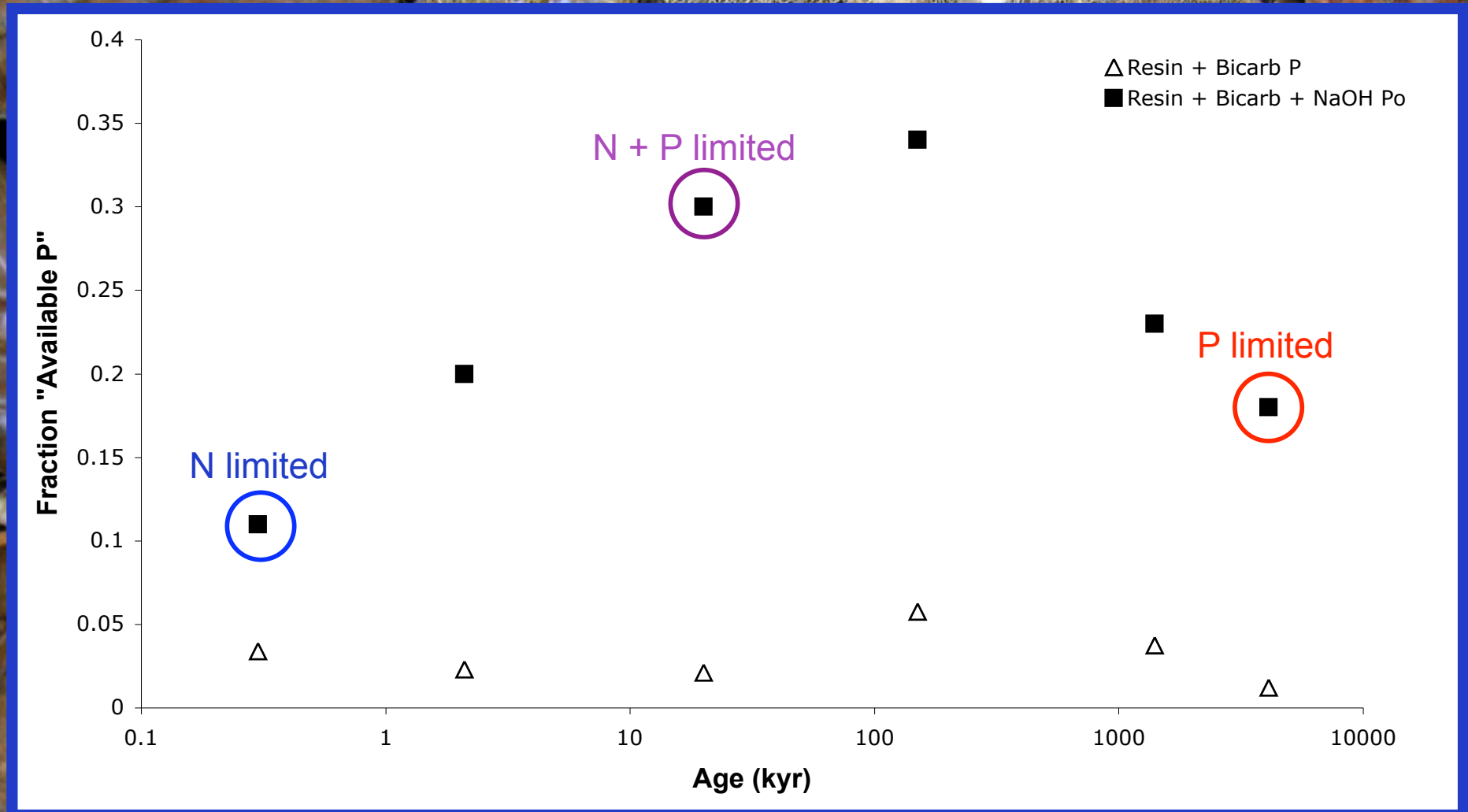
Residue P (total P only)

Fig. 1. Hedley sequential phosphorus fractionation method for soils (after Tiessen et al., 1984).

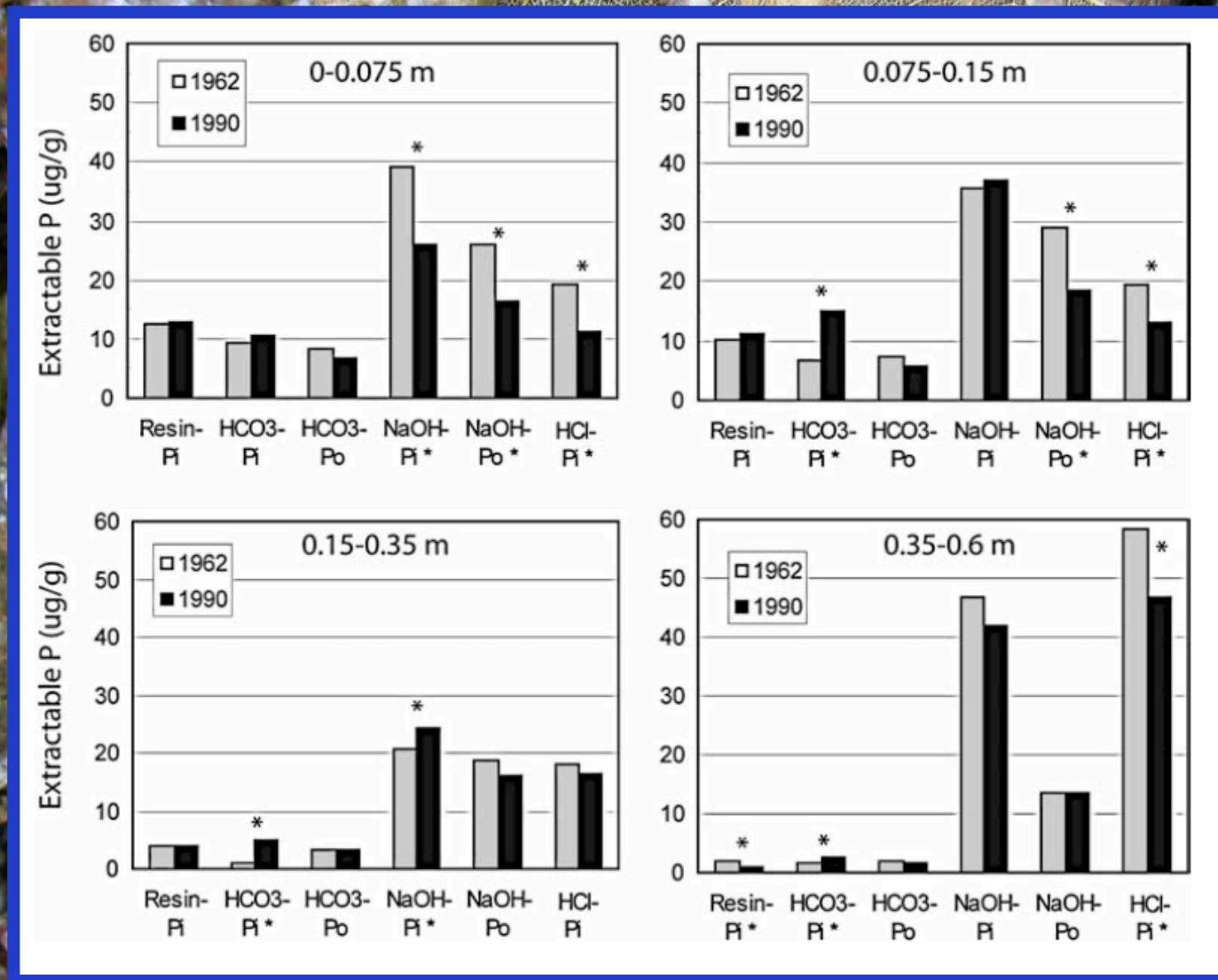
What P forms are biologically available?



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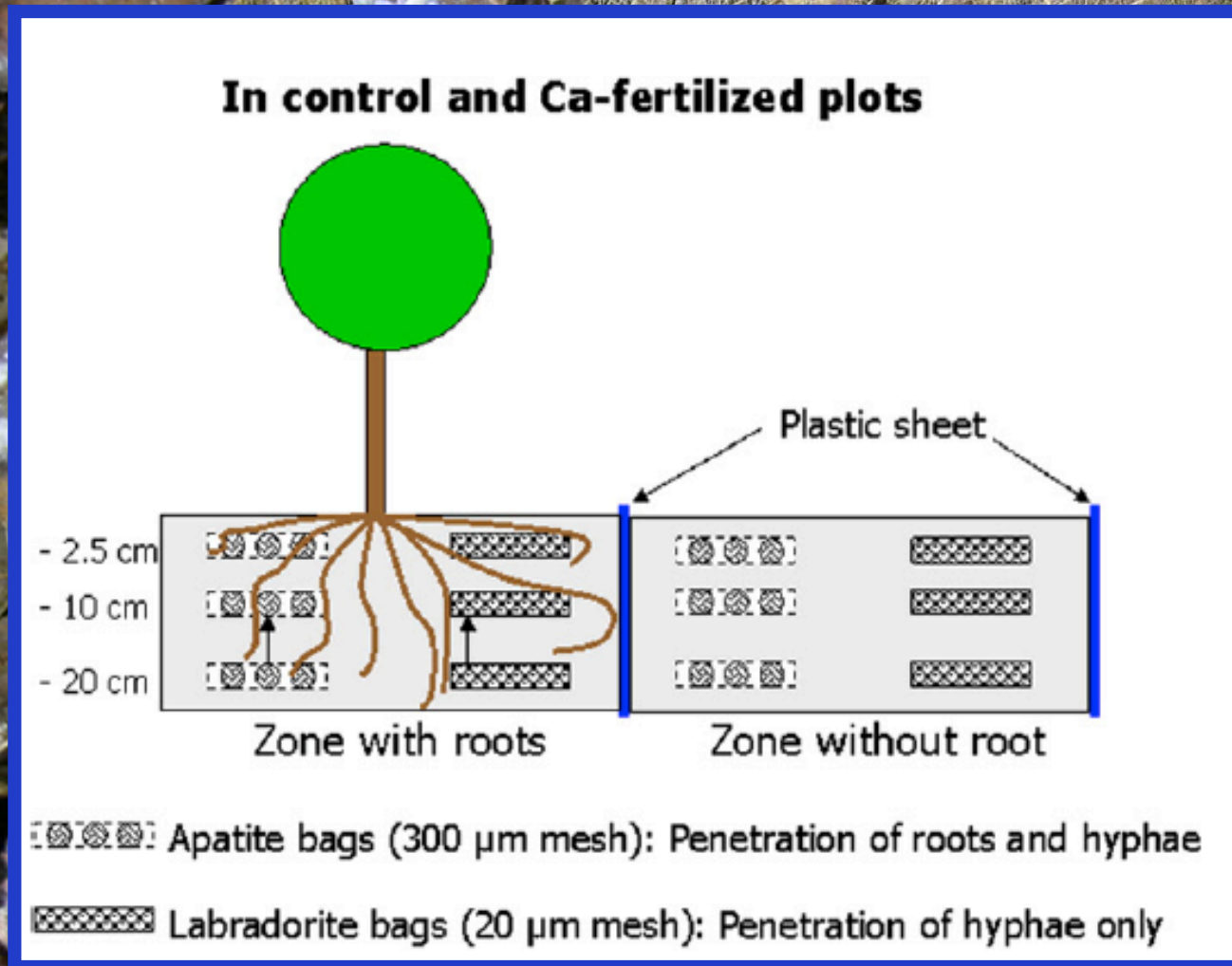


What P forms are biologically available?

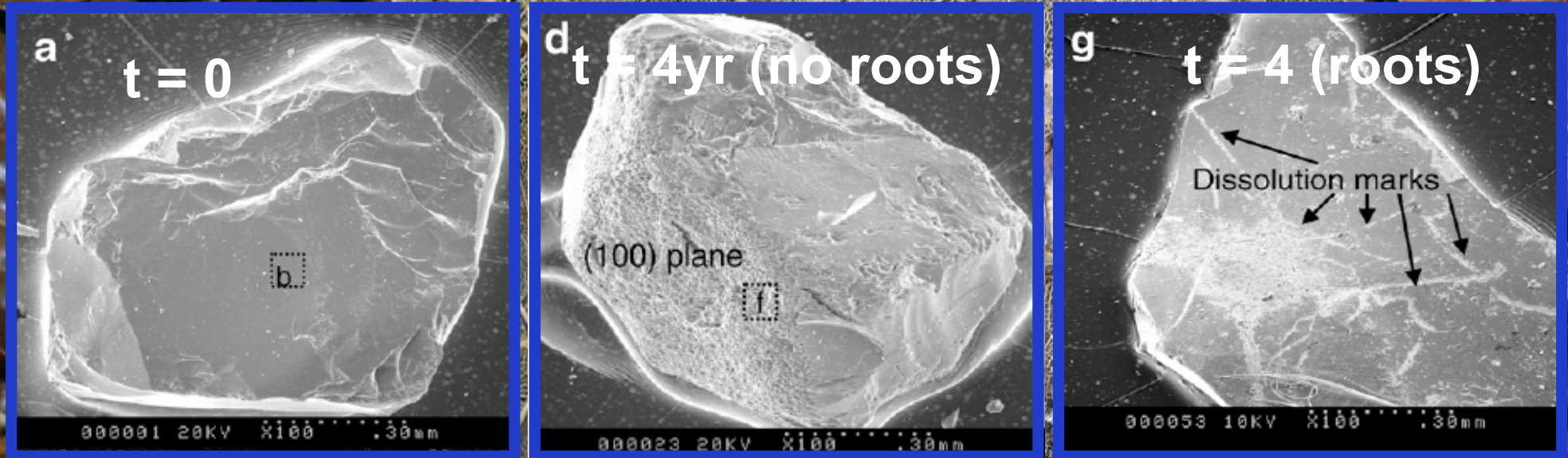


Calhoun experimental forest planted w/seedlings in 1962

What P forms are biologically available?



What P forms are biologically available?



- Is fungal mining the P equivalent of N fixation?
- What determines how long apatite stays around?



Potential Directions



Agriculture:

- A review of P binding capacities of Oxisols and Ultisols, coupled with assessments of likely farm expansion and readily available P deposits.

Limiting nutrient (?):

- Revisiting (gathering new) chronosequence data to get a handle on P depletion rates in different climate zones.
- Experiments that get at availability/turnover times for P.



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